

**Summary of PhD Thesis**

**Mass Spectrometric Behaviour of  
Phosphorus-containing Heterocycles and  
Ring-chain Tautomerism of some 1,3-*O,N*-heterocycles**

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**2008**

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## A. Introduction and aims

Compounds possessing an oxazaphosphorinane ring system are very important pharmacological building blocks of matrix metalloproteinase-inhibitors and alkylating anti-cancer drugs.

Because of their fundamental importance (bioactive and synthetic), we set out to elucidate the effects of various substituents on the ring nitrogen and phosphorus atoms, *cis-trans* isomerism and the phosphorus stereochemistry on the fragmentations of the diastereomeric pairs under 70 eV ionization for a few saturated 3,1,2-benzoxazaphosphinine-2-oxides (**1–14**), their nitrogen analogue 1,3,2-benzodiazaphosphinine-2-oxides (**15–31**) and various condensed 1,3,4,2-oxadiazaphosphinane derivatives (**32–39**).

The structures of numerous five- and six-membered, *N*-unsubstituted, 1,3-*X,N*-heterocycles (*X* = O, S and NR) can be characterized by the ring-chain tautomeric equilibria of the 1,3-*X,N*-heterocycles and the corresponding Schiff bases. This ring-chain tautomeric process influences the reactivity and therefore the synthetic applicability of these compounds.

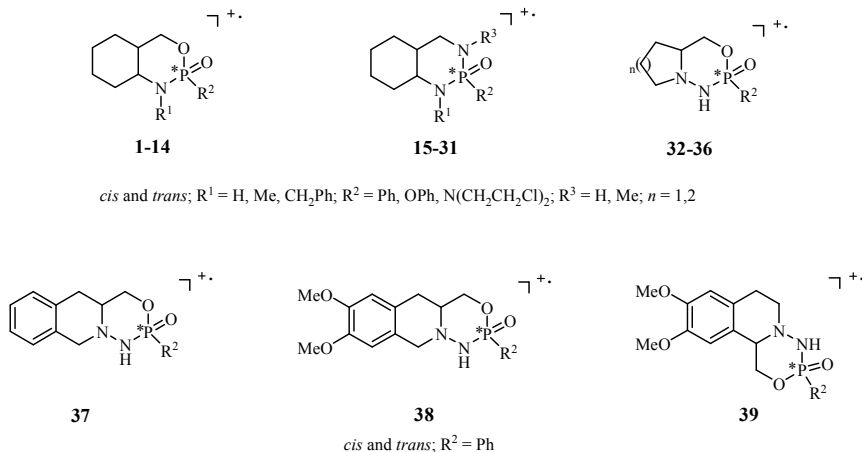
The electronic effects of 2-aryl substituents were studied most thoroughly in solution. A linear correlation was found between the log *K* values of the equilibria (*K* = [ring]/[chain]) and the Hammett-Brown parameters  $\sigma^+$  of the different substituents *X* on the 2-aryl group (Eq. 1):

$$\log K = \rho\sigma^+ + f \quad \text{Eq. 1}$$

As relatively little attention has been paid to the gas-phase tautomeric equilibria, it was important to investigate the electronic effects of 2-aryl substituents on the tautomeric equilibria for a few five- and six-membered 1,3-*O,N*-heterocyclic compounds. Electron ionization (EI) mass spectrometry can be used for this purpose by observing the ratio of the relative abundances of fragment ions associated with one or the other tautomeric form. A further aim was to inspect the effects of 4-alkyl or 4-phenyl substituents on the ring-chain tautomeric equilibria.

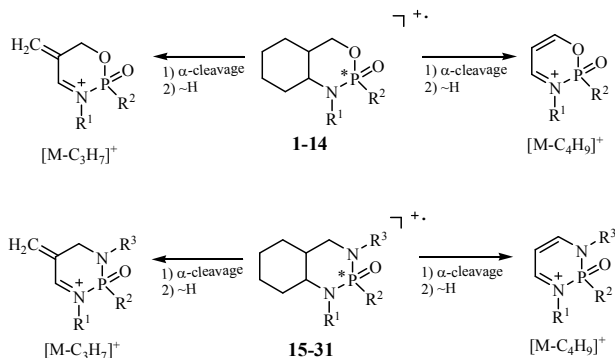
## B. Results

1. The derivatives of *cis*- and *trans* cyclopentane- or cyclohexane-fused 3,1,2-benzoxazaphosphinane- (**1–14**) and 1,3,2-benzodiazaphosphinane- (**15–31**) and 1,3,4,2-oxadiazaphosphinane-2-oxides (**32–39**) (Figure 1) gave rise to similar EI mass spectra, because the isomers underwent fast ring cleavage reactions, during which the differences in stereochemistry were lost.



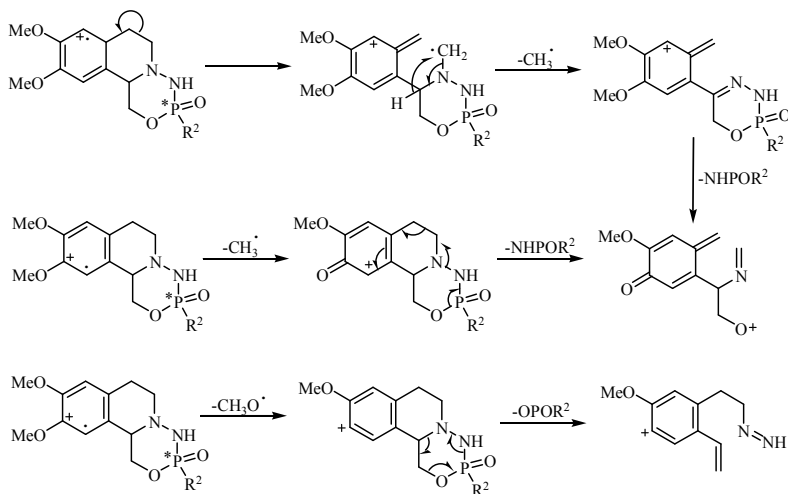
**Figure 1**

2. The ready ejection of various alkyl radicals was typical for compounds **1–14** and **15–31**. The formations of fragment ions [M–C<sub>3</sub>H<sub>7</sub>]<sup>+</sup> and [M–C<sub>4</sub>H<sub>9</sub>]<sup>+</sup> accompanied by H-transfers were favoured (Scheme 1).

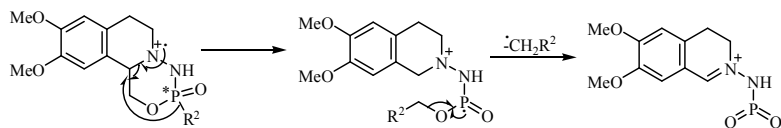


**Scheme 1**

3. Due to the factors stabilizing the positive charge, various fragmentation pathways of  $M^{+}$  (the molecular ion) were observed for the oxadiazaporphine compounds (**32–39**), e.g. fragmentations initiated from the aromatic moieties (Scheme 2a) and fragmentations occurring after the migration of  $R^2$  (Scheme 2b).

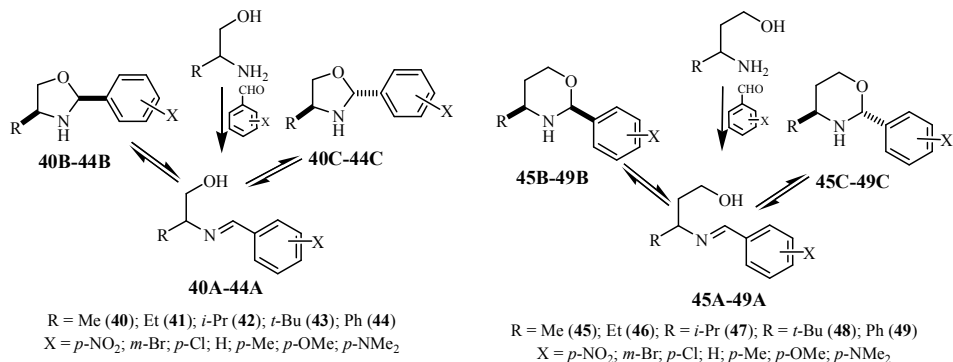


**Scheme 2a**



**Scheme 2b**

4. Compounds **40–49**, prepared by the condensation of the appropriate aminoalcohols with aromatic aldehydes, exhibited three-component tautomeric equilibria (open-chain (**A**) and cyclic *cis*(**B**) and *trans*(**C**)) (Scheme 3) in deuteriochloroform, and two-component tautomeric equilibria involving open-chain(**A**) and ring forms(**B+C**) in the gas phase, when both the *cis* ring form and the *trans* ring form gave the same fragment ions.



**Scheme 3**

**5.** Each equilibrium could be characterized by the Hammett equation (Eq. 1). Besides the steric effect ( $V^a$ , the Meyer parameter) of the 4-alkyl substituent, the different electronic (inductive and resonance) properties of the 2-aryl groups also influenced the tautomeric equilibria of the oxazolidinones, which could be described by Hansch-type equations (Eqs 2 and 3):

$$\log K = k + \rho^R V^a + \rho^X \sigma^{+X} \quad \text{Eq. 2}$$

$$\log K = k + \rho^R V^a + \rho_F^X \sigma_F^X + \rho_R^X \sigma_R^X \quad \text{Eq. 3}$$

Through multiple linear regression analysis of Eqs 2 and 3, we determined the free term ( $k$ ) and the coefficients of the independent variables ( $V^a$ ,  $\sigma_F^X$  and  $\sigma_R^X$ ), which are given in Table 1.

**Table 1.** Multiple linear regression analysis of the equilibrium constant  $\log K_B$  for the *cis*  $\rightleftharpoons$  open-chain and  $\log K_C$  for the *trans*  $\rightleftharpoons$  open-chain equilibria for compounds **40–43** and **45–48** according to Eq. 3

Equilibrium	$k$	$\rho^R$	$\rho_F^X$	$\rho_R^X$	$r$
<b>40–43B</b> $\rightleftharpoons$ <b>40–43A</b>	-1.532	0.189	0.386	1.114	0.974
<b>40–43C</b> $\rightleftharpoons$ <b>40–43A</b>	-1.748	0.190	0.567	1.384	0.983
<b>45–48B</b> $\rightleftharpoons$ <b>45–48A</b>	0.422	<sup>a</sup>	0.735	1.607	0.966
<b>45–48C</b> $\rightleftharpoons$ <b>45–48A</b>	-0.938	<sup>a</sup>	0.930	1.590	0.976

<sup>a</sup> Insignificant results.

A significant correlation was found between the Meyer parameter (which characterizes the steric effect of the 4-alkyl substituent) and the tautomeric equilibria for the oxazolidines, *i.e.* increasing bulk of the 4-alkyl group favoured the ring-closed tautomers, but the Meyer parameter did not indicate a significant effect for the corresponding 1,3-oxazines.

The opposite tendency was observed for the parameters  $\sigma_F^X$  and  $\sigma_R^X$ , *i.e.* the parameters describing the inductive and resonance effects of different aryl substituents influenced the equilibria more markedly for the tetrahydro-1,3-oxazines than for the oxazolidines.

**6.** For the oxazolidine derivatives, two equilibrium constants were defined. One of them was calculated with the relative abundances of the ring-related fragment ion  $[M-CH_2O]^{++}$  and of the open-chain-related  $[M-CH_2OH]^+$ . To describe the equilibrium closer, besides these two fragment ions, further fragment ions were also considered. For the corresponding 1,3-oxazine derivatives, the tautomeric equilibria could be described with the latter approach only.

**7.** For the oxazines and oxazolidines, better correlations were achieved by excluding the strong electron donor-substituted derivatives from the regression analysis (Tables 2a), due to some factor which probably makes the molecular ions of these derivatives favour the open-chain forms in the final gas-phase equilibria.

**Table 2a.** Linear regression analysis data for 4-*t*-Bu-2-aryl-substituted oxazolidines (**43**) and 4-*t*-Bu-2-aryl-substituted 3,4,5,6-tetrahydro-2*H*-1,3-oxazines (**48**) in gas phase

	Equilibrium	No. of points	Slope <sup>a</sup> ( $\rho$ )	Intercept <sup>a</sup> ( $f$ )	Correlation coefficient ( $r$ )
70 eV	<b>43(B+C) <math>\rightleftharpoons</math> 43A</b>	6 <sup>b,c</sup>	0.69( $\pm$ 0.06)	-1.40( $\pm$ 0.03)	0.987
	<b>48(B+C) <math>\rightleftharpoons</math> 48A</b>	6 <sup>b</sup>	0.54( $\pm$ 0.12)	1.60( $\pm$ 0.06)	0.908
14 eV	<b>43(B+C) <math>\rightleftharpoons</math> 43A</b>	6 <sup>b,c</sup>	0.45( $\pm$ 0.10)	-1.45( $\pm$ 0.05)	0.914
	<b>48(B+C) <math>\rightleftharpoons</math> 48A</b>	6 <sup>b</sup>	0.33( $\pm$ 0.02)	1.79( $\pm$ 0.01)	0.996

<sup>a</sup> Standard errors in parentheses. <sup>b</sup> *p*-NMe<sub>2</sub> derivatives excluded. <sup>c</sup> Calculated via  $\log K_1 = \log ([M-CH_2O]^{++})/([M-CH_2OH]^+) = \log [B+C]/[A]$

As shown in Tables 2a and b, better correlations were achieved for each case in deuteriochloroform than in the gas phase for the corresponding equilibria.

**Table 2b.** Linear regression analysis data for 4-*t*-Bu-2-aryl-substituted oxazolidines (**43**) and 4-*t*-Bu-2-aryl-substituted 3,4,5,6-tetrahydro-2*H*-1,3-oxazines (**48**) in solution

	Equilibrium	No. of points	Slope <sup>a</sup> ( $\rho$ )	Intercept <sup>a</sup> ( $f$ )	Correlation coefficient ( $r$ )
	<b>43B</b> $\rightleftharpoons$ <b>43A</b>	7	0.46( $\pm$ 0.01)	-0.13( $\pm$ 0.01)	0.996
	<b>43C</b> $\rightleftharpoons$ <b>43A</b>	7	0.54( $\pm$ 0.01)	-0.30( $\pm$ 0.01)	0.996
CDCl <sub>3</sub>	<b>43(B+C)</b> $\rightleftharpoons$ <b>43A</b>	7	0.49( $\pm$ 0.01)	0.10( $\pm$ 0.01)	0.999
	<b>48B</b> $\rightleftharpoons$ <b>48A</b>	7	0.65( $\pm$ 0.04)	0.63( $\pm$ 0.03)	0.992
	<b>48C</b> $\rightleftharpoons$ <b>48A</b>	7	0.80( $\pm$ 0.04)	-0.97( $\pm$ 0.03)	0.994
	<b>48(B+C)</b> $\rightleftharpoons$ <b>48A</b>	7	0.65( $\pm$ 0.03)	0.64( $\pm$ 0.03)	0.993

<sup>a</sup>Standard errors in parentheses.

## C. Publications

- I. **Márta Juhász**, Olli Martiskainen, Zita Zalán, Ferenc Fülöp, Kalevi Pihlaja  
Electron ionization mass spectra of phosphorus-containing heterocycles. I.  
1,4,4a,5,6,7,8,8a-Octahydro-2H-3,1,2-benzoxazaphosphinine 2-oxides  
*Rapid Commun. Mass Spectrom.* **2006**, *20*, 433-437. i.f.: 2.680
- II. Olli Martiskainen, **Márta Juhász**, Zita Zalán, Ferenc Fülöp, Kalevi Pihlaja  
Electron ionization mass spectra of phosphorus-containing heterocycles. II.  
1,2,3,4,4a,5,6,7,8,8a-Decahydro-1,3,2-benzodiazaphosphinine 2-oxides  
*Rapid Commun. Mass Spectrom.* **2006**, *20*, 1621-1627. 2.680
- III. **Márta Juhász**, Zita Zalán, Ferenc Fülöp, Kalevi Pihlaja  
Electron ionization mass spectra of phosphorus-containing heterocycles. III.  
1,3,4,2-Oxadiazaphosphinane 2-oxides  
*Rapid Commun. Mass Spectrom.* **2006**, *20*, 3595-3604. 2.680
- IV. **Márta Juhász**, László Lázár, Ferenc Fülöp  
Substituent effects in the ring-chain tautomerism of 4-alkyl-2-aryl substituted  
oxazolidines and tetrahydro-1,3-oxazines  
*J. Heterocycl. Chem.* **2007**, *44*, 1465-1473. 0.776
- V. **Márta Juhász**, Ferenc Fülöp, Kalevi Pihlaja  
Substituent effects on the gas-phase ring-chain tautomerism of 3,4,5,6-  
tetrahydro-2H-1,3-oxazines  
*Rapid Commun. Mass Spectrom.* **2007**, *21*, 3701-3710. 2.680
- VI. Kalevi Pihlaja, **Márta Juhász**, Henri Kivelä, Ferenc Fülöp  
Substituent effects on the ring-chain tautomerism of some 1,3-oxazolidine  
derivatives  
*Rapid Commun. Mass Spectrom.* accepted for publication. 2.680



## D. Conference lectures

- VII. **Juhász Márta**, Ovcharenko Vladimir, Zalán Zita, Pihlaja Kalevi, Fülöp Ferenc  
Néhány *O,N,P*-heterociklus fő fragmentációs útvonalainak tömegspektrometriás vizsgálata  
*MKE Vegyészkonferencia*  
Hajdúszoboszló, 2005. június 28-30., Abstr.: P-37.
- VIII. **Márta Juhász**, Zita Zalán, Ferenc Fülöp, Kalevi Pihlaja  
Mass spectrometric behaviour of 1,3,4,2-oxadiazaphosphinane 2-oxides under electron ionization  
*24th Informal Meeting on Mass Spectrometry*  
Ustron, Poland, 14-18 May, 2006, Abstr.: MoPo08.
- IX. Olli Martiskainen, **Márta Juhász**, Zita Zalán, Ferenc Fülöp, Kalevi Pihlaja  
Electron ionization mass spectra of phosphorus-containing heterocycles. 1,2,3,4,4a,5,6,7,8,8a-decahydro-1,3,2-benzodiazaphosphinine 2-oxides  
*24th Informal Meeting on Mass Spectrometry*  
Ustron, Poland, 14-18 May, 2006, Abstr.: MoPo18.
- X. **Márta Juhász**, Olli Martiskainen, Zita Zalán, Ferenc Fülöp, Kalevi Pihlaja  
Effects of *N*- and *P*-substitutions on the fragmentations of some *P*-containing heterocycles under EI  
*17th International Mass Spectrometry Conference*,  
Prague, Czech Republic, 27 August–1 September, 2006, Abstr.: TuP-112.
- XI. **Márta Juhász**, László Lázár, Ferenc Fülöp, Kalevi Pihlaja  
Ring-chain tautomerism in 4-alkyl-2-aryl-1,3-oxazolidines  
*25th Informal Meeting on Mass Spectrometry*  
Nyíregyháza-Sóstó, 6-10 May, 2007, Abstr.: TuPo19.
- XII. **Juhász Márta**, Lázár László, Fülöp Ferenc  
Szubsztituenshatások vizsgálata 4-alkil-2-aryl-1,3-*O,N*-heterociklusok gyűrű-lánc tautomériájában  
*MKE Centenáriumi Vegyészkonferencia*  
Sopron, 2007. május 29-június 1., Abstr.: SZ-P-26.