

**University of Szeged**  
**Faculty of Pharmacy**  
**Institute of Pharmaceutical Technology and Regulatory Affairs**  
Head: Prof. Dr. Ildikó Csóka, Ph.D.

**Ph.D. Thesis**

**INHALABLE KETOPROFEN NANOCRYSTAL-BASED DRY  
POWDERS FOR LOCAL AND SYSTEMIC PULMONARY DELIVERY:  
A PATIENT-CENTRIC APPROACH**

By  
**Heba Banat**  
Pharmacist

Supervisor:  
**Prof. Dr. Ildikó Csóka**  
and  
**Prof. Dr. Rita Ambrus**

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## ABBREVIATIONS

A549	Adenocarcinomic human alveolar basal epithelial cells
Ac	Acinar region
ACI	Andersen cascade impactor
A	Surface area of the membrane
ANOVA	Analysis of Variance
API	Active pharmaceutical ingredient
APS	Aerodynamic Particle Sizer
AUC	Area under the curve
BBD	Box-Behnken factorial design
BCS	Biopharmaceutical Classification System
BH	Breath-hold
Br	Bronchial region
C	Control
C <sub>A</sub>	Concentration of the acceptor phase
C <sub>D</sub>	Concentration of the donor phase
CF	Cystic fibrosis
CFBE	Cystic fibrosis bronchial epithelial cells
CFTR	Cystic fibrosis transmembrane conductance regulator
CMAAs	Critical Material Attributes
C <sub>max</sub>	Maximum concentration
CPPs	Critical Process Parameters
CQAs	Critical Quality Attributes
COPD	Chronic obstructive pulmonary disease
COX-1	Cyclooxygenase-1 enzyme
COX-2	Cyclooxygenase-2 enzyme
D[0.1]	10% of the volume distribution is below this value
D[0.5]	50% of the volume distribution is below this value
D[0.9]	90% of the volume distribution is below this value
DL	Drug loading
DLS	Dynamic light scattering
DMSO	Dimethyl sulfoxide
DNA	Deoxyribonucleic acid

DPI	Dry powder inhaler
DSC	Differential scanning calorimetry
EE	Encapsulation efficiency
EELV	End-expiratory lung volume
EF	Emitted fraction
EMA	European Medicines Agency
ET	Extra-thoracic
EXH	Exhaled
F	Relative bioavailability
FBS	Fetal bovine serum
FDA	U. S. Food and Drug Administration
FPF	Fine particle fraction
$f_t$	Tissue availability
FT-IR	Fourier-Transform Infrared Spectroscopy
G	Tissue damping
GSD	Geometric standard deviation
H	Tissue elastance
HEPES	4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid
HPMC	Hydroxypropyl methylcellulose
ICH	The International Council for Harmonization of Technical Requirements for Pharmaceuticals for Human Use
IVIVC	<i>In vitro-in vivo</i> correlation
IL-6	Interleukin 6
IUPAC	International Union of Pure and Applied Chemistry
$k_e$	Elimination rate constant
KTP	Ketoprofen
LPS	Lipopolysaccharide
MCh	Methacholine
MEM	Minimum Essential Medium
MMAD	Mass median aerodynamic diameter
MRT	Mean residence time
$M_w$	Molecular weight
NCA	Nanocrystal agglomerate
NS	Nanosuspension

NSAID	Non-steroidal anti-inflammatory drug
NTA	Nanoparticle Tracking Analysis
OVA	Ovalbumin
$P_{app}$	Apparent permeability coefficient
PDI	Polydispersity index
PEEP	Positive end-expiratory pressure
PK	Pharmacokinetics
PM	Physical mixture
PS	Particle size
PVA	Polyvinyl alcohol
QbD	Quality by Design
qPCR	Quantitative polymerase chain reaction
QTPP	Quality Target Product Profile
$R_{aw}$	Airway resistance
RI	Refractive index
RNA	Ribonucleic acid
RPMI	Roswell Park Memorial Institute medium
S.D.	Standard deviation
SDS	Sodium dodecyl sulphate
SEM	Scanning electron microscopy
SSA	Specific surface area
t	Time
TEER	Transepithelial electrical resistance
U937	Promonocyte cells
$X_c$	Crystallinity index
XRPD	X-ray powder diffraction
ZP	Zeta potential
$ZrO_2$	Zirconium-dioxide
$\rho_b$	Bulk density
$\rho_t$	Tapped density

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## 1. Introduction

Pulmonary drug delivery offers a non-invasive route for both local and systemic therapy, leveraging the lungs' large surface area, rich vascularization, and avoidance of first-pass metabolism to enhance bioavailability. Dry powder inhalers (DPIs) are particularly advantageous due to their stability, propellant-free design, and ease of use [1,2]. While inhaled non-steroidal anti-inflammatory drugs (NSAIDs) have shown promise for treating local lung inflammation, their systemic applications remain underexplored, despite the potential for lower therapeutic doses and reduced systemic side effects compared to oral administration [3].

NSAIDs are widely used to treat inflammatory conditions, including cystic fibrosis (CF) and acute pain; however, oral administration is limited by gastrointestinal toxicity, poor solubility, and variable bioavailability. Pulmonary delivery could overcome these challenges, yet no inhaled NSAID has been translated into a marketed product [4–6]. Nanocrystals, as a carrier-free platform (typically 1–1000 nm), offer distinct advantages for pulmonary delivery, including enhanced dissolution, improved mucus penetration, and high drug loading [7]. However, their small size necessitates precise particle engineering to ensure aerodynamic diameters within the optimal range (1–5  $\mu\text{m}$ ) for lung deposition.

This work introduces two patient-centric ketoprofen (KTP)-based DPI formulations derived from a unified nanocrystal platform. The first combines KTP with the mucolytic agent mannitol for localized pulmonary therapy, targeting both inflammation and mucus clearance in diseases such as CF. The second leverages nanocrystal engineering for systemic NSAID delivery, aiming to improve bioavailability while minimizing gastrointestinal irritation. Both formulations are designed to simplify treatment regimens and enhance tolerability.

By integrating Quality by Design (QbD) principles with nanocrystal technology, this study advances the development of inhaled NSAIDs through tailored strategies for local and systemic indications. The approach not only addresses current therapeutic gaps but also establishes a framework for repurposing poorly soluble drugs via pulmonary delivery—aligning with regulatory priorities for patient-centric, innovative drug products.

## **2. Aim of the work**

This study proposes a patient-centric strategy for pulmonary drug delivery through the development of two distinct DPI systems derived from a unified KTP nanocrystal platform. One formulation targets local lung inflammation and mucus obstruction in CF, while the other is designed for systemic delivery to enhance bioavailability via the pulmonary route. The overarching aim is to tailor drug formulation to meet different therapeutic needs while aligning with patient expectations for rapid relief, reduced dosing burden, and improved tolerability. The specific objectives are as follows:

### **A. Nanocrystal dispersion development**

I. Formulate and optimize a stable KTP nanosuspension as a versatile intermediate for both local and systemic DPI products, ensuring suitability for downstream powder engineering.

### **B. Local pulmonary delivery (Anti-inflammatory and mucolytic combination):**

II. Establish a rational inhalable combination of KTP and mannitol guided by a QbD approach.

III. Develop the combined dry powder formulation using mini spray-drying with varying mannitol ratios to optimize the combination while maintaining KTP integrity.

IV. Assess the dual functionality of the formulation through evaluation of local efficacy, biocompatibility, and epithelial permeability.

### **C. Systemic pulmonary delivery (Nanocrystal-engineered DPI):**

V. Optimize the nano spray-drying process parameters, with a factorial design, to produce nanocrystal-engineered powders suitable for deep lung deposition and systemic absorption.

VI. Investigate the influence of excipients (mannitol and leucine) on aerosol performance, biocompatibility, and cellular diffusion.

VII. Assess the pharmacokinetic profile of the inhaled nanocrystal formulation compared to the oral route, alongside tolerability and safety in ovalbumin-sensitized rats.

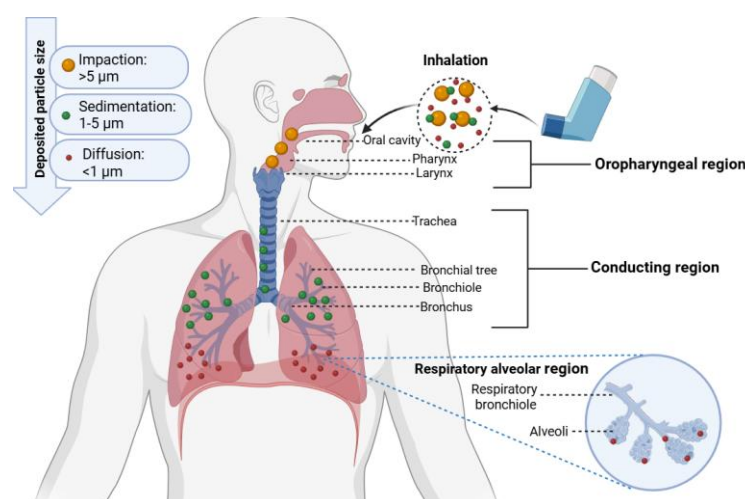
### **D. Stability evaluation:**

VIII. Determine the physicochemical stability of both DPI systems under accelerated storage conditions, ensuring maintenance of performance-critical attributes.

### 3. Literature background

#### 3.1. Pulmonary drug delivery: Local and systemic applications

The lungs provide a highly accessible and efficient route for drug administration due to their extensive surface area ( $\sim 100 \text{ m}^2$ ) and rich vascular network. The efficiency of drug delivery largely depends on where particles deposit within the respiratory tract. Particles with an aerodynamic diameter of  $<5 \mu\text{m}$  are optimal for reaching the lungs, with smaller particles penetrating deeper to the alveolar region, thereby enhancing systemic uptake [8–11]. As illustrated in Figure 1, deposition occurs through three principal mechanisms: inertial impaction ( $>5 \mu\text{m}$ ) in the upper airways, gravitational sedimentation ( $1\text{--}5 \mu\text{m}$ ) in smaller bronchioles, and Brownian diffusion ( $<1 \mu\text{m}$ ) in the alveoli [12]. While submicron particles can reach the deepest regions, they are also more likely to be exhaled [13,14]. In addition, particle morphology (e.g., corrugated or porous structures) and low particle density ( $<1 \text{ g/cm}^3$ ) are critical for improving dispersibility and achieving deep lung deposition [15–18].



**Figure 1:** An overview of the parts of the respiratory system and how particles deposit in the lungs based on the size. Adapted from [1], with slight modifications.

Pulmonary drug delivery can be classified into two main strategies: (i) local delivery for respiratory conditions, and (ii) systemic delivery of various drugs [19]. Although bronchodilators and corticosteroids remain the most common inhaled therapies for local lung targeting, other drug classes—such as antibiotics, anti-tuberculosis agents, anticancer drugs, and mucolytics—are gaining attention. For instance, inhaled antibiotics can achieve high local concentrations in the lungs, improving efficacy and minimizing systemic side effects, while potentially limiting resistance development by sparing non-target microbiota [20]. Table 1 summarizes recent developments in inhaled formulations for local pulmonary therapy.

**Table 1:** Inhaled products, marketed or in-developing, intended for local delivery [21–27].

Product	Indication	API	Advancement	Manufacturer	Status
Tobi® Podhaler	CF	Tobramycin	PulmoSphere®	Viartis	Marketed- 2013
Trixeo Aerosphere	COPD	Budesonide, Glycopy- ronium, Formetrol fumerate	Aerosphere™	AstraZeneca	Marketed- 2020
Ohtuvayre®	COPD	Ensifentrine	Unique MoC (Dual PDE3/4 inhibitor)	Verona Pharma	Marketed- 2024
RT234-PAH	PAH	Vardenafil	AOST™ DPI	Respira Therap- eutics	Phase II- 2024
AP01	IPF	Pirfenidone	Novel inhaled product (details undisclosed)	Avalyn Pharma	Phase II- 2025
AP02	IPF	Nintedanib	Novel inhaled product (details undisclosed)	Avalyn Pharma	Phase I- 2024

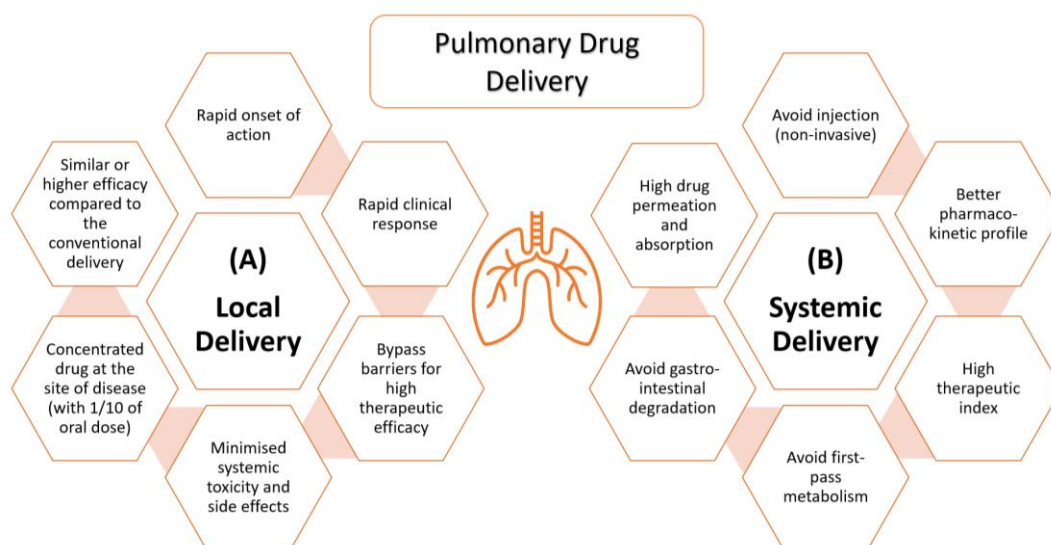
CF: cystic fibrosis, COPD: chronic obstructive pulmonary disease, PDE: phosphodiesterase, PAH: pulmonary arterial hypertension, IPF: idiopathic pulmonary fibrosis, MoC: mechanism of action.

Although pulmonary delivery is well established for local treatment, achieving efficient systemic absorption remains challenging. Success relies on deep lung deposition, where high surface area and vascularization enable optimal uptake, unlike the upper airways with limited absorption and rapid clearance [28]. Nonetheless, progress has been made, with several inhaled products reaching clinical development or market approval. Table 2 highlights key examples, demonstrating the growing interest in this non-invasive route.

**Table 2:** Inhaled products, marketed or in-developing, intended for systemic delivery [1,29–31].

Product	Indication	API	Manufacturer	Status
Exubera®	Diabetes	Insulin	Pfizer, Nektar Therapeutics	Withdrawn- 2007
Afrezza®	Diabetes	Insulin	MannKind Corporation	Marketed- 2014
AIR® insulin	Diabetes	Insulin	Lilly	Withdrawn- 2007
Inbrija®	Parkinson's disease	Levodopa	Merz Therapeutics	Marketed- 2018
Adasuve®	Schizophrenia	Loxapine	Alexza Pharmaceuticals	Marketed- 2012
Levadex®	Migraine	Dihydro- ergotamine	MAP Pharmaceuticals	Phase III- 2014 (declined)
CVT427	Migraine	Zolmitriptan	Acorda Therapeutics	Phase I- 2016

Several pulmonary barriers—such as mucociliary clearance, alveolar macrophages, and enzymatic degradation—limit drug delivery and reduce bioavailability [32,33]. Addressing these challenges is critical in the design of effective inhalable formulations. Despite such limitations, the substantial advantages of pulmonary drug delivery—such as rapid onset, reduced systemic side effects, and lower dosing—support the continued development of inhalable therapies for both local and systemic applications. These benefits are summarised in Figure 2.



**Figure 2:** Advantages of local and systemic pulmonary drug delivery [1].

### 3.2. Inhaled non-steroidal anti-inflammatory drugs

NSAIDs are increasingly recognized for their role in managing chronic respiratory conditions such as CF, where persistent neutrophilic inflammation drives lung damage. High-dose oral ibuprofen has shown efficacy in slowing lung function decline in CF and is included in treatment guidelines [34–38]. However, long-term systemic use is limited by significant side effects. Despite their therapeutic value, no inhaled NSAID formulations are currently approved for CF or other pulmonary conditions, highlighting a gap for targeted pulmonary delivery to enhance efficacy and reduce systemic risks.

Beyond respiratory diseases, NSAIDs remain among the most widely used drugs globally, indicated for pain, fever, and inflammation in conditions like arthritis and migraines [39–44]. Their mechanism—primarily inhibition of cyclooxygenase (COX) enzymes—offers broad therapeutic benefits but is offset by gastrointestinal, cardiovascular, and renal adverse effects [45–48]. Additionally, poor water solubility hinders formulation and bioavailability [49].



Inhaled NSAIDs, such as ibuprofen and meloxicam, have been explored for local pulmonary action [4–6], but their systemic potential remains underutilized. Preclinical data suggest that pulmonary administration of NSAIDs can achieve therapeutic effects at lower doses than oral routes [50], offering a promising alternative with fewer side effects. Given regulatory recommendations to minimize NSAID exposure, alternative delivery strategies like inhalation are urgently needed.

### **3.3. Ketoprofen as a candidate for inhaled delivery**

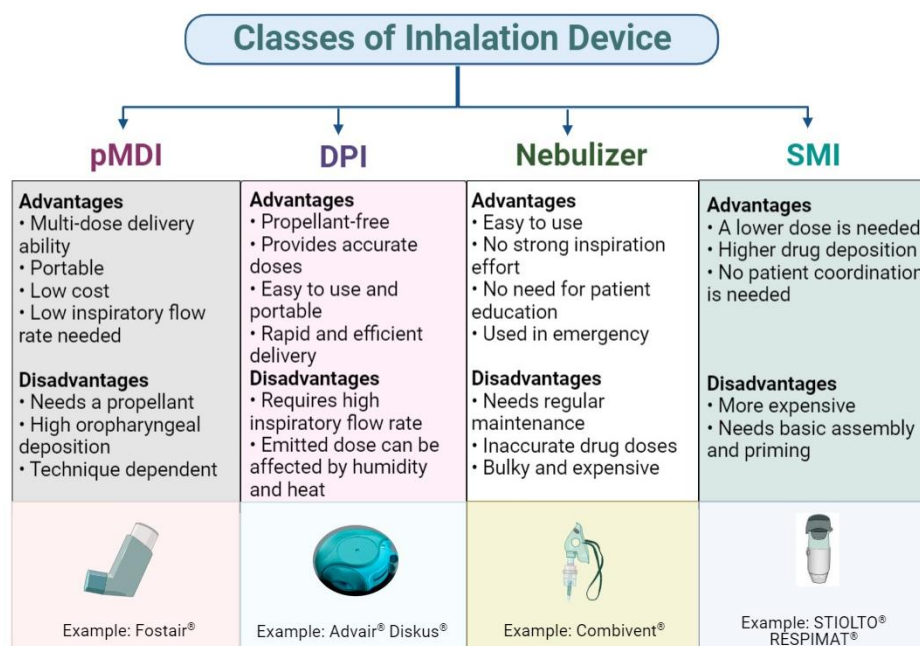
KTP is a potent NSAID with antipyretic, analgesic, and anti-inflammatory properties. Compared to other NSAIDs, it offers superior analgesic efficacy—particularly in inflammatory and postoperative pain—with a favorable safety profile. Clinical and preclinical studies support its dual anti-inflammatory and antinociceptive effects, including inhibition of edema, IL-8 suppression, and enhanced antioxidant activity [51,52].

Meta-analyses further show KTP outperforms ibuprofen in managing rheumatoid arthritis and provides better outcomes in moderate-to-severe pain while maintaining tolerability [53–55]. Topical formulations, such as KTP plasters, have shown comparable efficacy to diclofenac with good safety in osteoarthritis [56]. Despite its therapeutic potential, KTP remains underexplored for pulmonary delivery, with only limited investigations [57], indicating a promising area for further development.

### **3.4. Dry powder inhalers: Formulation, advantages and challenges**

A critical determinant of therapeutic efficacy in pulmonary drug delivery is the inhalation device, which ensures that the drug is effectively deposited within the bronchial tree, with potential for deep lung penetration [58,59]. Among the various pulmonary delivery devices shown in Figure 3, DPIs have gained increasing preference in pulmonary delivery due to their solid-state stability, absence of harmful propellants, user-friendly design, and breath-actuated mechanism, which removes the need for patient coordination during inhalation [29].

Unlike nebulizers and pressurized metered dose inhalers (pMDIs), which rely on external power sources or propellants, DPIs are portable and environmentally friendly. Their compatibility with engineered particles, makes them particularly suitable for modern inhalation therapies targeting both local and systemic effects.



**Figure 3:** Illustration of devices used for inhaled products, alongside examples of selected marketed brands. Abbreviation: pMDI: pressurized metered dose inhaler, DPI: dry powder inhaler, SMI: soft mist inhaler [1].

DPIs are generally formulated using one of two strategies: (i) carrier-based systems, where the drug is blended with larger excipient particles, and (ii) carrier-free systems, which contain only the drug particles, often engineered using advanced techniques and excipients to enhance aerodynamic properties. In carrier-based DPIs, the active pharmaceutical ingredient (API) adheres to coarse carrier particles—typically lactose, mannitol, or glucose—ranging from 50 to 100  $\mu\text{m}$  in size [60]. However, during inhalation, drug detachment from the carrier is often incomplete, resulting in reduced deposition of the drug in the lower lungs and accumulation in the upper airways. This inefficiency underscores the importance of optimizing aerosolization and dispersion performance. In light of these limitations, carrier-free DPIs have emerged as a promising alternative, offering greater potential for efficient lung delivery and improved therapeutic outcomes.

Carrier-free DPI formulations aim to reduce interparticle cohesion, enhancing dispersion and improving drug release efficiency from the inhaler. These systems typically involve engineering complex powders by combining the API with functional excipients to optimize aerodynamic properties and lung deposition [61–63]. Additives such as hydrophobic agents, lipids, amino acids, and biodegradable polymers have been explored to enhance aerosolization and stability [64,65]. However, pulmonary delivery imposes stricter safety requirements, and only a limited number of excipients are approved for inhalation due to

potential risks of accumulation and irritation in lung tissue [29,64,66,67]. This restricts formulation options and complicates the development process.

Building on these advances, nanoscale drug delivery systems—such as nanoparticles and nanocrystals—offer new opportunities to further improve pulmonary drug delivery by enhancing deposition, dissolution, and systemic absorption. These technologies are discussed in the following sections.

### **3.5. Impact of nanotechnology on pulmonary drug delivery**

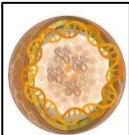
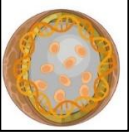
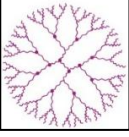
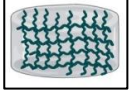
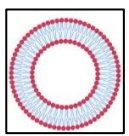
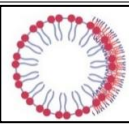
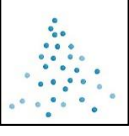
Nanotechnology has revolutionized drug delivery, including inhalation therapies. In pulmonary administration, nanoparticles improve the solubility and bioavailability of poorly soluble drugs, enabling enhanced therapeutic efficacy at lower doses and potentially minimizing systemic side effects [68,69]. They also help overcome biological barriers within the lungs. Notably, particles smaller than 200 nm can evade clearance by alveolar macrophages, promoting deeper lung deposition and prolonged retention [70,71]. Additionally, nanoparticles can be tailored through surface modifications to improve critical aspects of pulmonary delivery, such as enhancing cellular uptake [72].

Nanoparticles used in inhalation therapy are typically classified as carrier-based or carrier-free, depending on their composition. Among carrier-based systems, lipid-based nanoparticles—including liposomes, solid lipid nanoparticles, and nanoemulsions—are widely applied for their capacity to encapsulate both hydrophilic and hydrophobic drugs [73,74]. Polymeric nanoparticles, such as those made from PLGA or chitosan, offer benefits like biodegradability, mucoadhesiveness, and extended lung retention [75–78]. Inorganic nanoparticles, such as mesoporous silica, provide high drug-loading capacity and modifiable surfaces, contributing to targeted delivery and controlled release [79]. Hybrid nanoparticles, which combine materials (e.g., lipids with polymers), are engineered to harness the advantages of multiple platforms and further optimize drug delivery [80]. Each nanoparticle class presents distinct benefits, supporting a broad range of therapeutic and diagnostic applications in pulmonary medicine.

In contrast, nanocrystals represent a carrier-free nanotechnology platform that has attracted growing interest for inhalation delivery. These are pure drug particles, typically 1–1000 nm in size, stabilized by surfactants or polymers [81,82]. Nanocrystals enhance the dissolution rate and saturation solubility of poorly water-soluble drugs while maintaining favorable biological compatibility and low toxicity. Their high surface area improves bioavailability

and facilitates better drug adhesion to lung tissues. Compared to carrier-based systems, nanocrystals offer advantages such as higher drug loading, improved mucus penetration, reduced macrophage clearance, and efficient drug release—making them especially promising for DPI formulations [7]. Table 3 summarizes selected nanoparticle platforms under investigation for pulmonary delivery.

**Table 3:** An overview of nanoparticle systems used in pulmonary drug delivery [1].

Nanosystem	Preparation techniques	Example	Reference
 Nanosphere	Nanoprecipitation Ionic gelation	Combination of rifampicin/isoniazid nanosphere to target macrophage using solvent injection method.	[83]
 Nanocapsule	Solvent displacement Double emulsion Emulsion diffusion	Combination of fluticasone propionate/salmeterol nanocapsule using interfacial polycondensation based on L-lysine.	[84]
 Dendrimers	Microfluidic process Ionic gelation Cross-linking strategy	Combination of nucleic acid and anti-cancer loaded in dendrimers carrier.	[85]
 Nanogel	Polymerization Precipitation	Combination of graphene oxide -hybridized nanogels for combinative anticancer therapy.	[86]
 Liposome	Then film hydration Reverse phase evaporation Solvent injection Cross flow injection Membrane contractor Supercritical fluid Microfluidics	Synergistic combinations of levofloxacin/serratiopeptidase to improve antimicrobial and antibiofilm activity in treating <i>S. aureus</i> infection.	[87]
 Lipid nanoparticles	High shear homogenisation Microemulsion Solvent emulsification	Combination of afatinib/paclitaxel loaded lipid nanoparticle in porous microsphere for the treatment of EGFR TKIs resistant NSCLC.	[88]
 Nanocrystals	A. Top-down techniques Media milling High-pressure homogenisation B. Bottom-up techniques Nanoprecipitation	Combination of doxorubicin/ellagic acid nanocrystals using spray dryer to target lung cancer cells.	[89]

### **3.6. Nanocrystalline suspension: Preparation and stabilization**

#### **3.6.1. Production of nanosuspension by wet media milling**

The top-down approach is commonly used to produce drug nanocrystals by mechanically reducing larger particles to the nanoscale. Wet media milling is a cost-effective and scalable nanosizing technique that avoids the use of organic solvents, valued for its simplicity and reproducibility. It enables the preparation of stable nanosuspensions of poorly water-soluble APIs. Key factors affecting wet media milling include milling speed, duration, bead size, and filling volume, while stabilizer type and concentration are critical for preventing uncontrolled agglomeration and achieving optimal particle size [90–92].

#### **3.6.2. Stabilization of nanosuspension**

Stabilizers are indispensable in wet media milling process to prevent particle aggregation and maintain long-term stability [93]. They function by adsorbing onto the surface of drug nanoparticles and providing steric or electrostatic repulsion [94]. Commonly used stabilizers include polymers such as polyvinylpyrrolidone (PVP), hydroxypropyl methylcellulose (HPMC), hydroxypropyl cellulose (HPC), polyethylene glycol (PEG), polyvinyl alcohol (PVA), and poloxamer 188. Surfactants also play a crucial role: ionic surfactants like sodium dodecyl sulfate (SDS), sodium lauryl sulfate (SLS), poly(ethylene imine) (PEI), and chitosan stabilize via electrostatic interactions, while nonionic surfactants such as polysorbates (Tweens) and Pluronics<sup>®</sup>, act primarily through steric hindrance [95]. The choice and concentration of stabilizer must be carefully optimized, as they significantly influence particle size, surface properties, and suspension stability. Moreover, the stabilizer must be suitable for pulmonary administration, making biocompatibility and regulatory acceptance critical considerations [96].

### **3.7. Nanocrystal-based dry powder inhalers**

#### **3.7.1. Particle-engineered dry powder using spray drying**

Particle engineering is a versatile and essential approach in the development of DPIs, enabling control over the surface characteristics of drugs, carriers, or excipients [97]. This technique helps balance interparticle forces, enhance stability during processing and storage, and improve aerodynamic performance [98]. As previously discussed, achieving an aerodynamic diameter of 1–5  $\mu\text{m}$  is critical for efficient deposition in the lower respiratory tract [99]. While nanoparticles such as nanocrystals are favorable for deep lung penetration due to their small size, they face a high risk of being exhaled before deposition can occur [100–102]. Therefore, nanocrystals often require particle engineering strategies

to increase their aerodynamic diameter into the respirable range, ensuring effective lung deposition.

Among various particle engineering techniques, spray drying stands out for producing inhalable powders with optimized characteristics. This method allows fine-tuning of critical parameters such as drying temperature, feed rate, solvent composition, and atomization settings, making it highly adaptable for tailoring particle size, morphology, flowability, and moisture content [103,104]. In a single-step process, spray drying converts liquid formulations—such as solutions, suspensions, or emulsions—into dry powders suitable for inhalation [105].

Nano spray drying, a more recent advancement, offers enhanced control over droplet formation, producing highly uniform particles with narrow size distributions. Unlike traditional spray dryers that use pressure or centrifugal atomization, nano spray dryers utilize a vibrating mesh nebulizer to create fine and consistent microdroplets. This enables efficient powder production from small-volume samples and results in higher yields [106,107]. Its ability to precisely control particle size and morphology makes it particularly suited for pulmonary applications requiring deep alveolar deposition. However, it is important to carefully choose the excipients during spray drying, as they play a critical role in engineering the size and shape of nanocrystals and achieving desirable aerosol performance.

### **3.7.2. Role of excipients in spray drying**

The selection of excipients during spray drying significantly influences the physicochemical properties of the final powder. Common excipients, such as mannitol, trehalose, and leucine, play a critical role in stabilizing formulations during drying and ensuring product stability. Mannitol is commonly used as a cryoprotectant, bulking agent, preserving structural integrity and enhancing formulation stability [108]. However, its tendency to crystallize can impact aerosol performance and fine particle fraction (FPF) upon storage [109]. In contrast, leucine improves powder dispersibility and stability in dry powder formulations by reducing particle cohesion and enhancing aerosolization [110]. Since excipients in spray drying serve multiple roles, including thermal protection, aggregation prevention, and maintaining API bioactivity, the ideal spray-dried preparations may necessitate a combination of various excipients [111]. Moreover, their influence on

spray-dried powder should be carefully evaluated to ensure optimal characteristics and to achieve the desired powder properties.

### **3.8. Regulatory considerations and the QbD framework**

Regulatory authorities such as the U.S. Food and Drug Administration (FDA) and the European Medicines Agency (EMA) impose strict requirements to ensure the safety, efficacy, and consistency of pulmonary drug delivery systems, especially DPIs [112]. Inhaled products require thorough characterization, including aerodynamic particle size distribution, emitted dose uniformity, and moisture sensitivity [1]. For advanced systems such as nanocrystal-based DPIs, stability remains a particular concern, as hygroscopicity, particle aggregation, and morphological changes can compromise delivery efficiency [113]. Preclinical studies are essential to assess lung toxicity and local tolerability, followed by clinical trials to establish therapeutic efficacy.

To support regulatory compliance and streamline development, the QbD framework offers a risk-based, science-driven strategy for product optimization [114]. The process begins with a pre-formulation or “Zero Phase”, where unmet therapeutic needs and key characteristics of the drug–dosage form–delivery route triangle are evaluated. This phase lays the foundation for defining a suitable design space [115].

Subsequent QbD stages involve identifying the Quality Target Product Profile (QTPP), Critical Quality Attributes (CQAs), Critical Process Parameters (CPPs), and Critical Material Attributes (CMAs) [116–118]. This structured methodology ensures that development objectives, materials, and manufacturing processes align with both regulatory expectations and clinical performance. For inhaled products, QbD can guide decisions related to aerosolization efficiency, powder flowability, and formulation-device compatibility—ultimately supporting cost- and time-efficient scale-up [119].

In addition to facilitating smoother regulatory approval, integrating QbD strengthens product design by balancing patient-centric needs with industrial feasibility and regulatory standards. This is especially valuable in the context of next-generation inhalation therapies, where personalized treatment approaches are becoming increasingly important—the focus of the following section.

### **3.9. Patient-centric pulmonary drug delivery systems**

Patient-centricity in pharmaceutical development means designing therapies to meet real-world patient needs, including efficacy, ease of use, tolerability, and convenience. Patient-

centricity is a relatively new term as applied to the relationship between quality attribute levels and the impact on safety and efficacy [120]. While not classified as such until recently, a patient-centric approach has been shaping industry development strategies for more than two decades. Inhaled drug delivery—especially for chronic respiratory diseases—requires thoughtful consideration of user preferences to ensure long-term adherence [121]. This concept now extends beyond patient interviews and clinical trials to include the formulation of drug products that integrate usability, safety, adherence, and overall patient experience. By involving both patient needs and therapeutic outcomes early in the development process, inhalation therapies can be optimized to improve treatment satisfaction and disease management [64,112]. The transition to patient-centricity is a major step toward driving patient engagement and satisfaction, bringing positive direction to drug development and medical care.

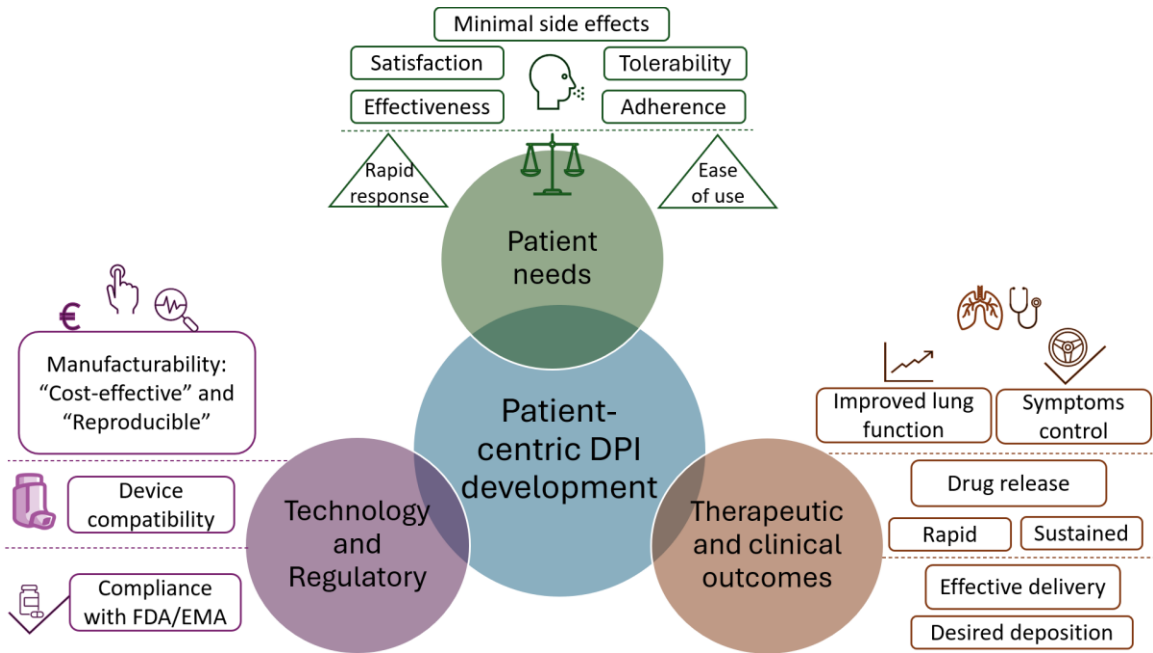
One practical patient-centric strategy is the development of combination inhalers, which deliver two or more therapeutic agents within a single device. Originally introduced in the mid-1990s, these systems were designed to enhance asthma and chronic obstructive pulmonary disease (COPD) management by simplifying treatment regimens [122–124]. Patients often favor drugs that provide fast symptom relief (e.g., bronchodilators), but long-term control requires anti-inflammatory agents like corticosteroids [125]. Administering both in a single inhaler reduces dosing complexity, minimizes training errors, and increases compliance—particularly important given that poor adherence is a leading cause of uncontrolled respiratory disease [124,126–128]. Furthermore, patients often gauge treatment effectiveness by how quickly they feel relief [129,130]. Thus, combined inhalers—designed to provide both rapid relief and long-term control—better align with patient behavior and therapeutic goals. However, this strategy has primarily been applied to asthma and COPD, highlighting a broader opportunity to extend combination inhaler approaches to other respiratory conditions.

Another key strategy in patient-centric drug delivery is enhancing the bioavailability of inhaled drugs, which can lead to faster therapeutic effects, fewer side effects, and improved patient satisfaction. The pulmonary route offers notable advantages for systemic drug delivery, including bypassing first-pass metabolism and enabling rapid absorption—particularly beneficial for compounds with poor oral bioavailability (as discussed in Section 3.1).



Regulatory authorities, such as the FDA, have shifted toward science- and risk-based frameworks that prioritize patient outcomes over rigid compliance checklists [120]. For instance, guidance like SUPAC-IR emphasizes physiologically relevant *in vitro* dissolution testing as a surrogate for *in vivo* bioavailability, especially for highly soluble drugs. This aligns with the broader adoption of Patient-Centric Quality Standards (PCQS), where clinical relevance, therapeutic performance, and real-world patient variability take precedence [131,132]. Within this framework, optimizing dissolution is not just a manufacturing requirement but a critical step in developing effective pulmonary drug products. By improving systemic bioavailability, these innovations can accelerate therapeutic action, minimize side effects, and ultimately enhance patient adherence—fulfilling the core objectives of patient-centric delivery.

Translating these frameworks to novel DPI designs—particularly for non-traditional indications—demands a departure from conventional development paradigms. Central to this evolution is the synergy between particle engineering, device design, and patient behavior, where rapid symptom relief remains a frequently neglected priority in chronic disease therapeutics. As Figure 4 demonstrates, successful patient-centric DPI development emerges from the integration of clinical needs, technical feasibility, and real-world usability—a triad that unlocks new potential for pulmonary delivery, particularly in indications extending beyond classical respiratory applications.



**Figure 4:** A conceptual framework illustrating the three core pillars of patient-centered DPI development.

## 4. Materials

### 4.1. Active pharmaceutical ingredient

KTP, with the IUPAC name *2-(3-benzoylphenyl)propanoic acid*, was used as the primary API (TCI Chemicals, Shanghai, China). Its chemical structure is shown in Figure 5. KTP is an NSAID classified under Biopharmaceutical Classification System (BCS) class II. This class indicates its poor water solubility (51 mg/L at 22 °C) and high permeability (log P = 3.12). It is a propionic acid derivative with a molecular weight ( $M_w$ ) of 254.28 g/mol, a melting point of 94 °C, and a pKa of 4 [133].

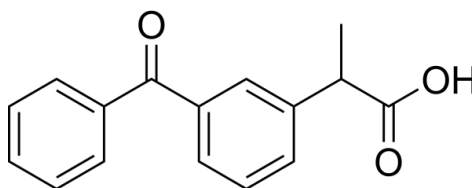
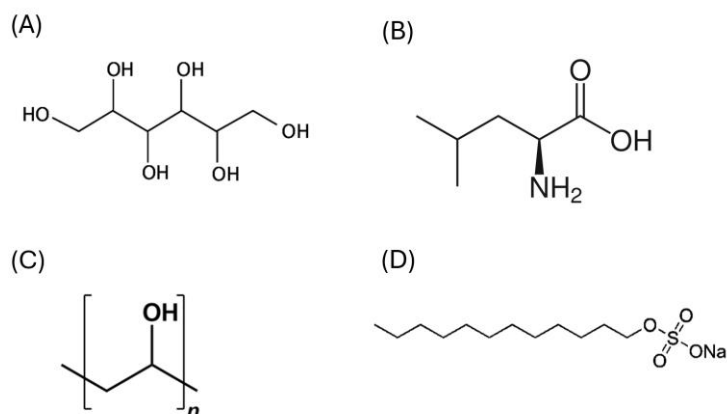


Figure 5: KTP chemical structure [133].

### 4.2. Excipients

PVA ( $M_w$  ~27,000 g/mol; Aldrich Chemistry, Darmstadt, Germany) was employed as a steric stabilizer during the wet media milling process. It adsorbs onto nanoparticle surfaces, providing steric hindrance to prevent uncontrolled aggregation [134,135]. SDS ( $M_w$  288.38 g/mol; VWR Chemicals, Leuven, Belgium) was utilized as a co-stabilizer in the nanosuspension preparation. It acted as an electrostatic stabilizer by imparting surface charge, thereby inducing repulsive forces between particles and enhancing colloidal stability [136–138]. The combination of steric (PVA) and electrostatic (SDS) stabilizers helps maintain nanosuspension stability [96].

D-mannitol ( $M_w$  182.17 g/mol; Molar Chemicals Ltd., Budapest, Hungary) was incorporated as a secondary API in the spray-dried formulation intended for local pulmonary delivery, owing to its muco-active properties. Mannitol is used as a mucolytic agent (Marketed brand name Bronchitol<sup>®</sup>), particularly as an add-on therapy for CF [139]. Additionally, its role as a formulation excipient was evaluated in the nano spray-dried formulations intended for systemic delivery. L-leucine ( $M_w$  131.17 g/mol; AppliChem GmbH, Darmstadt, Germany) was included as a dispersibility enhancer during spray drying. Leucine is well recognized in pulmonary drug delivery for its ability to improve powder flow properties and aerosolization performance [140,141]. Chemical structures of excipients used are illustrated in Figure 6.



**Figure 6:** Chemical structures of (A) mannitol [142], (B) leucine [143], (C) PVA [144] and (D) SDS [145].

## 5. Methods

### 5.1. Production of the nanosuspension by wet media milling

A KTP-containing nanosuspension (KTP-NS) was developed using a two-step process. First, KTP particles were dispersed in an aqueous stabilizing solution and homogenized using an Ultra-Turrax homogenizer (T-25, IKA-Werke, Germany) at 17,000 rpm for 10 minutes. The resulting pre-suspension was then subjected to wet media milling using a planetary ball mill (Retsch PM 100 MA, Retsch GmbH, Haan, Germany). Milling was performed with 20.00 g of 0.3 mm zirconium dioxide ( $\text{ZrO}_2$ ) beads at 400 rpm for 90 minutes in a 50 mL milling chamber. To avoid overheating, a 5-minute cooling pause was implemented every 15 minutes during the milling process. The final drug concentration in the nanosuspension was 10% (w/v), and samples were diluted as required for further analysis or processing.

#### 5.1.1. Selection of the stabilizers

Three different stabilizers were evaluated for their ability to produce stable KTP-NS, each tested at three concentrations: i) HPMC at 0.25, 0.50, and 1.00% (w/v); ii) PVA at 1.00, 2.50, and 5.50% (w/v); iii) Poloxamer 188 at 0.20, 0.50, and 1.00% (w/v). SDS was included in all formulations at a fixed concentration of 0.10% (w/v) to enhance electrostatic stability during the milling process. The resulting nanosuspensions were characterized in terms of particle size, and the most effective stabilizer was selected for further investigations.

#### 5.1.2. Particle size analysis of the nanosuspension

##### 5.1.2.1. Dynamic light scattering

Particle size (PS), polydispersity index (PDI), and zeta potential (ZP) of KTP-NS were measured using a Zetasizer Nano ZS (Malvern Instruments, UK). The refractive index was

set to 1.592. Measurements were performed at 25 °C using disposable folded capillary cells (DTS1070). Each sample was analyzed in triplicate, and the average values were reported.

#### **5.1.2.2. Nanoparticle tracking analysis**

The particle size distribution of KTP-NS was determined using the NanoSight NS3000 (Malvern Instruments, UK), equipped with a 565 nm laser, sCMOS camera, and syringe pump (speed: 50). Samples were diluted and analyzed in script control mode at the refractive index for KTP set to 1.592, with three 30-second videos (1500 frames total). Particle size metrics included D[0.1], D[0.5], and D[0.9], while the size distribution span was calculated according to Equation 1.

$$\text{Span} = \frac{D[0.9] - D[0.1]}{D[0.5]} \quad (1)$$

#### **5.1.3. Stability of the nanosuspension**

To assess the stability of the KTP-NS prior to the drying process, a short-term stability study was conducted. The nanosuspension was stored at refrigeration temperature (+4 °C) and evaluated for particle size and distribution at predefined time intervals.

### **5.2. Inhalable ketoprofen-mannitol combination powder: Local delivery**

#### **5.2.1. Application of QbD to define the rationale for novel powder combinations**

The implementation of QbD enables a rational and systematic formulation strategy to ensure high-quality inhaled therapies from early development [114]. In this study, the QbD framework was applied during the pre-formulation design (Zero Phase) to rationally select and justify the components of the novel inhalable combination [115]. Risk assessment and process parameter analysis were not within the scope of this work. The process began with structured brainstorming to identify the needs, main goals and critical steps in manufacturing. A flowchart was used to outline key challenges in CF and highlight the need for combination inhaled therapies. To further identify factors affecting product quality, an Ishikawa (fishbone) diagram was developed, helping to pinpoint variables influencing CQAs. Based on this analysis, a QTPP was defined, outlining the essential attributes for safety, efficacy, and performance, specifically tailored to inhaled combination therapies for CF.

#### **5.2.2. Formulation of combined DPI using mini spray drying**

To achieve local pulmonary targeting, combination powders were prepared using Mini Spray Dryer B-19 (Büchi, Flawil, Switzerland). Spray drying parameters were optimized based on preliminary trials and set as follows: inlet temperature, 90 °C; aspirator setting,

80%; pump rate, 2%; and air flow rate, 600 L/h. For powder preparation, the KTP-NS was diluted to 1% (w/v), and mannitol—used for its mucolytic properties—was incorporated at varying concentrations to determine the highest amount that could be included without adversely affecting the integrity of KTP. Four spray-dried formulations were prepared with mannitol-to-KTP mass ratios of 0, 0.5:1, 1:1, and 2:1, and were named F0, F0.5, F1, and F2 according to their mannitol content. Additionally, a dispersity enhancer (leucine) was included in all formulations at a fixed mass ratio of 1:1 (leucine:KTP) [146].

### 5.3. Inhalable ketoprofen nanocrystal-based dry powder: Systemic delivery

#### 5.3.1. Optimization of nano spray-drying parameters

Nano Spray Dryer B-90 HP (Büchi, Flawil, Switzerland) equipped with a medium nebulizer was used to engineer KTP-containing nanocrystal agglomerates (NCAs) for targeting systemic effect. The Box-Behnken factorial design (BBD) was employed to optimize the drying parameters, assisted by TIBCO Statistica® 13.4 (Statsoft Hungary, Budapest, Hungary) for experimental design generation. Three independent factors were chosen: pump rate (%) ( $X_1$ ), inlet temperature (°C) ( $X_2$ ), and feed concentration (% w/v) ( $X_3$ ). The effect of these factors on the dependent variables (D[0.5], span, and yield percentage) was investigated at three levels. Temperature was studied at 90, 100, and 110 °C, feed concentration was evaluated at 10%, 5% and 2.5% (w/v), while pump percentage was assessed at 10%, 30%, and 50% [147].

#### 5.3.2. Incorporation of mannitol and/or leucine as excipients

Using optimized nano spray dryer parameters, we evaluated the impact of excipients—mannitol (K1M), leucine (K1L), their combination (K1ML), and the absence of excipients (K1)—on the properties of NCAs for pulmonary delivery. The formulations were prepared as detailed in Table 4, mannitol and/or leucine were added accordingly while ensuring a constant total concentration [3,147].

**Table 4:** Samples annotation, composition and concentration\*.

Sample	KTP (% w/v)	Mannitol (% w/v)	Leucine (% w/v)
K1	5.00	0	0
K1M	2.50	2.50	0
K1L	2.50	0	2.50
K1ML	2.50	1.25	1.25

\*The concentrations of the samples were maintained to meet the optimized feed concentration of 5% w/v, as determined by the factorial design study.

## 5.4. Powder characterization

### 5.4.1. Particle size

#### 5.4.1.1. *Dynamic light scattering*

Spray-dried powders were re-dispersed in deionized water and analyzed for PS, PDI, and ZP using a Zetasizer Nano ZS (Malvern Instruments, UK), as described in Section 5.1.2.1.

#### 5.4.1.2. *Laser diffraction*

Laser diffraction was used to assess the particle size and distribution of the spray-dried samples with a Mastersizer Scirocco 2000 device (Malvern Instruments Ltd., UK). A dry dispersion unit was employed with 0.5–1.0 g of powder placed in the feeding tray. The air pressure was adjusted to 3.0 bar with a vibration feed of 75%. Triplicate measurements were performed for each sample. Parameters including D[0.5], and specific surface area (SSA) were evaluated, and the span was calculated as previously defined (Equation 1).

### 5.4.2. Density measurements

Spray-dried samples were evaluated for tapped and bulk densities using an Engelsmann Stampfvolumeter (Ludwigshafen, Germany). The bulk density ( $\rho_b$ ) was determined by accurately weighing the samples and filling them into a 10 cm<sup>3</sup> graduated cylinder, then calculating the mass-to-untapped volume ratio ( $m/v_0$ ). The tapped density ( $\rho_t$ ) was measured after 1250 taps and calculated by dividing the mass by the tapped volume ( $m/v_f$ ).

### 5.4.3. Drug loading, encapsulation efficiency and yield

To calculate drug loading (DL) and encapsulation efficiency (EE), a measured amount of the spray-dried samples was dissolved in 50% (v/v) methanol and analyzed by UV/VIS Spectrophotometer (ATI-Unicam, Cambridge, UK) at 258 nm (Equations 2 and 3). Moreover, percentage yields of dry powder collected after spray drying was calculated (Equation 4).

$$DL (\%) = \frac{\text{Weight of the drug}_{(\text{measured})}}{\text{Total weight of the dry powder}_{(\text{drug} + \text{excipients})}} \times 100\% \quad (2)$$

$$EE (\%) = \frac{\text{Weight of the drug}_{(\text{measured})}}{\text{Initial weight of the drug added}} \times 100\% \quad (3)$$

$$\text{Yield} (\%) = \frac{\text{Weight of dry powder collected}_{(\text{after spray drying})}}{\text{Weight of the initial solid composition}_{(\text{before spray drying})}} \times 100\% \quad (4)$$

### 5.4.4. Morphology

The morphology was analysed using a scanning electron microscope (SEM) (Hitachi Scientific Ltd. in Tokyo, Japan). Operating at 10 kV, a sputter coater from Bio-Rad coated

the samples with a gold-palladium layer using 2.0 kV electric potential, 10 mA current, and a 10-minute duration. Air pressure was maintained between 1.3 and 13.0 mPa. Also, the particle size was calculated using ImageJ 1.53e software, with measurements taken from 50 to 100 particles per sample.

#### 5.4.5. Solid-state analysis

Differential Scanning Calorimetry (DSC) and Thermogravimetric Analysis (TGA) were performed using Mettler Toledo DSC 821e and TGA/DSC1 systems, respectively, equipped with STARe software v9.3 (Mettler Inc., Schwerzenbach, Switzerland). Analyses were conducted from 25 °C to 300 °C at 10 °C/min under an argon flow (10 L/h). TGA was coupled to a ThermoStar™ GSD 320 quadrupole mass spectrometer (Pfeiffer Vacuum) for residual water analysis.

X-ray Powder Diffraction (XRPD) was carried out using a BRUKER D8 Advance diffractometer (Bruker AXS GmbH, Germany) with Cu K $\alpha$ 1 radiation ( $\lambda=1.5406$  Å), operated at 40 kV and 40 mA. Scans were recorded over a 3°–40° 2 $\theta$  range with a step size of 0.01° and step time of 0.1 s. Crystallinity index (Xc) was calculated using OriginPro v9.6.5.1, referencing pure KTP as 100% crystalline (Equation 5).

$$Xc (\%) = \frac{A_{\text{crystalline}}}{A_{\text{crystalline}} + A_{\text{amorphous}}} \times 100\% \quad (5)$$

Fourier-Transform Infrared Spectroscopy (FT-IR) was performed using an AVATAR 330 FT-IR spectrometer (Thermo Fisher Scientific, USA). Samples were blended with dry potassium bromide (KBr) and pressed into pastilles with a Specac® hydraulic press (10-ton force). Spectra were recorded from 4000 to 400 cm<sup>-1</sup> at 4 cm<sup>-1</sup> resolution.

Physical mixtures (PMs) corresponding to each sample, alongside individual excipients were also analysed for their solid-state.

#### 5.5. *In vitro* release study

A modified paddle method (Hanson SR8 Plus) using 100 mL vessels and mini paddles assessed KTP release in 50 mL simulated lung fluid (pH 7.4, 37 °C). Spray-dried samples and a PM (represents KTP and stabilizers) were dispersed in SLF, stirred at 50 rpm for 120 min. Aliquots (5 mL) were collected at set intervals, filtered (0.45  $\mu$ m), and analyzed spectrophotometrically at 258 nm. All tests were run in triplicate. Full method is detailed in our papers [3,146].

## **5.6. Aerosol performance**

### **5.6.1. Andersen cascade impactor**

The aerosolization performance was evaluated using an Andersen Cascade Impactor (ACI, Copley Scientific Ltd., UK) at 60 L/min airflow, generated by a high-capacity pump and verified by a mass flow meter (DFM 2000, Copley Scientific Ltd.). Samples were filled in HPMC capsules and administered via a Breezhaler<sup>®</sup> device (Novartis AG, Switzerland) with a 4-second inhalation. Post-inhalation, KTP was recovered by 50% methanol and quantified spectrophotometrically at 258 nm. Key aerodynamic parameters were calculated including: MMAD (Mass Median Aerodynamic Diameter), FPF (Fine Particle Fraction), GSD (Geometric Standard Deviation), and EF (Emitted Fraction).

### **5.6.2. Aerodynamic particle sizer**

Aerodynamic characterization was further evaluated by Aerodynamic Particle Sizer (APS 3321; TSI Inc., Shoreview, MN, USA). The setup included an induction port, a high-capacity vacuum pump (HCP5), and a critical flow controller (TPK 2000) (all from Copley Scientific Ltd., UK). Inhalation was simulated using a rectangular breathing waveform (4 s, 60 L/min). Three size-based parameters were analyzed: number particle size, surface particle size, and volume particle size. Detailed method was reported in our paper [147].

### **5.6.3. *In silico* modeling**

The latest Stochastic Lung Model (SLM) was used to estimate regional drug deposition in the respiratory tract via Monte Carlo simulation ( $\sim 10^5$  particles) [148]. Inputs included particle properties and inhalation parameters: 69.5 L/min peak flow, 1.7 L volume, and 2.04 s inhalation time. Deposition fractions were expressed as lung deposition (bronchial + acinar), extra-thoracic, and exhaled percentage. Simulations also evaluated breath-hold (BH) effects at 5 and 10 s [147].

## **5.7. Viscosity measurement**

Viscosity was assessed to evaluate potential effects on respiratory mucus properties, relevant to the management of pulmonary diseases. Type II mucin from porcine stomach (Sigma-Aldrich) — the primary glycoprotein in respiratory mucus involved in airway defense and drug transport [149] — was prepared at various concentrations, mixed with spray-dried samples, and stirred for 30 min. Viscosity was measured at 37 °C using a Rotavisk viscometer (IKA-Werke GmbH & Co. KG, Germany) equipped with spindle SP-11 at 120 rpm for 30 s. Sample tested in triplicate, and results reported as mean  $\pm$  SD.



## 5.8. Anti-inflammatory effect

Anti-inflammatory activity was assessed using A549 (alveolar epithelial) and U937 (human pro-monocyte) cells (ATCC, Manassas, VA, USA). A549 cells were cultured in MEM and U937 cells in RPMI, both supplemented with 10% FBS, gentamicin, glucose, L-glutamine, and HEPES. Cells ( $1 \times 10^6$ /well) were seeded into 6-well plates ( $n=3$ /group), treated with Lipopolysaccharides (LPS) and either raw KTP or spray-dried samples, and incubated for 48 h at 37 °C with 5% CO<sub>2</sub>. LPS-only and untreated cells served as positive and negative controls, respectively. mRNA extraction, cDNA synthesis, and qPCR analysis of IL-6 expression were performed as detailed in our published paper [146].

## 5.9. Cell line measurements

### 5.9.1. Cell culture

Two cell lines were used, alveolar (A549) and bronchial (CFBE), and were cultured at 37 °C with 5% CO<sub>2</sub> in their respective supplemented media. For co-culture, endothelial cells were seeded at  $8 \times 10^4$  cells/cm<sup>2</sup> on the underside of 3 µm PET membrane inserts pre-coated with collagen, followed by A549 or CFBE cells on the upper surface. For assays, raw KTP and spray-dried samples were dispersed in culture medium or Ringer-HEPES buffer (pH 7.4) containing 1% FBS, sonicated, and adjusted to 0.5 mg/mL before dilution. Full methods are detailed in our published papers [3,90].

### 5.9.2. Cell viability

The response of alveolar and bronchial epithelial cells to raw KTP and spray-dried samples was assessed via real-time impedance measurements at 10 kHz using the RTCA-SP system (ACEA Biosciences, San Diego, CA). For baseline readings, 50 µL of culture medium was added to collagen-coated E-plate 96 wells with integrated gold electrodes, followed by seeding  $5 \times 10^3$  cells/well. At confluence, cells were exposed to raw KTP or spray-dried samples at 50, 300, and 500 µg/mL for 48 h. Triton X-100 ((TX), 10 mg/mL) served as a cytotoxic control. Full method is detailed in our papers [3,90].

### 5.9.3. Permeability measurement

Permeability was assessed using 12-well plate inserts with Ringer-HEPES buffer in the donor (0.5 mL) and acceptor (1.5 mL) compartments. Raw KTP or spray-dried samples were added to the donor side, and inserts were shaken at 120 rpm for 60 minutes. KTP concentrations were quantified spectrophotometrically at 258 nm, and apparent permeability coefficients ( $P_{app}$ ) were calculated according to Equation 6.

$$P_{app} \text{ (cm/s)} = \frac{\Delta[C]_A \times V_A}{A \times [C]_D \times \Delta t} \quad (6)$$

Transepithelial electrical resistance (TEER), marker molecule permeability assays, and immunocytochemistry were conducted to assess the integrity of intercellular junctions and confirm that barrier function remained intact post-permeability testing. Detailed methods are provided in our published papers [3,90].

## **5.10. *In vivo* pharmacokinetics**

### **5.10.1. Animal model**

Healthy 7-week-old male Sprague-Dawley rats ( $226 \pm 18$  g) were used. Animals were housed under standard laboratory conditions with food and water ad libitum. All procedures were approved by the *Hungarian Ethical Committee for Animal Research (IV./2706/2024)* and complied with FELASA recommendations.

### **5.10.2. Administration and sampling protocol**

Rats ( $n = 48$ ) were divided into oral ( $n = 4/\text{group}$ ) and inhalation ( $n = 20/\text{group}$ ) groups. Oral groups received either raw KTP unprocessed suspension (KTP-RAW) or nanosuspension (KTP-NS) by gavage. Inhalation groups received dry powders (K1 or K1ML) via intratracheal insufflation under sevoflurane anesthesia using a Penn-Century DP-4 insufflator. All animals received 3 mg/kg KTP. Blood and lung samples were collected at 15, 30, 60, 120, and 360 min post-dose.

### **5.10.3. UHPLC-MS/HRMS analysis for quantification**

KTP concentrations in lung and plasma were quantified by UHPLC-MS/HRMS (Waters Acuity I-Class UPLC–Thermo Orbitrap Exploris 240) in PRM mode. Full method is detailed in our paper [3].

### **5.10.4. Pharmacokinetic analysis**

Non-compartmental analysis was performed using Phoenix WinNonlin 8.5.2.4, applying the sparse sampling method. Key pharmacokinetic (PK) parameters ( $C_{\max}$ ,  $t_{\max}$ ,  $AUC_{0-6}$ ,  $AUC_{0-\infty}$ ,  $k_e$ ,  $t_{1/2}$ ) were determined from the concentration-time profiles. Relative bioavailability (F) and tissue availability ( $f_t$ ) were also determined as follows:

$$F (\%) = \frac{AUC_{0-\infty} (\text{intratracheal})}{AUC_{0-\infty} (\text{Oral})} \times \frac{\text{Dose}_{(\text{oral})}}{\text{Dose}_{(\text{intratracheal})}} \quad (7)$$

$$f_t = \frac{AUC_{0-6} (\text{lung})}{AUC_{0-6} (\text{plasma})} \quad (8)$$

## **5.11. Safety and tolerance in sensitized rats**

### **5.11.1. Animal model**

Six-week-old male (n=24) and female (n=24) Wistar rats were housed under standard conditions with ad libitum access to food and water. All procedures complied with EU Directive 2010/63/EU and national regulations and were approved by the *Hungarian authorities* (No. XXXII./1576/2024). The study followed ARRIVE guidelines.

### **5.11.2. Allergen-sensitized animal model**

Allergen sensitization was employed based on a previously published method [150]. In brief, rats received an intraperitoneal injection of ovalbumin (OVA, 1 mg) with aluminum hydroxide (50 mg) on day 0, followed by daily inhalation of aerosolized OVA (25 mg/mL, 15 min) from days 14 to 21.

### **5.11.3. Treatment protocol and animal grouping**

Rats were divided into five groups: three OVA-sensitized and treated with K1, K1ML, or placebo (spray-dried stabilizer), one positive control (OVA-sensitized, untreated), and one negative control (non-sensitized, untreated). Intratracheal treatments were administered every third day (days 0–14) and every other day (days 14–21).

### **5.11.4. Measurement of respiratory mechanics and lung responsiveness**

End-expiratory lung volume (EELV) was measured using whole-body plethysmography, with box and tracheal pressures recorded during 15 s end-expiratory occlusion at positive end-expiratory pressure (PEEP) levels of 0, 3, and 6 cmH<sub>2</sub>O, based on the Boyle-Mariotte law [151]. Respiratory impedance was assessed using the forced oscillation technique, quantifying airway resistance (Raw), tissue damping (G), and elastance (H) [152–154]. Airway hyperresponsiveness was evaluated via bronchoprovocation with methacholine (MCh) and ovalbumin (OVA), as described in our previous study [3].

## **5.12. Stability study**

Stability testing was conducted in a Binder KBF 240 chamber at  $40 \pm 2$  °C/ $75 \pm 5\%$  relative humidity (RH) per ICH guidelines. Formulations placed in HPMC capsules were assessed over 3 months, with samples taken at 0, 1, and 3 months, as detailed in [90,147].

## **5.13. Statistical Analysis**

GraphPad Prism 8.0.1 (CA, USA) was employed using a two-way ANOVA test, with significant values at  $p \leq 0.05$ . All data was expressed as mean  $\pm$  SD.

## 6. Results and discussion

### 6.1. Nanosuspension characterization

#### 6.1.1. Stabilizer selection

Stabilizer effects were assessed via dynamic light scattering (DLS) (Table 5). HPMC led to the largest PSs (~450–2570 nm) and highest PDIs, indicating poor stabilization. Compared to PVA, poloxamer 188 showed good zeta potential but larger particles. PVA provided the smallest PS and lowest PDI, especially at 1% (w/v), therefore; it was selected for this study. Combining PVA and SDS optimized stability, leveraging PVA's wettability enhancement and SDS's electrostatic repulsion [155].

**Table 5:** A preliminary study of KTP-NS using different stabilizers and concentrations. All stabilizers were combined with 0.1% SDS [146].

Stabilizer	Conc (w/v)	PDI	PS (nm)	ZP (mV)
HPMC	0.25	$0.454 \pm 0.11$	$2570.0 \pm 42.6$	$-5.50 \pm 1.34$
	0.5	$0.739 \pm 0.02$	$764.3 \pm 2.79$	$-8.10 \pm 4.24$
	1	$0.319 \pm 0.03$	$453.0 \pm 4.15$	$-6.20 \pm 2.87$
PVA	1	$0.082 \pm 0.01$	$238.3 \pm 1.37$	$-9.97 \pm 5.31$
	2.5	$0.227 \pm 0.04$	$316.4 \pm 1.91$	$-11.60 \pm 2.75$
	5	$0.205 \pm 0.02$	$431.4 \pm 3.16$	$-4.22 \pm 0.23$
Poloxamer 188	0.2	$0.428 \pm 0.14$	$583.2 \pm 12.6$	$-13.10 \pm 1.18$
	0.5	$0.213 \pm 0.05$	$3112.1 \pm 36.4$	$-25.20 \pm 3.17$
	1	$0.321 \pm 0.16$	$415.7 \pm 13.1$	$-17.50 \pm 2.94$

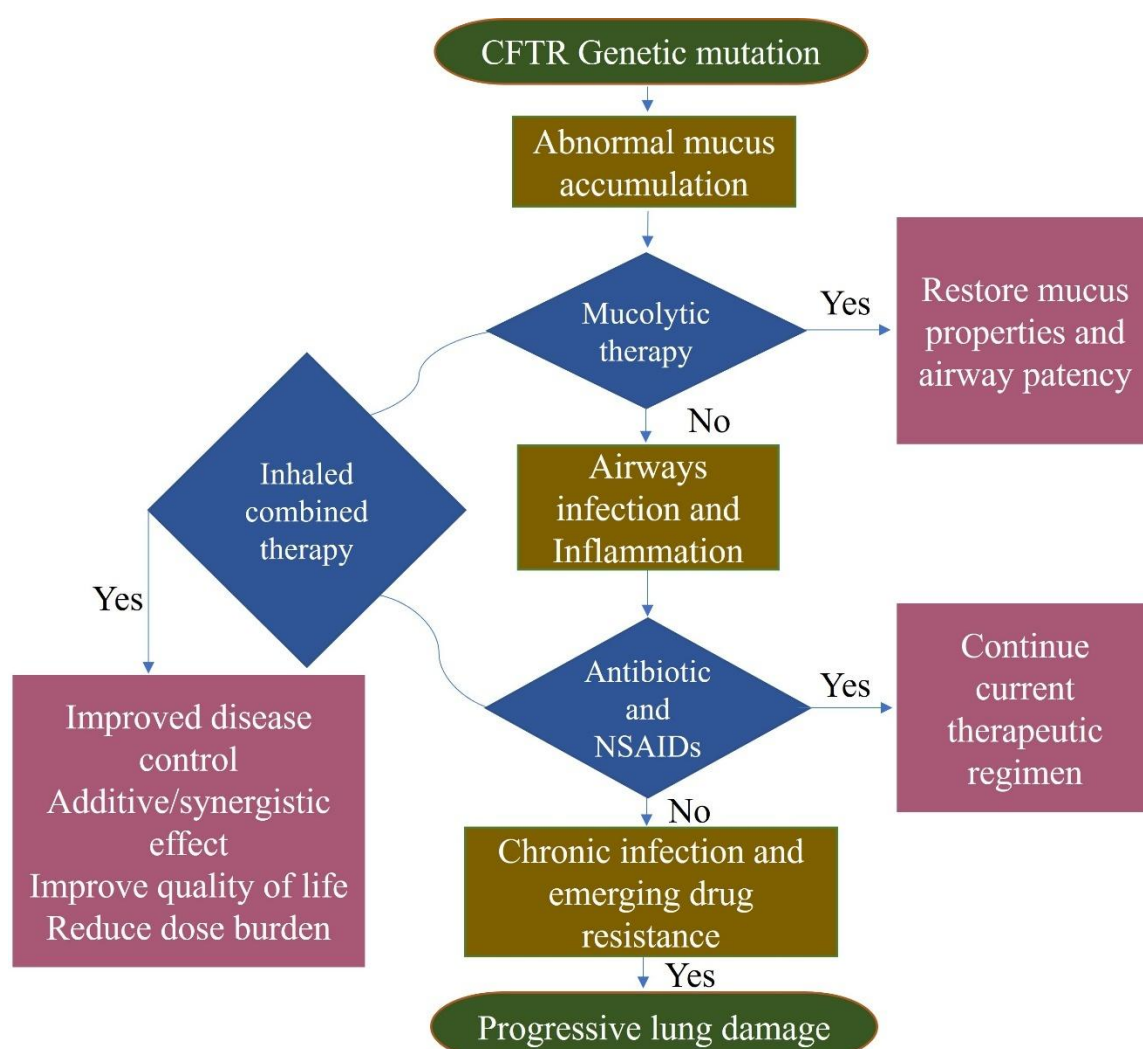
#### 6.1.2. Nanosuspension stability

Although DLS is commonly used for particle sizing, it tends to overrepresent larger particles [156]. To overcome this, nanoparticle tracking analysis (NTA) was used, providing more accurate size distribution by tracking individual particle motion [157]. NTA results were as follows: D[0.1]:  $119.60 \pm 3.30$  nm, D[0.5]:  $168.70 \pm 4.00$  nm, D[0.9]:  $230.80 \pm 11.70$  nm, with a span of  $0.659 \pm 0.024$ . Short-term stability over 45 days at +4 °C showed a slight D[0.5] increase (+29 nm,  $P \leq 0.001$ ) but stable span, indicating minimal aggregation [94]. Thus, the nanosuspension was deemed suitable for dry powder production [146,147].

## 6.2. Powder combinations for local pulmonary delivery assessments

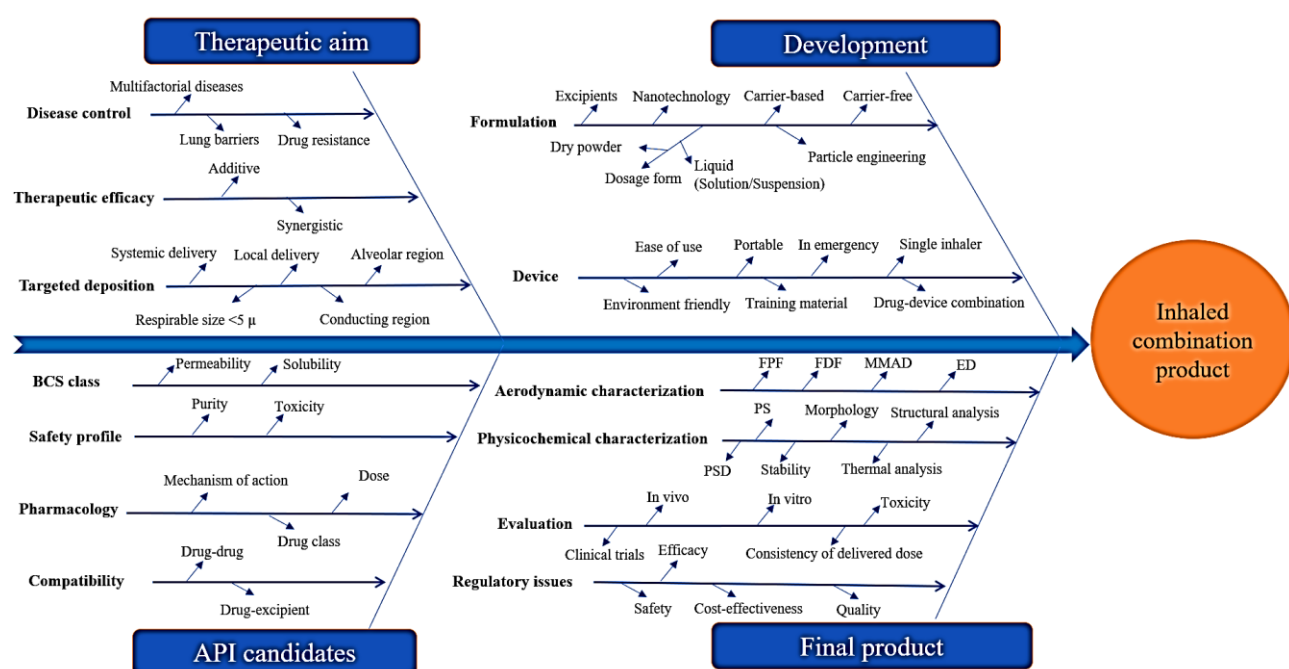
### 6.2.1. QbD approach for combined DPI development

A flowchart was employed to highlight challenges in CF treatment and the rationale for developing multi-drug inhaled therapies (Figure 7). Current single-agent treatments, such as TOBI® (tobramycin) for infection and Bronchitol® (mannitol) for mucus clearance, address limited aspects of CF pathology. NSAIDs (e.g., KTP) offer anti-inflammatory benefits and may slow disease progression [158]. While some combinations of antibiotics and mucolytics have been explored [159–161], NSAID-mucolytic combinations remain unexplored.



**Figure 7:** Flowchart illustrating the pathophysiological cascade of CF and current treatment strategies, highlighting the potential role of novel inhaled combination therapies (e.g., mucolytics, antibiotics, and NSAIDs) in enhancing airway clearance, controlling infection, improving clinical outcomes and reducing dose burden. CFTR: Cystic fibrosis transmembrane conductance regulator. Adapted from [1], with slight modifications.

Furthermore, an Ishikawa (fishbone) diagram—a tool used to identify key factors influencing product quality—was employed to support the development process (Figure 8). This diagram mapped critical elements affecting inhaled combination therapies, based on extensive literature and prior studies of pulmonary drug combinations [162–164]. Its design incorporated established scientific knowledge, considering therapeutic goals, API selection, development stages, and final product profiles.



**Figure 8:** Fishbone diagram defining factors affecting the development of combined products for inhalation [1].

The Ishikawa diagram supported the definition of the QTPP for combination therapies, such as KTP–mannitol (Table 6). Based on the QTPP, key CQAs were identified, including formulation technique, deposition site, particle size, aerosol performance, drug compatibility, and stability. Accordingly, spray-dried KTP–mannitol combinations were prepared and characterized to align with the defined QTPP.

The proper application of a QbD approach from the early development phase offers clear advantages in inhaled combination therapy development, including: (i) enhanced production capability, (ii) more efficient use of development time and resources, (iii) reduced product variability and defects, and (iv) a smoother FDA submission and approval process.

**Table 6:** QTPP of KTP-mannitol combination for inhalation as dry powders [1].

QTPP	Target	Justification	Reference
Route of administration	Pulmonary	Enables direct lung targeting with rapid onset and reduced systemic side effects.	[165]
Dosage form	Inhalable powder	Offers high stability and allows higher drug loading.	[166]
Delivery target	Local/systemic	Inhalation enables both local targeting (e.g., for inflammation) and systemic absorption via alveolar transfer.	[29,167]
Formulation design	Carrier or carrier-free	Allows flexibility using traditional carriers or advanced particle engineering.	[168]
Device	Capsule or non-capsule based	Both formats are feasible for pulmonary delivery.	[169,170]
Deposition site	Conducting/alveolar zone	Site selection depends on drug action and disease localization.	[171]
Therapeutic effect	Synergistic/additive	Combining drugs may enhance efficacy through synergy or additive effects.	[172,173]
Particle size	<5 $\mu\text{m}$	Required for deep lung deposition.	[33]
Morphology	Spherical/corrugated	Corrugated particles improve lung deposition.	[17,18]
Compatibility	Drug-drug, drug-excipient, drug-device	Ensures stability and safety across formulation components.	[174]
Dissolution	Rapid	Minimizes mucociliary clearance and enhances bioavailability.	[175]
Stability	No agglomeration or size change	Maintains consistent delivery performance.	[176]
Density	$\rho_b < 0.3 \text{ g/mL}$ , $\rho_t < 0.4 \text{ g/mL}$	Improves lung deposition.	[15,16]

$\rho_b$ : bulk density,  $\rho_t$ : tapped density.

### 6.2.2. Particle size and density

Table 7 shows that our particle engineering strategies were effective, as spray drying did not significantly increase KTP particle size, indicating good redispersibility and minimal aggregation. ZP values outperformed previous PVA-stabilized formulations for pulmonary delivery [177]. The highest PDI was observed in the mannitol-free sample (F0), suggesting that mannitol improved dispersity.

Furthermore, our formulations had low tapped densities (0.18–0.22 g/cm<sup>3</sup>) (Table 7), which are optimal for pulmonary delivery [178–180].

**Table 7:** Characteristics of spray-dried samples (F0-F2), including yield (%), drug loading (DL), particle size (DLS), density measurements, aerodynamic characteristics (ACI), and viscosity measurement [146].

Analysis	Parameter	F0	F0.5	F1	F2
Yield (%)	Yield (%)	52.95 ± 3.54	54.86 ± 6.12	57.29 ± 1.98	58.68 ± 9.73
Drug content	DL (%)	57.47 ± 1.44	84.12 ± 2.91	84.87 ± 7.13	84.72 ± 3.65
	PS (nm)	204.9 ± 3.07	222.5 ± 4.11	240.7 ± 6.32	251.4 ± 2.84
Particle size (DLS)	PDI	0.336 ± 0.003	0.127 ± 0.021	0.064 ± 0.008	0.156 ± 0.039
	ZP (mV)	-8.88 ± 0.27	-12.3 ± 0.43	-7.44 ± 0.18	-11.9 ± 0.33
Density	ρ <sub>t</sub> (g/mL)	0.180 ± 0.002	0.192 ± 0.011	0.201 ± 0.009	0.228 ± 0.052
	ρ <sub>b</sub> (g/mL)	0.124 ± 0.012	0.123 ± 0.003	0.120 ± 0.031	0.139 ± 0.024
Aerosol performance	MMAD (μm)	2.40 ± 0.17	2.80 ± 0.06	4.51 ± 0.41	4.90 ± 0.16
	FPF <5 (%)	56.16 ± 2.51	71.02 ± 1.19	64.32 ± 1.34	32.21 ± 3.67
	EF (%)	97.06 ± 3.22	96.60 ± 1.65	94.82 ± 2.79	95.70 ± 2.89
Rheology	Viscosity (Pa.s)	0.033 ± 1.98	0.031 ± 2.12	0.025 ± 1.37	0.030 ± 4.19

DLS: dynamic light scattering, PS: particle size, PDI: polydispersity index, ZP: zeta potential, ρ<sub>t</sub>: tapped density, ρ<sub>b</sub>: bulk density, MMAD: mass median aerodynamic size, FPF: fine particle fraction and EF: emitted fraction.

### 6.2.3. Drug loading and yield

Spray-dried samples showed yields of 53–59% (Table 7). Increasing mannitol concentration positively affected the yield by reducing cohesive forces between particles, as low spray-drying yields are often associated with particle cohesion [181]. These yields surpass those reported in similar spray-drying studies [182–184].

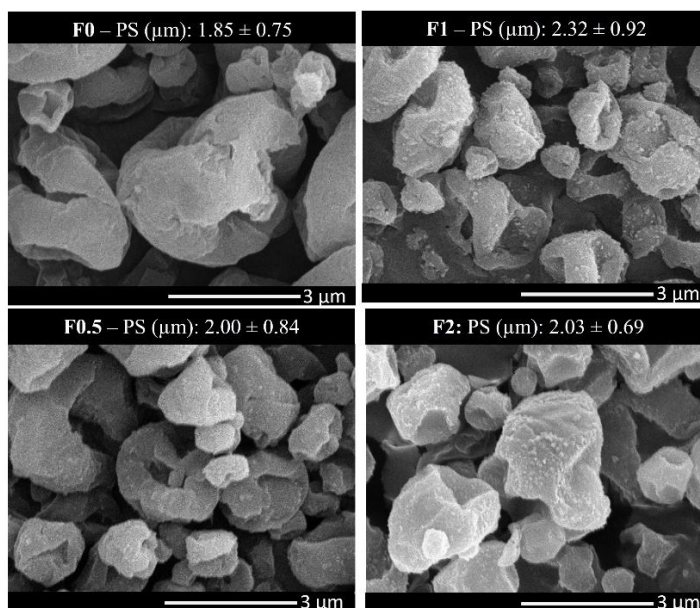
DL ranged from 57% to 85%, with no significant impact from mannitol concentration (F0.5–F2).

### 6.2.4. Morphology

Figure 9 shows SEM images of spray-dried samples (F0–F2), revealing nearly spherical particles with corrugated rough surfaces. Spherical shapes offer stability for pulmonary delivery [185], while rough surfaces improve lung deposition by increasing contact angles and enhancing cell attachment [17,18].

Particle sizes, measured by ImageJ, revealed that mannitol-free sample (F0) had the smallest size, but increasing mannitol content (F0.5–F2) had no significant effect on particle size.





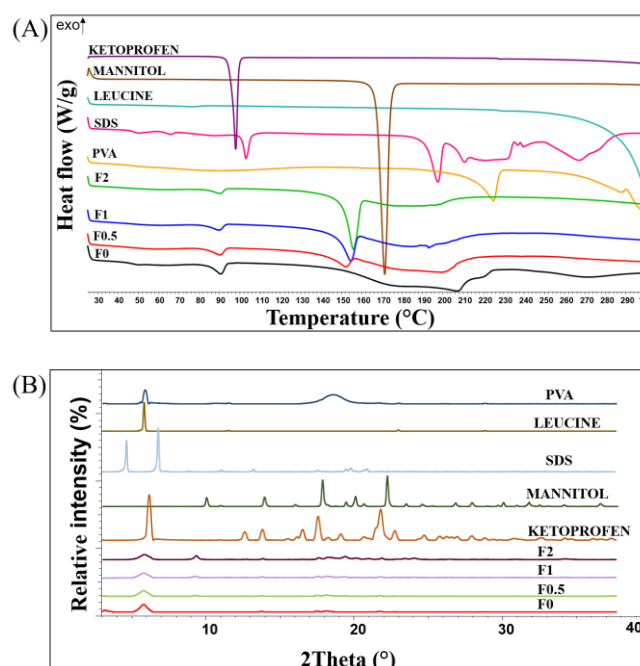
**Figure 9:** Morphology images using scanning electron microscope (SEM) of spray-dried samples with the diameter of final product measured by Image-J software. PS: particle size [146].

#### 6.2.5. Solid-state analysis

DSC thermograms showed a sharp endothermic peak for raw KTP at 97.83 °C, indicating its crystalline nature (Figure 10A). In formulations, KTP's melting point shifted to ~90 °C, reflecting nanosizing effects and increased amorphous content. Mannitol exhibited a distinct peak at 171.33 °C, confirming its crystalline form. In samples F0.5, F1, and F2, mannitol retained crystallinity, with melting peaks observed at 152.17 °C, 154.50 °C, and 155.83 °C, respectively. The presence of crystalline mannitol is advantageous, since it contributed to improved *in vitro* dissolution and diffusion profiles in pulmonary delivery [186,187]. Increasing mannitol also reduced KTP peak intensity and broadened its thermal signal, suggesting reduced crystallinity—consistent with previous findings [188]. Overall, the results support the compatibility of KTP–mannitol combinations.

Furthermore, DPI formulations require low moisture for effective aerosolization. TGA analysis showed residual water content of 1.08–2.12% in our samples (results not shown), lower than values reported in previous studies [189,190].

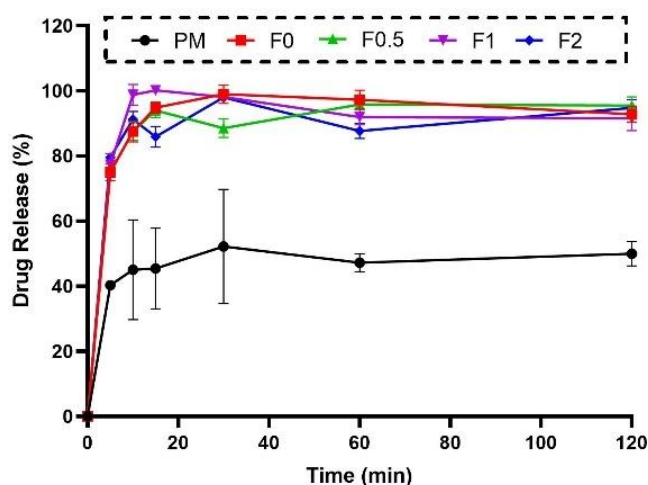
XRPD was used to assess crystallinity (Figure 10B). Raw KTP showed sharp peaks at 5.79°, 12.9°, 14.20°, 17.12°, 18.30°, and 21.83° 2θ, confirming its crystalline nature. All formulations (F0–F2) showed lower-intensity KTP peaks, indicating partial amorphization. Crystallinity percentages of KTP were 60.30% (F0), 44.54% (F0.5), 41.96% (F1), and 28.36% (F2), confirming that increasing mannitol content reduced KTP crystallinity, consistent with DSC results.



**Figure 10:** Thermal and structural analysis of raw materials and spray-dried samples analysed by (A) DSC (B) XRPD [146].

#### 6.2.6. *In vitro* release study

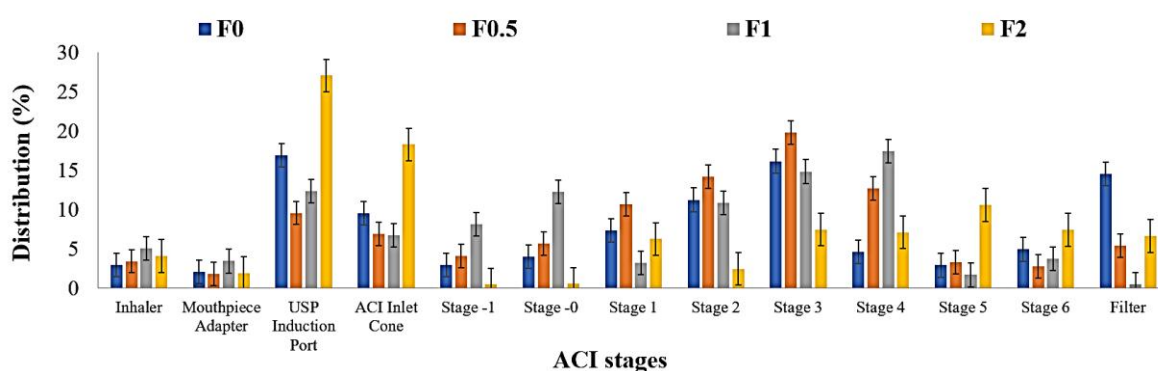
*In vitro* dissolution in simulated lung fluid (pH 7.4) showed ~80% KTP release from spray-dried samples within 5 min, compared to 40% from the PM (Figure 11). This improvement is mainly due to nanosized KTP, while mannitol's wettability also enhanced disintegration and drug release [191]. F1 showed the fastest release (~100% in 10 min), whereas both lower (F0.5) and higher (F2) mannitol concentrations reduced this effect, highlighting the need for precise optimization of mannitol concentration, which is crucial for maximizing the therapeutic potential and performance of this combination.



**Figure 11:** *In vitro* release study of physical mixture (PM) and samples F0, F0.5, F1 and F2 in simulated lung media (SLM). Results are expressed as mean  $\pm$  SD (n=3) [146].

### 6.2.7. *In vitro* aerosol performance

Aerodynamic evaluation using ACI confirmed the suitability of all samples for pulmonary delivery (Table 7). Increasing mannitol content increased the MMAD, while maintaining high FPF of 71.02% for F0.5 and 64.32% for F1—values exceeding some marketed products [192]. All samples met the EF criteria (95–97%) [182]. Deposition data (Figure 12) showed that F0.5 and F1 favored deeper lung regions (stages 3 and 4), whereas the high mannitol formulation (F2) deposited mainly in the USP induction port and ACI inlet cone, indicating predominant extra-thoracic deposition. The mannitol-free sample (F0) also showed high deposition in the USP induction port and filter. These findings highlight that both the inclusion and concentration of mannitol significantly influence deposition behavior and optimal lung targeting.



**Figure 12:** *In vitro* aerodynamic distribution of spray-dried samples at a flow rate of 60 L/min. Results are expressed as mean  $\pm$  SD (n=3) [146].

### 6.2.8. Viscosity measurement

Viscosity abnormalities contribute to many respiratory diseases [193], so we assessed our formulations' effect on mucin viscosity (Table 7). Mucin solutions were prepared at 2%, 5%, and 10% based on previous studies [194–196]. The 2% solution was too dilute to measure, while 5% and 10% had viscosities of 0.019 and 0.035 Pa·s, respectively—within the range reported in CF and COPD patients [197,198]. Consequently, 10% mucin ( $0.035 \pm 5.44$  Pa·s) was selected for testing. Among all formulations, only F1 resulted in a notable viscosity reduction. This effect may be attributed to the optimal mannitol concentration in F1, which enhances mucin hydration.

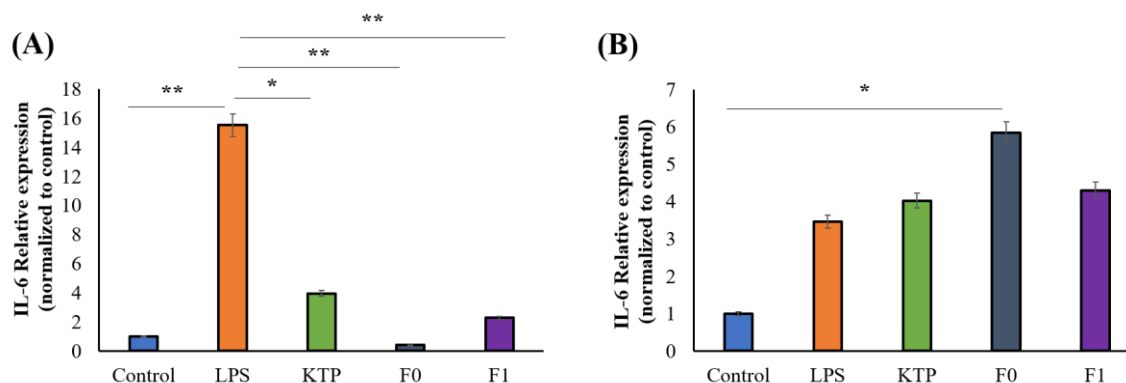
These findings suggest a concentration-dependent mechanism, where mannitol at an appropriate level facilitates mucus thinning—a key therapeutic goal in conditions like CF. Thus, F1 represents a promising formulation, meriting deeper exploration for its mucolytic

potential in combination with KTP. Accordingly, F1 was selected for continued evaluation, with F0 serving as a reference.

### 6.2.9. Anti-inflammatory effect

We assessed IL-6 expression in LPS-stimulated A549 and U937 cells to evaluate the anti-inflammatory effects of our formulations. LPS, a proinflammatory trigger [199], induced IL-6—a key cytokine in pulmonary inflammation [200]. U937 cells, known for strong IL-6 response due to high COX-2 expression [201,202], showed significantly higher IL-6 levels than A549 cells (15.5 vs. 3.4,  $p < 0.01$ ).

All samples significantly reduced IL-6 in U937 cells (Figure 13A), and mannitol did not impair this effect, confirming a promising KTP-mannitol combination. In contrast, no reduction was observed in A549 cells (Figure 13B), aligning with previous findings [203].



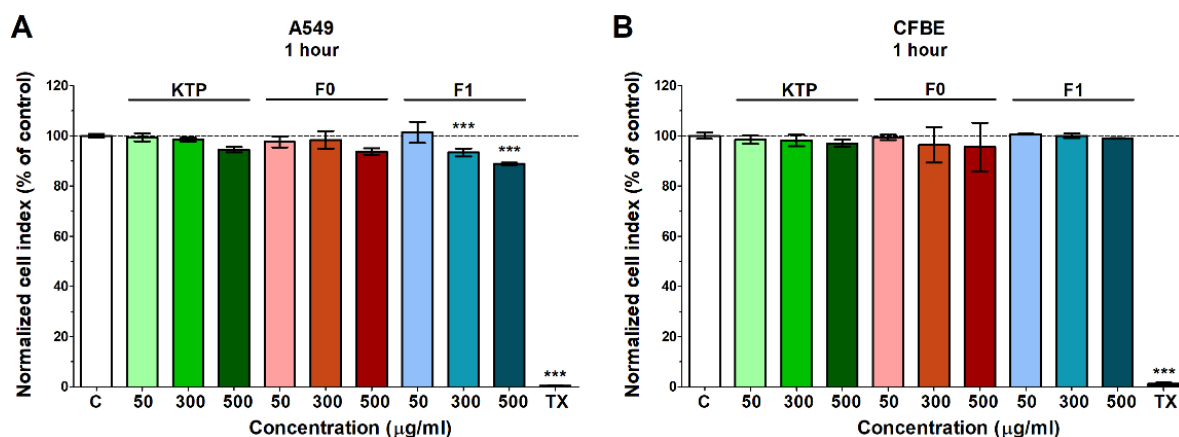
**Figure 13:** Relative expression of IL-6 on two cell lines; (A) U937 and (B) A549. Control is the untreated cell line, LPS is the treated cell line, raw KTP, F0, F0.5, F1 and F2 are treated with LPS. Results are expressed as mean  $\pm$  SD (n=3). Level of significance: \* $p < 0.05$ , \*\* $p < 0.01$  [146].

### 6.2.10. Cell line evaluation

#### 6.2.10.1. Cell viability

To ensure pulmonary safety, cell viability was assessed using real-time impedance measurements. In A549 cells, only sample F1 slightly reduced impedance at high concentrations (Figure 14A), while both F0 and F1 had no effect on CFBE cells (Figure 14B), indicating good biocompatibility across formulations and nearly all concentrations which is consistent with previously published findings [204].

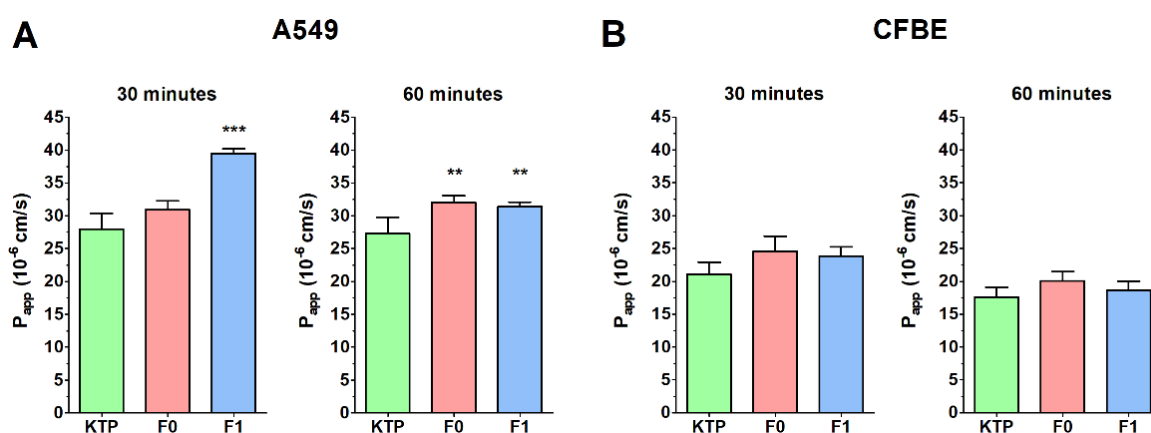
Accordingly, KTP concentration of 50  $\mu\text{g/mL}$  was selected for permeability studies.



**Figure 14:** Cell viability of (A) A549 alveolar and (B) CFBE bronchial epithelial cells after 1-h treatments with raw KTP and spray-dried samples F0, F1 measured by impedance. Values are presented as means  $\pm$  SD,  $n = 5-6$ . \*\*\*  $p < 0.001$  compared to the C (control group) [90].

#### 6.2.10.2. Cell permeability

Permeability was evaluated in A549 and CFBE cells at 30 and 60 min (Figure 15). KTP showed lower permeability across bronchial (CFBE) than alveolar (A549) barriers, likely due to cell line nature differences [205]. Both F0 and F1 improved KTP permeability in the alveolar model, with F1 enhancing it 1.4-fold at 30 min—possibly due to mannitol's effect on particle surface or epithelial interaction, facilitating better permeation [206].



**Figure 15:** Permeability of raw KTP, F0, and F1 (50  $\mu$ g/mL KTP concentration in the donor compartment) across the co-culture model of (A) alveolar epithelial cells and (B) bronchial epithelial cells after 30- and 60-min assay time. Values are presented as means  $\pm$  SD,  $n = 4$ . \*\*  $p < 0.01$ , \*\*\*  $p < 0.001$  [90].

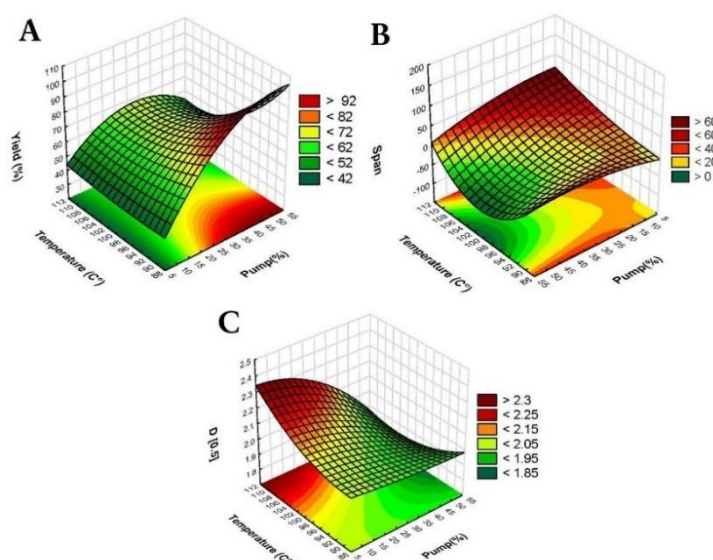
Barrier integrity was confirmed by TEER measurements and permeability assays using sodium fluorescein and Evans Blue-labeled albumin, with no observed morphological changes in cell–cell junctions (results not shown) [90].

Hence, this study confirms that combining mannitol and KTP in 1:1 mass ratio as a single DPI is promising for targeting pulmonary inflammations and simultaneously improving cell diffusion and mucus clearance.

### 6.3. Nanocrystal-based dry powder for systemic delivery evaluations

#### 6.3.1. Factorial design

A BBD was used to optimize the nano spray-drying process by varying pump rate, inlet temperature, and feed concentration across 15 runs. Key responses—median particle size ( $D[0.5]$ ), span, and yield—were analyzed using Pareto charts and 3D surface plots (Figure 16) [147]. Since  $D[0.5]$  and span are critical for pulmonary delivery performance, they were prioritized. Increasing pump rate reduced  $D[0.5]$  and span due to finer atomization [207], while higher feed concentrations increased particle size [208]. Low feed concentrations led to broader spans, indicating poor droplet stability. Higher inlet temperatures (100–110 °C) caused particle aggregation and size growth [106], whereas 90 °C produced smaller, more uniform particles. Based on these findings, optimal conditions—50% pump rate, 90 °C inlet temperature, and 5% feed concentration—were selected to produce mannitol- and/or leucine-based NCA formulations.



**Figure 16:** 3D surface plots of the effect of temperature (°C) and pump (%) on (A): yield (%), (B): span and (C):  $D[0.5]$  [147].

#### 6.3.2. Particle size and density

Following process optimization, laser diffraction (Table 8) showed K1 had a  $D[0.5]$  of  $1.708 \pm 0.074 \mu\text{m}$ . Adding leucine (K1L) slightly increased size ( $1.893 \pm 0.017 \mu\text{m}$ ) due to droplet stabilization during drying [209], while mannitol (K1M) led to larger particles ( $3.337 \pm 0.190 \mu\text{m}$ ) from its high crystallinity [210]. The combination (K1ML) yielded an intermediate size ( $2.046 \pm 0.064 \mu\text{m}$ ). Compared to K1, Leucine and/or mannitol also reduced span values, suggesting more uniform particles and potentially better lung deposition.

**Table 8:** Characteristics of NCAs, including particle size (laser diffraction), drug loading (DL), encapsulation efficiency (EE), density measurements, and aerodynamic performance assessed by ACI [3,147].

Analysis	Parameter	K1	K1M	K1L	K1ML
Particle size	D[0.5]	1.708 ± 0.074	3.337 ± 0.190	1.893 ± 0.017	2.046 ± 0.064
		1.412 ± 0.071	1.390 ± 0.118	1.372 ± 0.070	1.387 ± 0.008
	span	1.412 ± 0.071	1.390 ± 0.118	1.372 ± 0.070	1.387 ± 0.008
		3.945 ± 0.247	2.09 ± 0.156	3.56 ± 0.001	3.290 ± 0.057
Drug content	DL (%)	41.38 ± 3.54	26.24 ± 6.12	30.08 ± 1.98	20.55 ± 9.73
		84.83 ± 1.44	80.04 ± 2.91	91.73 ± 7.13	83.24 ± 3.65
	EE (%)	84.83 ± 1.44	80.04 ± 2.91	91.73 ± 7.13	83.24 ± 3.65
		95.56 ± 3.22	97.05 ± 1.65	96.23 ± 2.79	92.67 ± 2.89
Density	ρ <sub>t</sub> (g/mL)	0.225	0.097	0.225	0.34
	ρ <sub>b</sub> (g/mL)	0.144	0.064	0.144	0.189
Aerosol performance	MMAD (μm)	2.016 ± 0.17	4.50 ± 0.06	0.927 ± 0.41	2.661 ± 0.16
	FPF <5 (%)	84.07 ± 1.51	25.05 ± 1.19	89.96 ± 1.34	69.30 ± 3.67
		74.73 ± 1.23	20.65 ± 1.91	79.88 ± 1.42	52.75 ± 1.57
	FPF <3 (%)	74.73 ± 1.23	20.65 ± 1.91	79.88 ± 1.42	52.75 ± 1.57
		95.56 ± 3.22	97.05 ± 1.65	96.23 ± 2.79	92.67 ± 2.89
	EF (%)	95.56 ± 3.22	97.05 ± 1.65	96.23 ± 2.79	92.67 ± 2.89
Aerosol performance	GSD	1.658	2.95	N/A	1.986

ρ<sub>t</sub>: tapped density, ρ<sub>b</sub>: bulk density, SSA: specific surface area. Results are expressed as mean ± SD (n=3 independent measurements).

The samples showed different densities (Table 8), influencing aerosol performance. K1L's density was similar to K1, indicating minimal impact from leucine. K1M had a lower density, forming a lighter powder, while K1ML was denser, potentially favoring deep lung deposition. All formulations met inhalation criteria, with bulk densities below 0.3 g/mL and tapped densities under 0.4 g/mL [15,16], confirming suitability for pulmonary delivery.

### 6.3.3. Drug loading and encapsulation efficiency

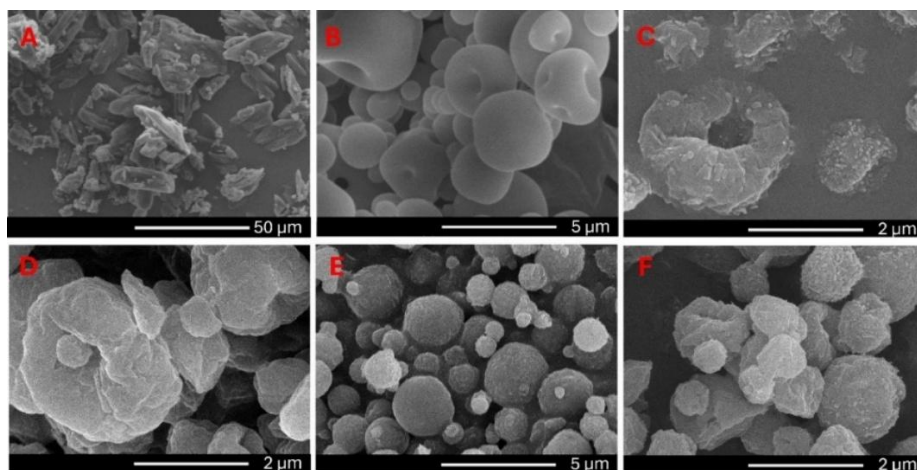
High DL and EE values in K1 demonstrated efficient drug incorporation in NCA without excipients. K1M and K1L saw reduced DL but maintained high EE, with leucine improving encapsulation due to its surfactant properties (Table 8).

### 6.3.4. Morphology

Morphology results are shown in Figure 17. Compared to the nano spray-dried stabilizer (Figure 17B), SEM images confirmed that KTP in K1 (Figure 17C) existed as nanocrystals agglomerated on the stabilizer surface, indicating high crystallinity without surface modification. Incorporating leucine (K1L, Figure 17D) produced a corrugated shell structure due to leucine's surface activity. In K1M (Figure 17E), mannitol promoted the



formation of larger, rough-surfaced, spherical particles. K1ML (Figure 17F) combined features of both a crystalline surface and shell-like structure, reflecting the effects of both excipients. Despite morphological differences, all samples exhibited rough surfaces, beneficial for lung deposition by enhancing contact angles and cell adhesion [17,18].



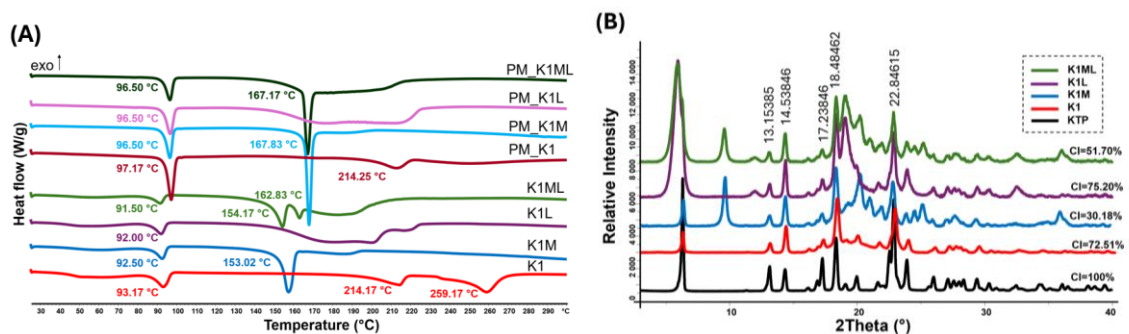
**Figure 17:** SEM images of (A) raw KTP, (B) nano spray dried-stabilizer system, (C) K1, (D) K1L, (E) K1M and (F) K1ML [3].

### 6.3.5. Solid-state analysis

Raw KTP showed a melting peak at 97.83 °C, which shifted to 91.5–93.17 °C in NCA samples due to particle size reduction. Mannitol-containing samples (K1M, K1ML) exhibited additional peaks at ~153 °C, corresponding to mannitol melting. Interestingly, K1ML showed an extra peak, suggesting distinct mannitol polymorphs, likely from spray drying at different ratios [211]. Further studies are needed to confirm this (Figure 18A). Also, TGA showed low water content across samples (0.16–1.8%; data not shown). K1 (without mannitol/leucine) had the highest (1.8%), while mannitol and leucine reduced moisture levels. Water content was within the acceptable range for DPI formulations [212,213].

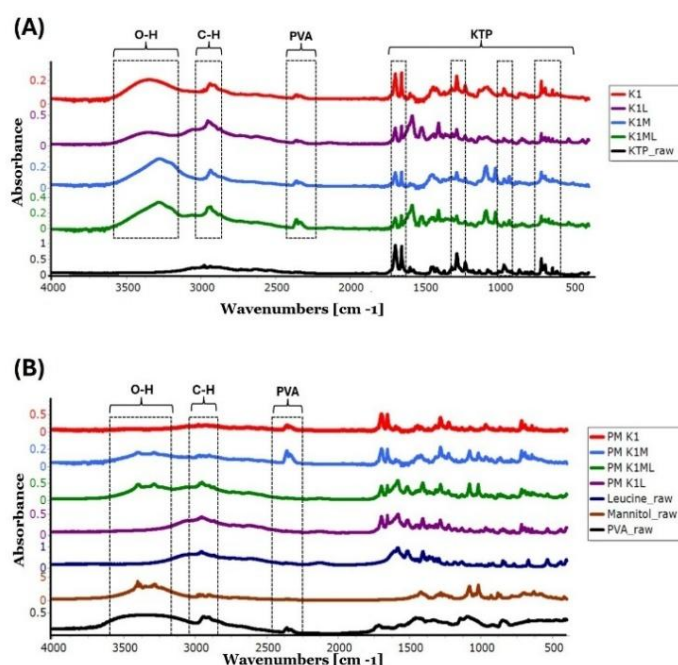
XRPD analysis (Figure 18B) showed that all formulations (K1, K1M, K1L, K1ML) retained the characteristic crystalline peaks of raw KTP. Crystallinity, calculated relative to 100% for raw KTP, was 72.51% (K1), 30.18% (K1M), 75.20% (K1L), and 51.70% (K1ML). Mannitol reduced crystallinity, consistent with its known impact during co-spray drying [188], while leucine led to the highest crystallinity due to its surface coverage and solid-phase properties.





**Figure 18:** Results of solid-state analysis (A) Thermal analysis by DSC of NCAs samples and PMs, (B) Structural analysis by XRPD of raw KTP and NCAs samples with values of crystallinity index (%), PM: physical mixture, CI: crystallinity index [3].

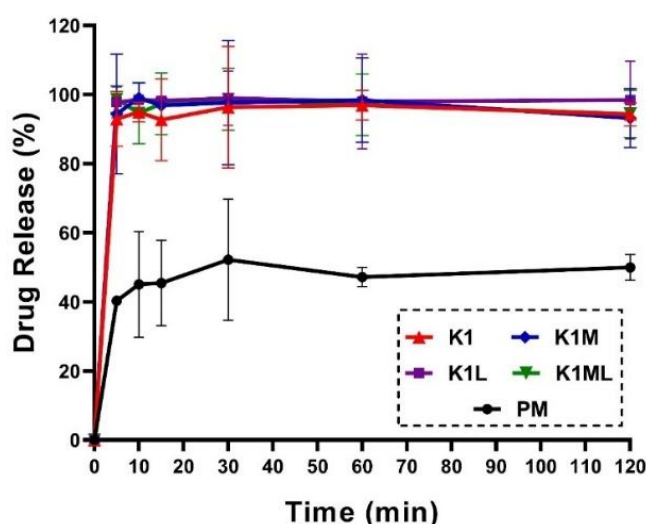
FTIR analysis (Figure 19A) showed that all samples retained KTP's characteristic peaks (719, 971, 1286, 1658, and 1701  $\text{cm}^{-1}$ ), indicating no structural changes. Variations in peak intensity suggested differences in crystallinity. The C-H stretching peak ( $\sim 2932 \text{ cm}^{-1}$ ) was preserved, while leucine reduced hydrogen bonding in PVA, seen by a narrower, weaker  $\sim 3367 \text{ cm}^{-1}$  peak, likely due to its surface activity and interference with PVA's hydroxyl groups [214]. Mannitol caused a shift in the O-H stretching from 3367 to 3285  $\text{cm}^{-1}$ , indicating new hydrogen bonding with PVA. FTIR spectra of PMs (Figure 19B) showed partial loss of PVA peaks, suggesting that nano spray drying prevents phase separation and aggregation.



**Figure 19:** (A) FTIR results of raw KTP and NCAs samples with characteristic peak of KTP in fingerprint region, and (B) FTIR results of corresponding unprocessed physical mixture and raw materials. O-H: stands for O-H stretching vibration, and C-H: stands for C-H stretching vibration [3].

### 6.3.6. *In vitro* release study

For systemic effect in pulmonary delivery, it is essential for the drug to dissolve and release quickly upon deposition to minimize the risk of clearance mechanisms [175,177]. The data revealed that 93%, 94%, 98%, and 99% of the drug was released from K1, K1M, K1L, and K1ML, respectively, within the first 5 minutes, compared to 40% released from the PM (Figure 20). Although all NCAs samples exhibited an improved release profile, K1ML recorded the highest dissolution rate. A similar trend was observed in a previously published study [215]. Hence, the release profiles are affected by the size, surface area, crystallinity, and roles of mannitol and leucine in influencing dissolution rates through their effects on wettability and particle dispersion. The interplay of these factors resulted in slight variations in the release rates observed for NCAs samples.



**Figure 20:** *In vitro* release study of physical mixture (PM) and developed NCAs (K1, K1M, K1L, and K1ML) in simulated lung media (SLM). Results are expressed as mean  $\pm$  SD (n=3 independent measurements) [3].

### 6.3.7. Aerosol performance

#### 6.3.7.1. *Andersen cascade impactor*

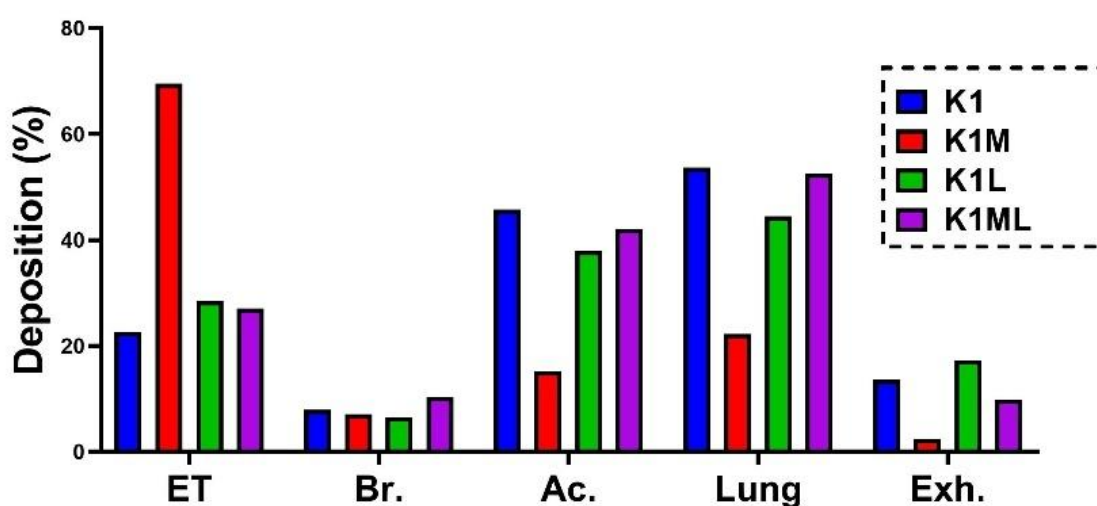
The ACI was used at 60 L/min to assess aerodynamic properties (Table 8). MMAD values ranged from 0.927 to 4.5  $\mu\text{m}$ , indicating suitability for lung deposition. Mannitol in K1M increased MMAD and GSD, limiting deep lung delivery and suggesting broad size distributions. K1L showed the highest FPF due to leucine's dispersibility enhancement, but its low MMAD ( $<1 \mu\text{m}$ ) raised exhalation risk. K1ML balanced MMAD and FPF, favoring deep lung deposition, consistent with a previous study [212]. EF values (92.67–97.05%) confirmed efficient delivery.

### 6.3.7.2. Aerosol particle sizer

In line with ACI analysis, APS results showed that K1L had the smallest median and geometric mean, while K1M had larger particle sizes, confirming that mannitol alone may not sufficiently optimize particle size for inhalation (results not shown) [147]. The combination of mannitol and leucine in K1ML resulted in intermediate particle sizes, offering a balance between the two excipients' properties. This was in line with a previous study by Pasero et al., (2025) where the aerosolization properties microparticle system (mannitol and salbutamol sulphate) was significantly improved up to maximum FPF (48 %) and MMAD (2  $\mu$ m) when combined and spray freeze-dried with leucine at 10 % (w/w db) [216]. Another study found a highly significant and positive impact of leucine on enhancing the FPF of spray-dried mannitol particles [217].

### 6.3.7.3. In silico modelling

In silico predictions aligned with *in vivo* data for pulmonary delivery (Figure 21). The 5-second BH model assessed aerosol behavior, matching *in vitro* trends. K1M (mannitol) showed larger particles, lower lung deposition, and minimal exhalation due to size. K1L (leucine) had smaller particles but higher exhaled fraction. K1ML (mannitol + leucine) achieved optimal lung deposition, confirming prior findings [216]. Extended BH (10 s) improved deposition for all formulations. Based on aerosol performance, K1ML was selected for further study, with K1 as reference.



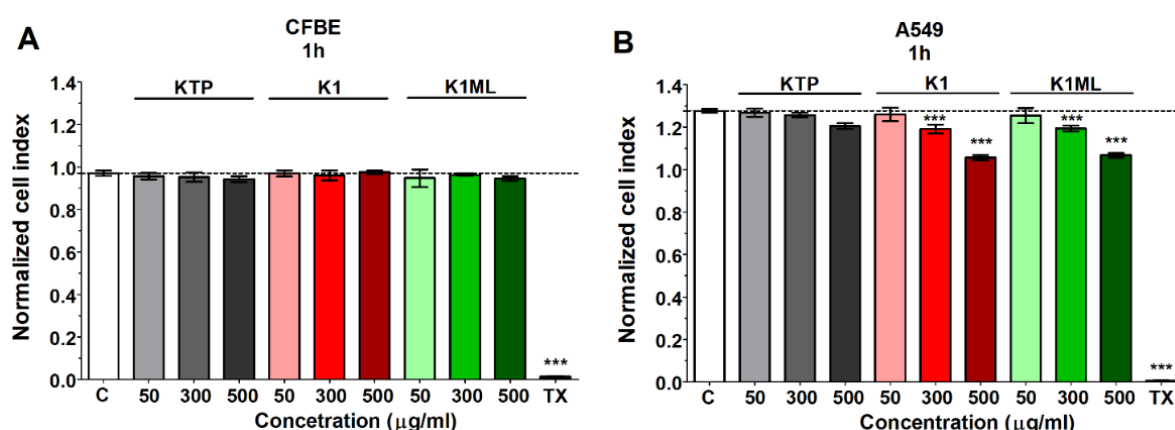
**Figure 21:** *In silico* results of deposited mass of NCA samples at a 5-second BH. ET: extra-thoracic, Br: bronchial, Ac: acinar, and EXH: exhaled [147].

### 6.3.8. Cell line measurements

#### 6.3.8.1. Cell viability

While no significant changes were observed over a 1-hour period in CFBE (Figure 22A), a slight reduction in impedance was observed at high concentrations in A549 (treated with K1 and K1ML) (Figure 22B).

Hence, 50  $\mu\text{g/mL}$  KTP concentration was selected for subsequent permeability experiments.

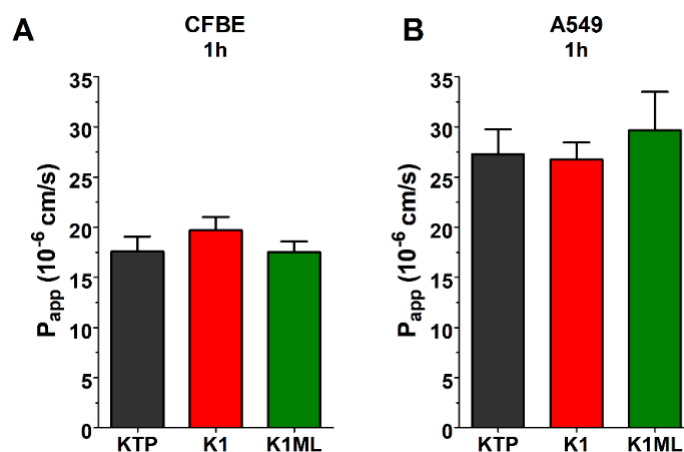


**Figure 22:** Cell viability of (A) CFBE bronchial epithelial and (B) A549 alveolar epithelial cells after 1-hour treatments with raw KTP, K1 and K1ML samples measured by impedance. Values are presented as means  $\pm$  SD,  $n = 5-6$ . \*\*\* $p < 0.001$  compared to the C (control group) [3].

#### 6.3.8.2. Cell permeability

The permeability of raw KTP and NCA formulations (K1, K1ML) was evaluated using bronchial and alveolar co-culture models over 1 hour (Figure 23). KTP showed efficient penetration, with higher  $P_{\text{app}}$  values in the A549 alveolar model, suggesting better absorption potential in the alveolar region [218,219].

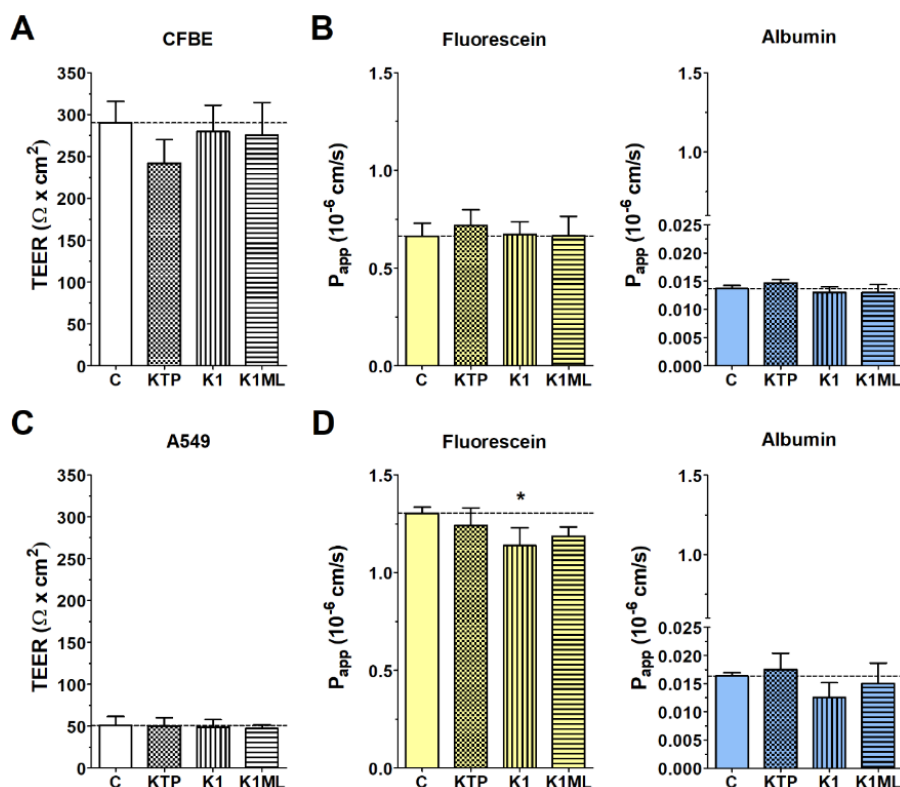
K1ML slightly increased permeability in A549 cells, while K1 showed higher permeability in CFBE cells, indicating excipient effects vary by cell type [220]. However, neither formulation significantly altered KTP permeability compared to the raw drug, which is considered a favorable outcome. As KTP is a BCS Class II drug, the results suggest that the formulations may improve solubility without impairing epithelial permeability.



**Figure 23:** Permeability results of raw KTP, K1, and K1ML across co-culture models (A) CFBE cells and (B) A549 alveolar cells after a 1-hour assay. Data are expressed as mean  $\pm$  SD, n = 4 [3].

### 6.3.8.3. Barrier integrity

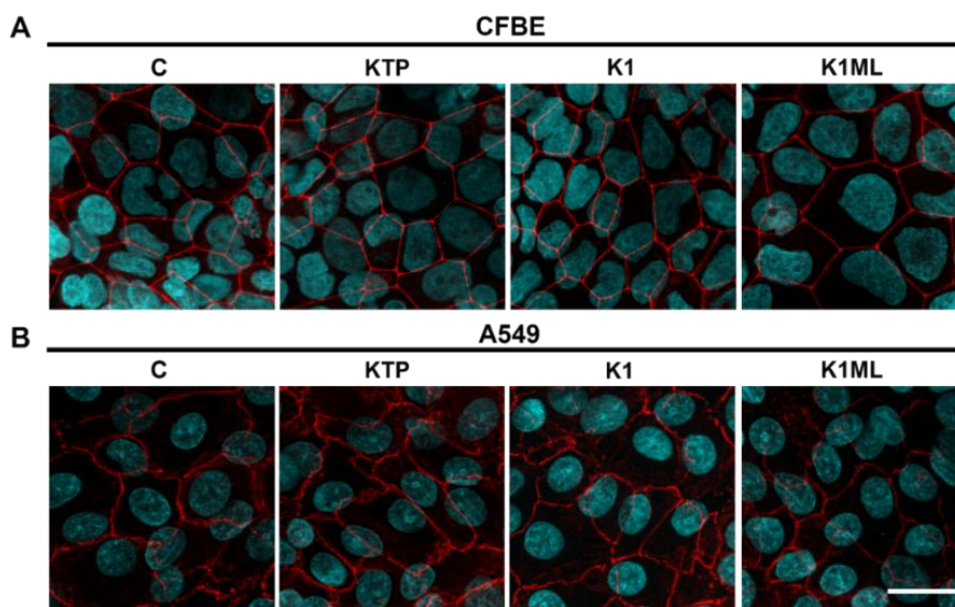
Neither KTP nor NCA formulations (K1, K1ML) significantly affected barrier integrity, as confirmed by unchanged TEER values and low  $P_{app}$  for fluorescein and albumin (Figure 24). Fluorescein permeability was higher in A549 than CFBE models, consistent with permeability results. Interestingly, K1 reduced fluorescein permeability and lowered albumin permeability without affecting TEER (in A549), suggesting a barrier-tightening effect, similar to that seen with meloxicam in nasal epithelial models [221].



**Figure 24:** Assessment of barrier tightness in CFBE and A549 co-culture models following permeability assays: (A) and (C) TEER measurements, and (B) and (D) permeability of fluorescein and albumin marker molecules. Data are expressed as means  $\pm$  SD, n = 4. \*p < 0.05. C: control [3].

#### 6.3.8.4. Immunocytochemistry

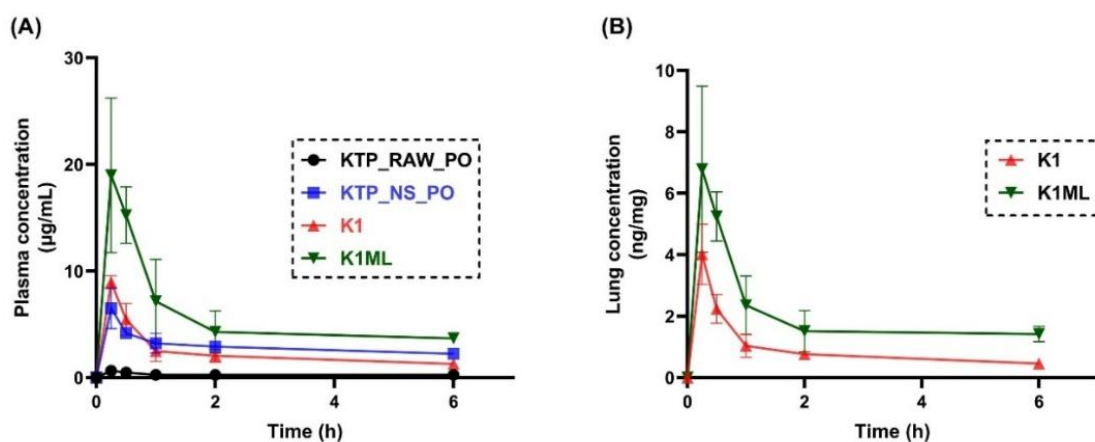
Immunostaining for ZO-1 (CFBE) and  $\beta$ -catenin (A549) was performed to assess cell morphology after exposure to raw KTP, K1, and K1ML (Figure 25). Strong, continuous staining was observed at cell junctions in both models, with no differences across groups. Morphological analysis confirmed that neither KTP nor NCA formulations compromised barrier integrity [3].



**Figure 25:** Barrier integrity assessed by immunostaining in (A) CFBE cells (ZO-1 staining) and (B) A549 cells ( $\beta$ -catenin staining) after permeability. Cyan: cell nuclei; red: junctional proteins. Bar = 20  $\mu$ m [3].

#### 6.3.9. Pharmacokinetic study in healthy rats

PK studies of KTP-RAW, KTP-NS, K1, and K1ML were conducted, and the concentration-time profiles in plasma and lungs are shown in Figure 26, with PK parameters in Table 9.



**Figure 26:** The concentration-time curve of KTP (A) in plasma after oral administration of KTP-raw and KTP-NS and intratracheal administration of K1 and K1ML, (B) in lungs after intratracheal administration of K1 and K1ML. Administered dose was 3 mg KTP/kg. Data are expressed as mean  $\pm$  SD (n = 4) [3].



In plasma samples, K1ML showed 779.9% relative bioavailability vs. KTP-RAW, outperforming KTP-NS (451.9%) and K1 (258.6%). Inhalation delivery further increased the  $C_{\max}$  values compared to orally delivered KTP-NS, by 1.4-fold for K1 and 3-fold higher than oral KTP-NS. Moreover, K1ML revealed superior  $AUC_{0-\infty}$  compared to KTP-NS (72.76 vs 42.16  $\mu\text{g}\cdot\text{h}/\text{mL}$ ). For comparison, K1ML's performance exceeded literature reports: its bioavailability (~73%) surpassed Huang et al.'s inhaled cyclosporine particles (44-48%) and Hou et al.'s IV KTP nanoparticles (52.04 %) [222,223].

The mannitol-leucine combination (K1ML) enabled 1.7-fold higher absorption than oral KTP-NS, demonstrating inhalation's advantages for systemic delivery with reduced gastric risks.

**Table 9:** The pharmacokinetic parameters of KTP obtained from both plasma and lung concentration-time curves by non-compartmental analysis after oral administration of KTP-RAW and KTP-NS, and intratracheal administration of NCA formulations (K1 and K1ML), (Mean  $\pm$  SD, n = 4) [3].

	PK parameter	KTP-RAW	KTP-NS	K1	K1ML
Plasma	$k_e$ ( $\text{h}^{-1}$ )	0.0653 $\pm$ 0.0423	0.0919 $\pm$ 0.0164	0.1269 <sup>a</sup>	0.1366 <sup>a</sup>
	$t_{1/2}$ (h)	21.306 $\pm$ 4.509	7.720 $\pm$ 1.354 <sup>a</sup>	5.460 <sup>a</sup>	5.072 <sup>a</sup>
	$t_{\max}$ (h)	0.25	0.25	0.25	0.25
	$C_{\max}$ ( $\mu\text{g}/\text{mL}$ )	0.646 $\pm$ 0.098	6.499 $\pm$ 1.9122	8.949 $\pm$ 0.611 <sup>a</sup>	18.973 $\pm$ 7.246 <sup>a,b,c</sup>
	$AUC_{0-\infty}$ ( $\mu\text{g}\cdot\text{h}/\text{mL}$ )	9.329 $\pm$ 8.897	42.163 $\pm$ 3.150 <sup>a</sup>	24.125 <sup>a,b</sup>	72.763 <sup>a,b,c</sup>
	F (to KTP_RAW)	-	4.519	2.586	7.799
	F (to KTP_NS)	0.221	-	0.572	1.726
Lung	$k_e$ ( $\text{h}^{-1}$ )	-	-	0.1535	0.1671
	$t_{1/2}$ (h)	-	-	4.515	4.147
	$t_{\max}$ (h)	-	-	0.25	0.25
	$C_{\max}$ ( $\mu\text{g}/\text{mg}$ )	-	-	0.00402 $\pm$ 0.00803	0.00678 $\pm$ 0.0136
	$AUC_{0-\infty}$ ( $\mu\text{g}\cdot\text{h}/\text{mg}$ )	-	-	0.00843	0.02037
	$f_t$	-	-	0.0003912	0.0003241

$k_e$ : elimination rate constant;  $t_{1/2}$ : elimination half-life;  $C_{\max}$ : maximum KTP concentration;  $t_{\max}$ : time to reach  $C_{\max}$ ;  $AUC_{0-\infty}$ : area under the zero to infinity concentration curve; F: relative bioavailability;  $f_t$ : tissue availability. <sup>a</sup>: indicates statistically significant difference from KTP-RAW ( $P < 0.05$ ). <sup>b</sup>: indicates statistically significant difference from KTP-NS ( $P < 0.05$ ). <sup>c</sup>: indicates statistically significant difference from K1 ( $P < 0.05$ ).

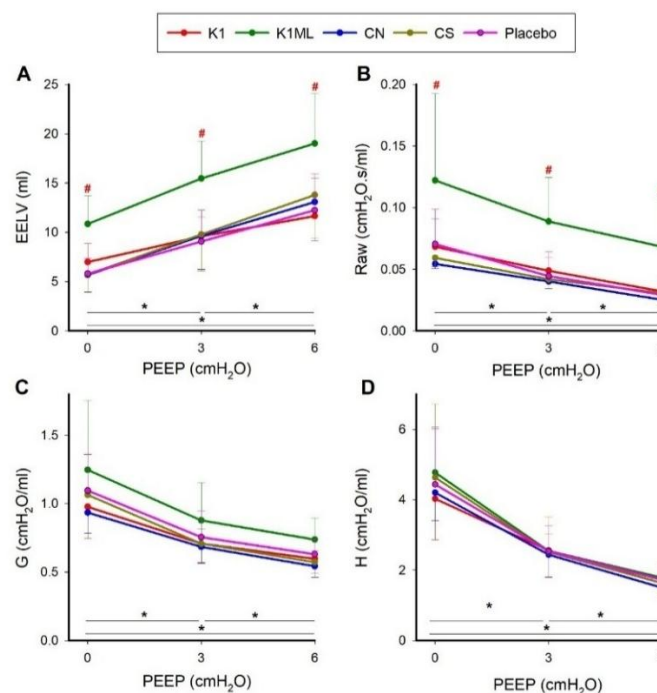
In lung samples, inhalation delivery resulted in minimal lung retention (ng-range KTP), with no significant differences in elimination kinetics between K1 and K1ML (Figure 26B). Both formulations showed rapid systemic absorption (low  $f_t$  values). While K1ML had higher (but non-significant)  $C_{\max}$  and  $AUC_{0-6}$  vs. K1, its significantly greater  $AUC_{0-\infty}$  suggests potentially prolonged exposure and increased local effect risk.

### 6.3.10. Safety and tolerance in ovalbumin-sensitized rats

#### 6.3.10.1. Lung function and respiratory mechanics measurement

The effect of increasing PEEP on four parameters (EELV, Raw, G, and H) was evaluated. As a result, increasing PEEP elevated EELV, while reduced Raw, G and H (all  $p < 0.001$ ), validating experimental consistency. Notably, Group K1ML exhibited significantly higher EELV and Raw ( $p < 0.001$ ) compared to other groups, suggesting airway narrowing and expiratory flow limitation. No differences were observed among other groups (Figure 27). The elevated Raw and EELV in K1ML may stem from:

- (i) Enhanced KTP bioavailability, leading to heightened COX inhibition and a shift toward leukotriene production, which promotes bronchoconstriction [224–227]. Jin et al. (2013) reported lung injury in rats at high COX-2 inhibitor doses, suggesting that reducing the K1ML dose could prevent adverse effects [228].
- (ii) Excipient-related effects: Mannitol, while aiding mucociliary clearance, may induce dose-dependent bronchoconstriction [229,230], potentially managed with bronchodilators like formoterol [231]. Leucine, at higher doses, might impair surfactant function or irritate airways [232]. These findings support exploring NSAID-bronchodilator combinations.



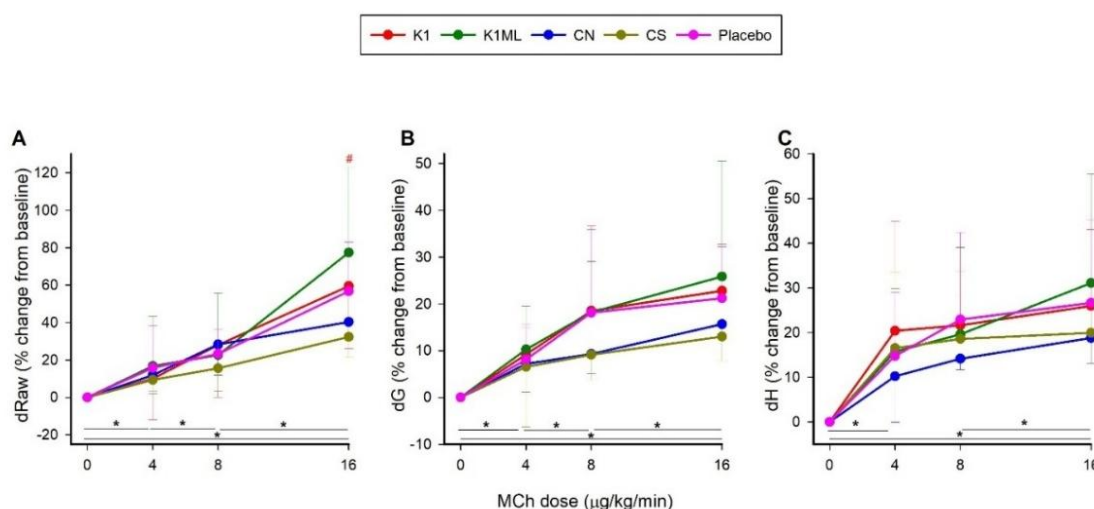
**Figure 27:** Lung function parameters in OVA-sensitized rats across varying PEEP levels: (A) End-expiratory lung volume (EELV), (B) airway resistance (Raw), (C) respiratory tissue damping (G) and (D) respiratory tissue elastance (H). Where \*:  $p < 0.05$  for PEEP change in all groups and #:  $p < 0.05$  K1ML vs. other groups [3].



### 6.3.10.2. Airway responsiveness to pharmacological challenges

- **Methacholine (MCh) Challenge**

The non-specific muscarinic agonist MCh induced dose-dependent increases in airway resistance (Raw) and respiratory tissue mechanics (G and H) ( $p < 0.001$  for all parameters), confirming effective bronchoconstriction (Figure 28). No significant differences were observed between treatment groups, indicating none of the interventions affected lung responsiveness to this non-specific agonist.



**Figure 28:** Changes during methacholine (MCh)-induced bronchoprovocation in (A) airway resistance (Raw), (B) respiratory tissue damping (G), and (C) respiratory tissue elastance (H) in naïve untreated animals (CN,  $n = 6$ ), OVA-sensitized untreated animals (CS,  $n = 6$ ), and OVA-sensitized animals receiving intratracheal administration of placebo ( $n = 12$ ), K1 ( $n = 12$ ), or K1ML ( $n = 12$ ) formulations. Where \*:  $p < 0.05$  for MCh dose change in all groups, #:  $p < 0.05$  K1ML vs. CN, CS [3].

- **Ovalbumin (OVA) Challenge:**

The specific allergen challenge confirmed successful sensitization, with sensitized animals showing pronounced bronchoconstriction responses ( $>25\%$  peak Raw increase) compared to non-sensitized controls ( $<10\%$  response).

Similar to MCh results, no differences were observed among sensitized treatment groups, demonstrating that neither K1 nor K1ML formulations altered the lung's response to specific allergen challenge (results not shown).

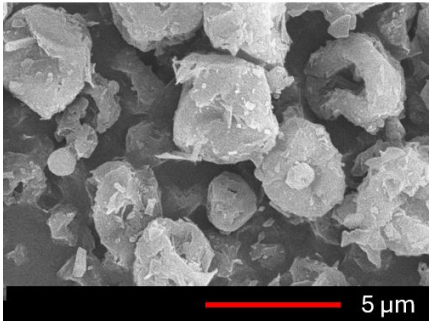
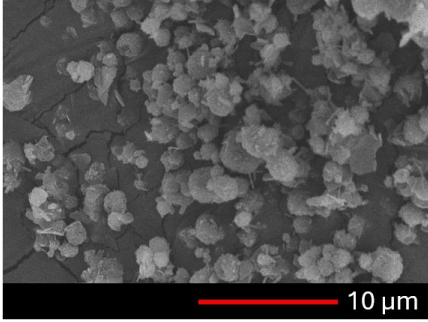
#### 6.4. Stability assessment of selected samples

Table 10 presents the 3-month stability results of F1 and K1ML under stress conditions. F1 showed a significant increase in span, along with reduced FPF and a compromised release profile—likely due to solid-state changes or increased interparticle interactions reducing surface area [233], which is in line with previously published results [234].

In K1ML, D[0.5], span, and MMAD remained stable, but FPF declined, likely due to moisture-induced surface changes such as roughness or partial aggregation. This aligns with findings by Das et al. (2009), who observed a drop in FPF of salmeterol xinafoate from 11.3% to 4.9% with rising RH [235]. Despite FPF changes, release profile and DSC analyses remained consistent with partially preserved crystallinity, suggesting that the formulation maintains consistent performance throughout its intended shelf life under normal storage conditions.

Future studies should focus on investigating stability under different stress conditions to address moisture sensitivity observed on the particle surface.

**Table 10:** Stability of selected formulations (F1 and K1ML) after 3-month storage under accelerated conditions [90,147].

Analysis	F1	K1ML
D[0.5] (μm)	2.44	2.19
Span	121.28	1.42
FPF (%) <5 μm	26.6	33.10
MMAD	5.43	3.52
Crystallinity (%)	33.81	40.15
Melting point (°C)	90.39	91.07
Morphology		
	Nearly spherical Rougher corrugated surface	Spherical Fiber-like surface
Release	75% in 5 min	90% in 5 min

## 7. Conclusion

This dissertation employed a patient-centric strategy to address distinct therapeutic needs—local and systemic—through pulmonary delivery using KTP-based DPIs. Two formulations were developed from a unified nanocrystal platform: one designed to treat lung inflammation and mucus obstruction in CF, and the other optimized for systemic delivery with enhanced bioavailability. The main findings, organized according to the defined objectives, are summarized as follows:

### A. Nanocrystal dispersion preparation

- I. A stable nanocrystal suspension was successfully optimized using 1% PVA and 0.1% SDS. Over 45 days of refrigerated storage, only minimal particle size growth was observed, while size distribution remained consistent, qualifying the dispersion for downstream solidification and engineering.

### B. Local pulmonary delivery (nanocrystal KTP-mannitol combination therapy)

- II. A QbD framework supported the rationale for combining KTP with mannitol in a single inhaler to address both inflammation and mucus accumulation in CF, aiming to improve disease control and patient quality of life.
- III. The optimized formulation (F1; containing KTP and mannitol in 1:1 mass ratio, with 84.9% DL) demonstrated efficient aerosol performance, with deposition primarily in ACI stages 3 and 4, confirming desired lung delivery. It achieved complete drug release (~100%) within 10 minutes in simulated lung fluid, notably reduced mucin viscosity without compromising KTP's anti-inflammatory effect, and showed good biocompatibility in alveolar and bronchial cells. A 1.4-fold increase in alveolar permeability was achieved. However, deviations from the 1:1 ratio (either higher or lower) negatively impacted performance.
- IV. These findings highlight the formulation's potential as a dual-function inhalable therapy for CF, though precise mannitol concentration remains critical to maintaining efficacy and performance. This novel inhaled combination supports patient-centric delivery by reducing doses burden, improving symptom control, and enhancing adherence.

### C. Systemic pulmonary delivery (nanocrystal-engineered DPI)

- V. A factorial design (BBD) optimized nano spray-drying parameters (50% pump rate, 5% (w/v) feed concentration and 90 °C inlet temperature), enabling the production of KTP engineered-nanocrystal agglomerates tailored for systemic absorption via the deep lung deposition.
- VI. The inclusion of both mannitol and leucine demonstrated a synergistic effect on particle engineering, significantly enhancing aerosol behavior and dispersion compared to formulations with either excipient alone. The optimized mannitol-leucine formulation (K1ML; containing 2.5% KTP, 1.25% mannitol and 1.25% leucine (w/v)) achieved excellent *in vitro* aerosol performance (MMAD:  $2.66 \pm 0.16 \mu\text{m}$ ; FPF:  $69.3 \pm 3.67\%$ ) and complete drug release (99% in 5 min). It also maintained biocompatibility across both CFBE and A549 cells, with greater permeability in alveolar cells.
- VII. Pharmacokinetic evaluation of inhaled K1ML confirmed a 3-fold increase in  $C_{\text{max}}$  and 1.73-fold improvement in relative bioavailability (73% for K1ML vs. 42% for oral KTP-NS). Low lung tissue retention indicated efficient systemic absorption, supporting its suitability for extra-pulmonary indications and rapid systemic absorption. However, repeated dosing in an OVA-sensitized rat model revealed bronchial side effects (e.g., airway narrowing and reduced expiratory flow) for K1ML, indicating that dose adjustment is required to balance efficacy and pulmonary safety. The nanocrystal-engineered DPI system aligns with patient-centric goals by improving bioavailability, reducing gastrointestinal-related side effects, accelerating therapeutic onset, and potentially lowering dosing frequency.

### D. Stability assessment

- VIII. Over three months of ICH storage (40°C/75% RH), F1 exhibited a broadened particle size distribution (span), reduced FPF, and a compromised release profile, indicating the need for further optimization of mannitol concentration or incorporation of an additional stabilizing agent. In contrast, K1ML maintained its particle size, MMAD, and *in vitro* release characteristics, demonstrating robust stability and greater suitability for potential commercial development.

## 8. Novelty and practical relevance of the work

The main innovations and practical contributions of this study include:

- First application of a unified KTP nanocrystal suspension engineered to serve both local and systemic delivery via inhalable powders, developed using a patient-centric strategy.
- Novel QbD-driven Zero Phase formulation rationale proposed for a DPI combination therapy in CF, supporting dual functionality—anti-inflammatory action (via KTP) and mucus hydration (via mannitol).
- First demonstration of mannitol–leucine synergy in enhancing aerosolization, cellular permeability, and systemic bioavailability of a nanocrystal-engineered DPI.
- Development of the first NSAID-based DPI intended specifically for systemic absorption, offering a needle-free alternative to injectable and oral routes.
- First-time comprehensive assessment of potential respiratory risks associated with prolonged inhaled NSAID (KTP) use, using an OVA-sensitized rat model to evaluate lung function and airway mechanics in an asthmatic-like condition.
- Integration of KTP and mannitol in a single DPI formulation to address polytherapy needs in CF, by targeting inflammations and mucus overproduction simultaneously, aiming to improve adherence and simplify treatment.
- Implementation of a scalable and clinically relevant spray-drying process aligned with patient usability and regulatory trends in patient-focused drug design.
- Advancement of pulmonary drug delivery as a route for repurposing poorly soluble drugs, with potential for broad application to other therapeutic classes such as antivirals or immunomodulators.
- Supports regulatory interest in IVIVC and patient-centricity by demonstrating rational formulation design tied to improved therapeutic performance and usability.
- These innovations represent an important shift toward more effective and patient-aligned therapies.

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