

Methane bioactivity and interactions with other biological gases

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Ph.D. Thesis

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- II. **Mészáros A**, Büki T, Varga G, Tőkés T, Kaszaki J, Boros M: Effects of the methane treatment on the early morphological and functional consequences of the experimental mesenteric ischaemia and reperfusion. *Archives of the Hungarian Medical Association of America* Vol 20 (2): 49, 2012.
- III. **Mészáros AT**, Büki T, Varga G, Tőkés T, Kaszaki J, Boros M: Effects of Methane on the Functional and Morphological Consequences of Experimental Mesenteric Ischemia and Reperfusion. *European Surgical Research* 50 (S1): 1, 2013.
- IV. **Mészáros A**, Fischer-Szatmári T, Szűcs Sz, Varga G, Boros M: A vékonybél nyálkahártya változásai az ischaemia-reperfúzió folyamán – megbízható-e a szövettani eredmény? *Magyar Sebészet* 66 (2): 99, 2013.
- V. Érces D, Nógrády M, **Mészáros A**, Varga G, Ghyczy M, Kaszaki J, Boros M: Methane Inhalation in the Early Phase of Resuscitation Reduces Intestinal Injury After Experimental Cardiac Tamponade. *Shock* 40, Suppl. 1: 23, 2013.
- VI. Érces D, Nógrády M, **Mészáros A**, Varga G, Földesi I, Ghyczy M, Kaszaki J, Boros M: Decreased intestinal injury after resuscitation with methane inhalation during experimental cardiac tamponade. *Cardiologica Hungarica* 43, G12, 2013.
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- V. Strifler G, **Mészáros A**, Chun C, Kaszaki J, Boros M: Metán kezelés hatása a mitokondriális működésre anoxia-reoxigenizáció után. *44. Membrán-Transzport Konferencia, Sümeg, 2014. május 20- 23.*
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- VII. Boros M, Tuboly E, Hartmann P, Strifler G, **Mészáros A**: Methane and mitochondria. *A Magyar Élettani Társaság 79. Vándorgyűlése, Szeged, 2015. május 27-30.*
- VIII. **Mészáros AT**, Büki T, Varga G, Kaszaki J, Boros M: Methane Treatment Improves Intestinal Morphology and Function in Experimental Mesenteric Ischemia and Reperfusion. *39th Seminar of the Austrian Society for Surgical Research, Wagrain, Austria. 2015. november 19-21.*

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Contents

Contents.....	3
List of abbreviations.....	5
Summary	7
Összefoglalás.....	9
I. Introduction	11
1. Oxygen and aerobic life	11
1.1. Oxygen physiology, normoxia.....	11
1.2. The mitochondrial ETS and O ₂	12
1.3. Sources and sinks of mitochondrial reactive oxygen species (ROS).....	12
2. Hypoxia and reoxygenation, ischemia and reperfusion, pathological ROS generation	13
2.1. Hypoxia and ischemia.....	13
2.2. Reperfusion injury	14
3. Hypoxia and inflammation.....	15
3.1. Inflammatory mediators.....	16
3.3. The intestinal atmosphere and the epithelial barrier	18
4. Biologically active gases	19
4.1. Nitric oxide (NO).....	19
4.2. Carbon monoxide (CO).....	20
4.3. Hydrogen sulfide (H ₂ S).....	20
4.4. Methane (CH ₄)	20
4.5. Relationship between O ₂ and other gases	22
5. Mitochondrion: the hub of interactions of gaseous compounds	24
II. Aims	27
III. Materials and methods	28
1. An <i>in vivo</i> study to investigate the effects of CH ₄ treatments in mesenteric IR	28
1.1. Animals	28
1.2. Surgical procedure and study design	28
1.3. Mucosal permeability measurements.....	29
1.4. Morphological analysis	30
1.5. Hemodynamics	31
1.6. Biochemical measurements	31
1.7. CH ₄ concentration measurement.....	32
2. <i>In vitro</i> and <i>ex vivo</i> studies for the detection of NO generation.....	33
2.1. Animals	33
2.2. LPS treatment and study design.....	33
2.3. EPR spectroscopy	34

2.4. NO trapping	34
3. An <i>in vitro</i> microrheological study	35
3.1. Study design, induction of oxidative stress and CH ₄ treatment.....	35
3.2. RBC deformability	35
3.3. RBC aggregation.....	36
4. <i>In vitro</i> analysis of anoxic NO release	36
4.1. Sample preparation	36
4.2. NO measurement protocol	36
5. Statistical analysis	36
IV. Results	38
1. The kinetics of CH ₄ transport.....	38
2. The small intestinal permeability	39
3. The detection of NO by EPR without exogenously added spin-trapping molecules	45
4. The effects of CH ₄ on NO release under ischemia.....	47
5. The effects of CH ₄ on the microhemorheological parameters of whole blood.....	49
V. Discussion	51
1. The kinetics of CH ₄ transport <i>in vivo</i>	51
2. The bioactivity of exogenous CH ₄ in the intestine.....	52
3. The detection of NO in biological samples	55
4. The interaction of CH ₄ with NO-related nitrosative stress	56
5. The direct action of CH ₄ on erythrocyte deformability.....	58
VI. Summary of the new findings	60
VII. Acknowledgements	61
VIII. List of references	62
IX. Annex	73

List of abbreviations

AJ	adherent junction
bw	body weight
CI, II, III, IV	mitochondrial electron transport chain complex I-IV
CH ₄	methane
CK	creatine kinase
CLSEM	confocal laser scanning endomicroscopy
CO	carbon monoxide
CO ₂	carbon dioxide
$\Delta\psi_m$	mitochondrial membrane potential
DETC-Fe	iron-diethylthiocarbamate
DMSO	dimethylsulfoxide
DNIC	dinitrosyl-iron complexes (see also NO-Fe)
e/i/nNOS	endothelial/inducible/neuronal nitric oxide synthase
Em	emission
EP	epithelial permeability
EPR	electron paramagnetic resonance
ET-1	endothelin-1
ETS	mitochondrial electron transport system
Ex	excitation
FD4	4 kDa fluorescein isothiocyanate-dextran
Fe	iron
GI	gastrointestinal
H ₂ S	hydrogen sulfide
Hb	hemoglobin
IL	interleukin
IMM	inner mitochondrial membrane
ip	intraperitoneal
IR	ischemia-reperfusion
K _M	Michaelis constant
LPS	lipopolysaccharide
MPO	myeloperoxidase
mPTP	mitochondrial permeability transition pore

mtROS	mitochondrial reactive oxygen species
MW	molecular weight
N ₂	nitrogen
NO	nitric oxide
NO ₂ ⁻	nitrite ion
NO ₃ ⁻	nitrate ion
NO-Fe	dinitrosyl-iron complex
NO-Hb	mononitrosyl-hemoglobin complex
O ₂ ^{•-}	superoxide radical
OMM	outer mitochondrial membrane
ONOO	peroxynitrite
OXPPOS	oxidative phosphorylation
PAS	photoacoustic spectroscopy
PMN	polymorphonuclear leukocyte
PMS	phenazine methosulfate
RBC	red blood cell
RNS	reactive nitrogen species
ROS	reactive oxygen species
RSNO	S-nitrosothiol
sGC	soluble guanylate cyclase
SMA	superior mesenteric artery
TJ	tight junction
TNF-α	tumor necrosis factor alpha
TLR4	Toll-like receptor 4
VP	vascular permeability
XDH	xanthine dehydrogenase
XO	xanthine oxidase
XOR	xanthine oxidoreductase

Summary

Today it is recognized that a variety of enzymes are generating gas mediators, molecules that exert important physiological functions. Also, there is a growing body of evidence which suggests that methane (CH₄) can be produced in the eukaryotic cell. The anti-inflammatory potential of exogenously supplemented CH₄ has been repeatedly confirmed as well, but important details of the mechanism of action still need to be clarified. The main purpose of the thesis was to summarize and interpret previous data on the *in vivo* properties of exogenous CH₄ in order to design and conduct further experiments, which help to extend our knowledge on the possible role and mechanisms of action of CH₄ in mammals.

Therefore, our first general aim was to explore the effects of normoxic CH₄ administration in a small animal model of mesenteric ischemia-reperfusion (IR) induced inflammation. The early and later consequences of intestinal IR were characterized by mucosal permeability, biochemical, macro- and microcirculatory and morphological parameters. The IR-induced circulatory and structural damage was accompanied by increased epithelial permeability, demonstrating mucosal barrier damage. An assessment of the kinetics of the inhaled CH₄ revealed that in our inhalation regime, at the end of ischemia significant amounts of CH₄ are already in the systemic circulation, allowing rapid equilibration with the intestinal tissue early in the reperfusion, as confirmed by photoacoustic spectroscopy measurements. Normoxic CH₄ inhalation effectively prevented the elevation of intestinal epithelial permeability, maintaining structural integrity of the mucosa and improving biochemical signs of inflammation in the early reperfusion. These data support previous findings on CH₄ bioactivity and establish a mucosa-protective role for exogenous CH₄ to modulate the IR-induced pro-inflammatory activity locally in the small intestine.

Our second general aim was to delineate the mechanisms of action of CH₄ in models of tissue hypoxia with elevated levels of reactive oxygen and nitrogen species, with special emphasis on the possible relationship of CH₄ with other biological gases, such as nitric oxide (NO).

Based on the above results two hypotheses for the mechanism of action were formulated and tested. First, we supposed that CH₄ is able to directly influence cell membrane rigidity. Accordingly, we set out to investigate the influence of CH₄ on changes in erythrocyte deformability provoked by oxidative stress *in vitro*. The results revealed that CH₄ *in vitro* improves the deformability of red blood cells during simulated oxidative stress, which may

contribute to the improvement of the effectiveness of mesenteric microcirculation in the postischemic small intestine.

The second hypothesis of action of exogenous CH₄ was a direct interaction with NO metabolism. A new, non-toxic, exogenous spin trap-free electron paramagnetic resonance (EPR) spectroscopy method was developed for NO detection. The method was validated in a bacterial endotoxin-based *in vivo* experimental model of inflammation where NO concentration was measured from both the intra- and extracellular compartments of the same samples in tissue biopsies. With EPR spectroscopy reduced NO levels were detected in the ischemic intestinal tissue after normoxic CH₄ administration. Considering this information, we set out to measure NO release in real-time *in vitro* from liver tissue homogenate under anoxic conditions, mimicking ischemic tissue. CH₄ treatment under such conditions inhibited NO release. Collectively, these results confirm that exogenous CH₄ administration effectively reduces NO levels in the ischemic tissue.

In conclusion, our data demonstrates an anti-inflammatory and mucosa-protecting role of CH₄ in intestinal IR. To shed light on the mechanism of action, we confirmed direct effects of CH₄ on oxidized erythrocyte membranes and that increased CH₄ input reduces NO production and nitrotyrosine levels in hypoxic and anoxic organs.

Összefoglalás

Napjainkra számos enzimről bizonyosodott be, hogy gáznemű jelátvivő molekulák termelésére képesek. Ezen mediátorok fontos élettani funkciókat szabályoznak. Egyre több bizonyíték szól amellett, hogy eukarióta sejtekben metán (CH_4) képződhet. A kívülről bejuttatott CH_4 gyulladásgátló hatását is többször megerősítették, azonban a hatásmechanizmus fontos részletei még tisztázatlanok. E dolgozat fő célja az exogén CH_4 *in vivo* hatásairól rendelkezésre álló adatok összegzése és értelmezése, majd további kísérletek tervezése és kivitelezése volt, melyek gyarapíthatják tudásunkat a CH_4 emlősökben betöltött szerepéről.

Első fő célunk a normoxiás CH_4 bevitel hatásainak és következményeinek vizsgálata volt mesenterialis ischaemia-reperfúzió (IR) által okozott gyulladásos reakció alatt. A vékonybél IR korai és késői következményeit a nyálkahártya permeabilitás meghatározásával, biokémiai, makro- és mikrohaemodinamikai valamint morfológiai vizsgálatokkal jellemeztük. A vékonybél IR mikrokeringési és strukturális károsodása megemelkedett epitheliális permeabilitással járt, jelezve a mucosa barrier károsodását. A belélegzett CH_4 időbeli szöveti eloszlásának fotoakusztikus spektroszkópos vizsgálata feltárta, hogy az általunk alkalmazott lélegeztetési módszerrel már az ischaemia végén jelentős mennyiségű CH_4 van a szisztémás keringésben, mely a bélben a koncentráció gyors kiegyenlítődését teszi lehetővé a reparfúzió elején. Normoxiás CH_4 belélegzése hatékonyan gátolta a vékonybél epitheliális permeabilitás növekedését, megőrizte a mucosa strukturális épségét és mérsékelte a gyulladás biokémiai jeleit a korai reparfúzió során. Az adatok megerősítik a CH_4 bioaktivitását bizonyító adatokat és megalapozzák az exogén CH_4 gyulladáscsökkentő, nyálkahártya-védő hatását IR alatt.

Másik fő célunk a CH_4 hatásmechanizmusának további feltárása volt szöveti hypoxiával, valamint fokozott reaktív oxigén és nitrogén származék képződéssel jellemzett modellekben. Ennek során a CH_4 egyéb gázokkal, pl. nitrogén-monoxiddal (NO) való kölcsönhatásaira fektettünk külön hangsúlyt.

A fenti eredmények alapján két elméletet dolgoztuk ki és vizsgáltuk meg. Elsőként azt feltételeztük, hogy a CH_4 közvetlenül képes a sejtmembrán-rigiditás befolyásolására. Ehhez megvizsgáltuk a CH_4 *in vitro* hatását az oxidatív stressz által kiváltott vörösvérsejt deformabilitás változásokra. Eredményeink szerint *in vitro* szimulált oxidatív stressz alatt a CH_4 javítja a vörösvérsejtek deformabilitását, mely ezáltal hozzájárulhat a mesenterialis mikrokeringés hatékonyságának javulásához ischaemiát követően.

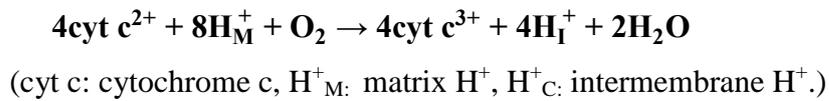
A második elmélet szerint az exogén CH₄ közvetlenül képes befolyásolni a NO metabolizmust. Egy új, nem toxikus, exogén spincsapda-mentes, elektron paramágneses rezonancia (EPR) spektroszkópos módszert dolgoztunk ki a NO mérésére. A módszert validáltuk egy bakteriális endotoxin-alapú gyulladás állatmodelljében ugyanazon szövetbiopsziából történő intra- és extracelluláris NO meghatározásra is. EPR spektroszkópiával csökkent NO szintet mutattunk ki vékonybél szövetben *in vivo* normoxiás CH₄ kezelés után. Ezen eredményeket megfontolva további vizsgálatokat végeztünk a NO felszabadulás valós idejű mérésére máj szövet homogenizátumból, anoxiás körülmények között, ischaemiás szövetet utánozva. A CH₄ kezelés ezen körülmények között gátolta a NO felszabadulást. Összességében ezen eredmények igazolják, hogy exogén CH₄ alkalmazása csökkenti az ischaemiás szövet NO szintjét.

Összefoglalva, adataink a CH₄ gyulladásgátló és nyálkahártya védő hatását bizonyítják vékonybél IR során. Hogy fényt derítsünk a hatásmechanizmusra, kimutattuk, hogy a CH₄ közvetlenül befolyásolja az oxidált vörösvérsejt membránt és hogy megnövekedett CH₄ bevitel csökkenti az anoxiás és hypoxiás szövetek NO termelését és a nitrotirozin szintet.

I. Introduction

1. Oxygen and aerobic life

In most vertebrate tissues the majority of oxygen (O₂) is consumed by mitochondria, the powerhouses of cells. During the aerobic metabolism ATP is produced by protein complexes of the electron transport system (ETS) embedded in the inner mitochondrial membrane (IMM). In the last cascade step of the electron transfer fed by the metabolites of the citrate cycle, the final electron acceptor O₂ is reduced at the cytochrome c oxidase (EC 1.3.3.1) producing H₂O:



This process contributes to more than 90% of the total O₂ consumption of the human body. In addition, over 60 enzymes are consumers of the cellular O₂, including oxidases (e.g. xanthine oxidase (EC 1.17.3.2), aldehyde oxidase (EC 1.2.3.1), amine oxidases, NADPH oxidases), and mono- and dioxygenases (e.g. lipoxygenases of the arachidonic acid cascade, cytochrome p450 monooxygenase, etc.) (Vanderkooi *et al.*, 1991). Interestingly, in contrast to most cell types, not mitochondria, but phagocytic NADPH oxidase is the major sink of O₂ in activated polymorphonuclear (PMN) leukocytes during the “respiratory burst” since these cells generally rely on glycolysis (Segal *et al.*, 1978).

1.1. Oxygen physiology, normoxia

Normoxia generally refers to 20-21% O₂ in the atmosphere. Within the human body the O₂ gradient gradually decreases in the respiratory system and across the cells, reaching its lowest value in the vicinity of mitochondria. The partial pressure of O₂ (pO₂) in the lung alveoli is 100-110 mmHg (approx. 13 kPa), 90–110 mmHg (12.0–14.7 kPa) in the arterial blood, well above the normal extracellular pO₂ levels in solid organs (40 mmHg in the neocortex and 10 mmHg in the luminal parts of the colon) (Erecinska *et al.*, 2001; Gnaiger, 2003). Nevertheless, the partitioning factor of O₂ is roughly 3, thus local concentrations of O₂ in cellular and mitochondrial membrane fractions are enhanced (Subczynski *et al.*, 1983).

Here, it should be mentioned that the O₂ level in cell cultures is usually much higher than *in vivo*, because in the majority of studies they are incubated at air oxygen pressure. In this case the O₂ concentration reaches approximately 200 μM (160 mmHg) in the incubation medium at 37°C and ambient pressure, representing relative hyperoxia compared to the *in vivo* situations (Gnaiger, 2003; van Faassen *et al.*, 2009).

1.2. The mitochondrial ETS and O₂

Mitochondria are spatially separated intracellular sites hosting a series of oxidoreductive reactions coupled to proton pumping. The initial shuttle carrier of electrons from the citric acid cycle to the membrane-bound respiratory chain of flavoproteins is the NAD⁺/NADH couple. Complex I (CI, NADH:ubiquinone oxidoreductase, EC 1.6.5.3) feeds electrons in the ubiquinone pool located in the IMM. Subsequently, electrons are carried by Complex III (CIII, cytochrome c reductase, EC 1.10.2.2) to cytochrome c and then to Complex IV (CIV, cytochrome c oxidase, EC 1.9.3.1), where O₂, the final electron acceptor, is reduced. The other main way of electrons converging at the ubiquinone pool is driven by Complex II (CII, succinate dehydrogenase, EC 1.3.5.1), the only membrane-bound enzyme of the citrate cycle. According to the chemiosmotic coupling hypothesis of Mitchell (1961), protons are pumped with the energy of oxidation-reduction at the CI, CIII and CIV across the IMM to the intermembrane space, which is the driving force of the ATP synthesis by the F₀-F₁ ATPase (Mitchell, 2011).

1.3. Sources and sinks of mitochondrial reactive oxygen species (ROS)

According to Vladimir Skulachev “*living with the risk of oxidative stress is a price that aerobic organisms must pay for more efficient bioenergetics*” (Skulachev, 1996). During the electron transfer over the ETS 1-4% of electrons “leak”, producing partially reduced superoxide (O₂^{•-}), the primary reactive oxygen species (ROS) in the eukaryotic cell (Chance *et al.*, 1961). There are at least 10 known enzymes in mammalian mitochondria which are able to produce ROS, some of them at several sites within the same protein complexes (i.e. CI, CII and CIII), these also include the cytochrome b5 reductase, monoamine oxidases A and B, dihydroorotate dehydrogenase, glycerol-3-phosphate dehydrogenase, aconitase, and the α -ketoglutarate dehydrogenase complex (Andreyev *et al.*, 2005; Quinlan *et al.*, 2012). However, most of the experiments that indicate net ROS production of these organelles were done on isolated mitochondria or on submitochondrial particles, usually by means of detection methods, which are based on trapping reactive species by molecules that have a high affinity to ROS, generating detectable adducts upon reaction. Optical and electron spin resonance spectroscopic techniques are commonly used to detect such adducts.

The rate of ROS production strongly depends on the metabolic state of the cells and the contribution of mitochondria to the cellular ROS pool is still debated (Hansford *et al.*, 1997; Tahara *et al.*, 2009). It has been suggested that *in situ*, under physiological conditions, mitochondria might be more like sinks than sources of ROS, if the high antioxidative capacity of mitochondria is taken into account (Brown *et al.*, 2012). The mitochondrial ROS-

detoxifying mechanisms among others include the membrane lipid peroxide removal systems, phospholipid hydroperoxide glutathione peroxidase, manganese superoxide dismutase, cytochrome c, catalase, glutathione, glutathione-S-transferase, glutathione reductase, glutathione peroxidase and peroxiredoxins. This suggests that ROS production is tightly regulated and secured by several lines of antioxidant defense systems in intra- and extra-mitochondrial sites. Today it is commonly accepted that physiological levels of ROS regulate a variety of signaling pathways (e.g. NF- κ B, Nrf-2, STAT3) directly and indirectly (Hamanaka *et al.*, 2010; Niture *et al.*, 2014; Weidinger *et al.*, 2015).

2. Hypoxia and reoxygenation, ischemia and reperfusion, pathological ROS generation

2.1. Hypoxia and ischemia

The impairment of arterial blood flow leads to tissue hypoxia, with lower-than-physiological levels of O₂; the exact value will depend on the energetic needs of the tissue which is already rate-limiting for mitochondrial oxidative phosphorylation (OXPHOS). Taking into account the fact that the K_M value of cytochrome c oxidase is well below 1 μ M for O₂ in uncoupled mitochondria, local hypoxia usually means O₂ levels below 2 μ M. However, the K_M value is between 1-10 μ M in states with a high proton motive force and a small difference in redox potential across the ETS (Petersen *et al.*, 1974; Krab *et al.*, 2011). Furthermore, K_M values for O₂ of the most important O₂-utilizing enzymes strongly depend on the presence of other substrates and gaseous compounds, such as nitric oxide (NO) and carbon monoxide (CO) (Thomas *et al.*, 2001; Cooper *et al.*, 2008).

The dysfunction of mitochondria is central to hypoxic tissue injuries. Several aspects of this condition have already been explored, such as impaired ATP production, an increase in ATP hydrolysis to maintain the mitochondrial membrane potential ($\Delta\psi_m$), dysregulated mitochondrial Ca²⁺ homeostasis and elevated ROS production (Di Lisa *et al.*, 2006). Mitochondrial permeability transition is a process leading to the assembly and opening of a voltage-dependent, high conductance channel in the IMM. In pathological conditions, the sustained opening of mPTP can lead to irreversible mitochondrial damage and apoptosis. ROS and Ca²⁺ are known triggers of pore opening. Upon permeability transition, $\Delta\psi_m$ collapses and ATP synthesis ceases. In a self-protecting endeavor of mitochondria to maintain $\Delta\psi_m$, the cytosolic ATP from glycolysis is hydrolyzed by reversed ATP synthase, further depleting the energy sources of the cell. As a secondary process, the outer mitochondrial membrane (OMM) disrupts and cytochrome c will be released in the cytoplasm, inducing apoptosis (Bernardi *et al.*, 2015).

2.2. Reperfusion injury

The mitochondrial and non-mitochondrial pathophysiology of ischemia-reperfusion (IR) injury is a rather well-explored topic in biomedical research (Granger *et al.*, 2015). In the absence of O₂, which is the final electron acceptor of the respiratory chain, the cascade of redox reactions stops and reducing equivalents accumulate. Generally speaking, the duration of ischemia determines the survival of the affected tissue and the timely restoration of nutritive blood flow is essential to recover organ function. Reperfusion, however, paradoxically worsens the initial damage done by the ischemia itself. This is mainly due to the reintroduction of molecular O₂ to previously ischemic tissue (originally proposed by Hearse (Hearse *et al.*, 1973)). The decisive role of ROS formation, or more precisely, an imbalance between ROS production and detoxification was later demonstrated by Neil Granger (Granger *et al.*, 1981). During hypoxia elevated mitochondrial ROS (mtROS) production occurs, mainly due to partial inhibition of respiratory chain enzymes (Kozlov *et al.*, 1999), but this is much less than ROS produced upon reperfusion (Parks *et al.*, 1986a).

An important contributing factor to elevated mtROS production upon reperfusion is succinate accumulation during ischemia (Chouchani *et al.*, 2014). Previously, mtROS production had been considered a nonspecific process after the re-establishment of O₂. The research team led by Murphy has shown that succinate is rapidly metabolized during the first minutes of reperfusion with concomitant increases of mtROS production by reverse electron transport at CI (Zweier *et al.*, 1987; Murphy *et al.*, 2008). In fact, the application of rotenone, an inhibitor of CI ameliorates the IR-induced mucosal damage in rodents (Ichikawa *et al.*, 2004).

In addition to elevated mtROS production, pathological ROS production in the reperfusion period is fueled by cytosolic or extracellular sources. The most important ROS source in the postischemic tissue is probably xanthine oxidoreductase (XOR) (Harrison, 2002; Khambata *et al.*, 2015). XOR is a complex molybdo-flavoenzyme, originally described by Schardinger in 1902 (Schardinger, 1902; Harrison, 2002). It exists as a homodimer of a single monomer with 150 kDa MW (Andrews *et al.*, 1964). Each monomer has a FAD domain, two iron-sulfur domains and a molybdopterin (Mo) binding site (Khambata *et al.*, 2015). The latter is the site of purine catabolism and NO₂⁻ reduction, which is an important feature of the enzyme.

XOR is ubiquitous in mammals, being generally recognized as the terminal enzyme of purine catabolism, oxidizing hypoxanthine to xanthine and xanthine to uric acid (Harrison, 2002). XOR has two states, which are interconvertible. Xanthine dehydrogenase (XDH) is the

predominant form under physiologic conditions and normoxia. XDH can be converted to xanthine oxidase (XO) by reversible sulfhydryl oxidation or by limited proteolysis. The latter process is irreversible and occurs during tissue hypoxia at high levels of intracellular Ca^{2+} (Granger *et al.*, 1986a). Both forms metabolize hypoxanthine, but XO produces $\text{O}_2^{\bullet-}$, which in turn forms H_2O_2 . During ischemia hypoxanthine accumulates, hence XO contributes substantially to $\text{O}_2^{\bullet-}$ induced tissue injury during reperfusion. In 1986, Granger proposed that XOR is the major source of $\text{O}_2^{\bullet-}$ during IR (Granger *et al.*, 1986b) and although concerns have been raised over the time (Moorhouse *et al.*, 1987), this model of postischemic $\text{O}_2^{\bullet-}$ generation is still widely accepted. Moreover, as has been recently shown, extracellular superoxide produced by XOR bound to GAGs on the endothelial surface of capillaries is a potent activator of PMNs through Toll-like receptor-4 (TLR4) (Lorne *et al.*, 2008).

The highest levels of XOR in rodents and humans have been found in the liver and in the intestine (Parks *et al.*, 1986b). Interestingly, Harrison and colleagues (Harrison, 2002) reported strongly positive anti-XOR antibody staining on the surface of human intestinal epithelial cells, and the enzyme was detected in the TJs of cultured intestinal epithelial cells as well.

3. Hypoxia and inflammation

Hypoxic conditions are inducing inflammation, and inflammatory disease states are frequently accompanied by tissue hypoxia. IR injury is also characterized by an inflammatory response and an important contributing factor to the oxidative tissue damage is the activation of the innate immune system. Extravasated leukocytes produce large quantities of $\text{O}_2^{\bullet-}$ by phagocytic NADPH oxidase and this mechanism can damage host cells and it also reduces tissue oxygenation by consuming O_2 (Hernandez *et al.*, 1987; Yang *et al.*, 2009).

Activated PMNs are not only important contributors to the oxidative stress, but also play a significant role in microcirculatory derangements. The massive expression of adhesion receptors induces leukocyte rolling and firm adherence to postcapillary endothelial cells, which leads to elevated flow resistance, endothelial dysfunction and tissue edema (Menger *et al.*, 1997; Szabo *et al.*, 2008).

An important feature of reperfusion injury is the no-reflow phenomenon. The maintenance of normal red blood cell deformability is a prerequisite of intact microcirculation, since only bullet-shaped erythrocytes can pass capillaries freely. The peroxidation of membrane lipids of RBCs by oxidant species, such as peroxynitrite (ONOO), decreases the fluidity of the affected cellular membranes, contributing to reduced capillary

blood flow (Dobretsov *et al.*, 1977) and in severe cases, capillary stasis can develop (Vollmar *et al.*, 2011). Notably, the direct effects of NO on membrane rigidity were shown earlier (Tsuda *et al.*, 2000) and the adsorption of non-polar molecules in lipid bilayers was demonstrated in model experiments (Gruen *et al.*, 1980).

3.1. Inflammatory mediators

The umbrella term “inflammatory mediator” covers molecules and cells contributing to the initiation, propagation and controlled cessation of the inflammation cascade. Most of them are “soluble” compounds such as cytokines, small protein products of immune cells or other cell types. The most prominent players in acute inflammation are tumor necrosis factor α (TNF- α), interleukins (ILs), lipid mediators (e.g. leukotrienes), and the members of the complement cascade. However, the role of damage-associated molecular patterns (circulating mitochondrial DNA, HMGB-1, hyaluronan fragments, etc.) is increasingly recognized in sterile inflammation (Piccinini *et al.*, 2010; Goodall *et al.*, 2014; Maeda *et al.*, 2014).

ROS (predominantly H₂O₂) are also local inflammatory signal transducer molecules. Gaseous compounds, such as NO, CO and H₂S represent a special class of inflammatory mediators (Liu *et al.*, 2012). From this perspective, the contribution of the inflammatory cells of the monocyte-macrophage axis is of especial importance in gas signal generation and the bulk release of cytokines as well (MacMicking *et al.*, 1997).

Endothelin-1 (ET-1) and interactions with NO

The regulation of the microcirculation is accomplished by a continuous interaction of vasoconstrictor and vasodilator forces. NO and ET-1 are natural counterparts in this complex mechanism. ET-1, a pro-inflammatory compound, is the most potent vasoconstrictor agent that predominantly acts on ET_A receptors, a subtype expressed mainly by vascular smooth muscle cells mediating vasoconstrictor effects. ET_B receptors on endothelial cells mediate mixed effects (Masaki, 2004; Schneider *et al.*, 2007). It has been shown that even low concentrations of NO can effectively inhibit ET-1 release and ET-1-linked functions (Bourque *et al.*, 2011). Shear stress induces NO production by eNOS, and NO, in turn, inhibits ET-1 release from cells (Kuchan *et al.*, 1993). Multiple pathways of inhibition have been shown, but the most relevant under acute conditions is the inhibition of ET_A-phospholipase C mediated intracellular Ca²⁺-release by NO (Goligorsky *et al.*, 1994; Bourque *et al.*, 2011). Reduced NO production during IR can leave ET-1-mediated vasoconstriction unopposed, leading to increased vascular tone and reduced microvascular blood flow, eventually exacerbating the IR injury (Martinez-Revelles *et al.*, 2012). Importantly, increased levels of plasma ET-1 after intestinal IR were previously reported (Kurtel *et al.*, 1999; Guzman-de la

Garza *et al.*, 2009) and ET-1 receptor blockers can alleviate mesenteric IR injury (Oktar *et al.*, 2002).

3.2. Intestinal IR injury

The mucosa forms a physical defense barrier between the intestinal lumen and the body, thus any factor which jeopardizes this defense is of vital concern. Occlusive or non-occlusive mesenteric ischemia with subsequent reperfusion has a common consequence, which results in oxidative injury to the mucosa. The causes are energy imbalance, the activation of inflammatory pathways and uncontrolled interactions between ROS and reactive nitrogen species (RNS) resulting in damage to membrane lipids, proteins and DNA. This can cause an irreversible structural damage to biological membranes impairing their permeability and facilitating the leakage of toxins and bacteria from the intestinal lumen to the blood.

Being a component of the cell wall of Gram-negative bacteria, which are known to occur in abundance in the intestine, bacterial lipopolysaccharide (LPS, endotoxin) is thought to play an important role in the development of systemic inflammation, sepsis and multiple organ dysfunction syndrome originating from the gut after low flow states and IR conditions (Olofsson *et al.*, 1985). The increased mucosal permeability of the intestine permits the translocation of LPS and even bacteria, initiating a TLR4-NF- κ B mediated acute inflammation. The translocation of LPS leads to the priming of PMNs (Deitch *et al.*, 1994; Moore *et al.*, 1994) as well and the production of inflammatory cytokines, which, in turn, exacerbate systemic inflammation.

However, circulating LPS is only the tip of the iceberg, while the majority of LPS is attached to thrombocytes, erythrocytes and TLR4 on immune cells and parenchymal cells inducing variety of pathologic signaling cascades in those cells (Cavaillon *et al.*, 2006).

The inflammatory response provoked by bacterial endotoxin in model systems is dose dependent. The LPS-linked reactions are mediated by TLRs and as part of the cellular stress response, the generation of both $O_2^{\bullet-}$ and NO is one of the major consequences of LPS exposure in various TLR4-expressing cell types, including macrophages, PMNs and endothelial cells (Mittal *et al.*, 2014; Vaure *et al.*, 2014). This, in turn, makes LPS a useful *in vivo* and *in vitro* tool for studying the activation of mammalian immune system and the production of inflammatory mediators. Since the majority of mechanistic studies on basic inflammatory reactions have been conducted in LPS models, a huge body of literature has accumulated and the exploration of novel reactions and pathways is easier under such well standardized circumstances.

3.3. The intestinal atmosphere and the epithelial barrier

The gastrointestinal (GI) mucosal barrier is one of the most important frontiers of the body with an average area of 32 m² in adults (Helander *et al.*, 2014), a large interface between the milieu interieur and the surrounding luminal compounds. Although the composition of the Earth's atmosphere is relatively well known, but only incomplete datasets are available on the gas composition of the mammalian GI tract. In 1971, the survey of Levitt and colleagues reported on the gas composition of healthy human volunteers (Levitt, 1971), very low O₂ levels (O₂: 0.69±0.49%) were found, and the average for N₂ was 64±21%, for CO₂: 14±7%, for H₂: 19±16% and for CH₄ levels ranging up to 20% in the producers. Much more data should be collected on this important subject, if we consider the dependency of biochemical processes on substrate concentrations. Nevertheless, it is well known that the small intestines and the colon have unique oxygenation profiles (Glover *et al.*, 2016) and a number of mechanisms have been developed by intestinal cells to cope with this biochemical environment. The gut mucosa is metabolically very active, but O₂ levels are rather low in the lumen of the bowel, reaching practically zero in the colon (Levitt, 1971; Glover *et al.*, 2016). The O₂ demand is highest and hypoxia tolerance is the lowest at the apical region of the villi (Vollmar *et al.*, 2011), and the countercurrent circulatory exchanger system along the villus axis also contributes to the development of significant intestinal hypoxia, especially in low-flow states (Åhrén *et al.*, 1973; Haglund *et al.*, 1975).

In spite of the relatively poor O₂ supply, the epithelial cells should sustain several processes with high energy demand, including, firstly, the selective absorption of nutrients. To cope with energy needs, epithelial cells are able to oxidize significant amounts of metabolites transported from the apical side, such as short-chain fatty acids and amino-acids (Blachier *et al.*, 2009; Blouin *et al.*, 2011).

Secondly, to maintain the function of the inter-epithelial tight junctions (TJs) and adherent junctions (AJs) of the selectively permeable mucosal barrier (Laukoetter *et al.*, 2006; Citalan-Madrid *et al.*, 2013; Glover *et al.*, 2016), a continuous energy supply is needed. It has been shown *in vitro* that ATP depletion leads to the dissolution of TJs and AJs, but existing junctional complexes can be “reused” if ATP levels are restored within a short period of time. (Tsukamoto *et al.*, 1997; Bush *et al.*, 2000). Similar results were obtained *in vivo*. It has been shown that mesenteric ischemia and reduced mucosal ATP levels lead to TJ dysfunction and elevated permeability (Salzman *et al.*, 1995; Wattanasirichaigoon *et al.*, 1999). Cytosolic creatine kinase (CK) plays an important role that covers intermittent phases of high energy demand, with the help of the cellular phosphocreatine pool, the CK-catalyzed ATP/ADP

conversion acts as a buffer of cytosolic phosphorylation potential (Glover *et al.*, 2013; Kitzenberg *et al.*, 2016).

Interestingly but not surprisingly, “physiological hypoxia” is critical for the expression of innate immune factors and TJ proteins (Kelly *et al.*, 2013; Saeedi *et al.*, 2015). At low O₂ levels HIF-1 α is stabilized, leading to the gene expression of proteins critically involved in the effective barrier function (Colgan *et al.*, 2015).

During inflammation the intestinal tissue can become profoundly hypoxic, or even anoxic despite the non-vascular origin of the disease. Contributing factors are edema, increased O₂ expenditure and decreased O₂ delivery to the inflamed mucosa, aggravated by the NADPH-oxidase dependent O₂ consumption of extravasated PMNs (Campbell *et al.*, 2014).

4. Biologically active gases

While the term “gasotransmitter” does not completely cover “biologically active gases”, there is a significant overlap between the groups. Gasotransmitters are defined by four characteristics (simplicity, availability, volatility and effectiveness, respectively), and six additional criteria were recently listed by Wang (Wang, 2014). Accordingly, gasotransmitters are (1) small molecules of gas; (2) freely permeable to membranes; (3) endogenously generated in mammalian cells with specific substrates and enzymes; (4) have well-defined specific functions at physiologically relevant concentrations; (5) the functions can be mimicked by their exogenously applied counterparts; and (6) have specific cellular and molecular targets. Today the most well-known gasotransmitter compounds are NO, H₂S and CO (Wang, 2014).

4.1. Nitric oxide (NO)

Furchgott and Zawadzki proposed the existence of an endothelium-derived vasorelaxing factor (EDRF) in 1980 (Furchgott *et al.*, 1980), which was later identified as NO (Ignarro *et al.*, 1987; Furchgott, 1988). Apart from vasodilation it has several physiological functions in cellular signaling, the inhibition of platelet aggregation and leukocyte adhesion (Hirst *et al.*, 2011). Under normoxic conditions, NO is mainly produced by various NO synthase isoforms (iNOS, nNOS, eNOS) and the majority of the physiological effects are based on the activation of soluble guanylate cyclase (sGC) by binding to its heme moiety, resulting in increased cGMP levels (Toledo *et al.*, 2012).

NO possesses an unpaired electron and is therefore regarded as a free radical. However, it is relatively stable, reacting only with other free radicals or metals and not

forming dimers spontaneously. Its biological life-time varies (approximately 2s under most conditions), which is extremely short for a signaling molecule but long as a free radical (most of the free radicals have a half-life of milliseconds). Being uncharged, NO readily crosses free biological membranes with passive diffusion in all directions. NO is more soluble in hydrophobic environments (by a factor of 10), which is an important determinant of its action (Thomas, 2015).

4.2. Carbon monoxide (CO)

CO is a small molecule (MW 28), non-polar gas messenger. Unlike NO, CO has no unpaired electron. Both molecules can bind to the same targets, forming complexes with hemoproteins and metalloenzymes. CO is produced endogenously in small quantities by a group of enzymes called heme oxygenases (Wu *et al.*, 2005). The therapeutic application of CO has been demonstrated in various disease models (Ryter *et al.*, 2006) and CO was successfully used in experimental transplantation to minimize IR injury as well (Ozaki *et al.*, 2012).

4.3. Hydrogen sulfide (H₂S)

H₂S has a structural similarity to H₂O. It is membrane-permeable and it can be formed both non-enzymatically and enzymatically in bacteria and mammals as well. Similar to CH₄, sulfate-reducing bacteria in the human intestine are one major source of H₂S. It is one of the most versatile biologically active gases, influencing an exceptionally wide range of physiological and parthophysiological processes (e.g. vascular tonus, myocardial contractility, mitochondrial OXPHOS, IR injury) (Wang, 2012).

4.4. Methane (CH₄)

CH₄ is the most reduced form of carbon, the smallest organic compound and the simplest alkane. It is odorless, colorless and occurs abundantly in the atmosphere. Because of its non-polar properties, the solubility of CH₄ is two orders of magnitude higher in membrane lipids than in the aqueous phase (Miller *et al.*, 1977; Meyer *et al.*, 1980). CH₄ is not toxic *per se*, but a simple asphyxiant, which means that by displacing O₂ in closed spaces hypoxia can develop. It should be stressed that, in physiologically relevant concentrations CH₄ has no toxic effects, unless O₂ delivery is heavily compromised. Under ambient conditions CH₄ is accepted to be not reactive, but in the troposphere CH₄ is oxidized to CO₂ by hydroxyl radicals (formed by the photoreaction of tropospheric ozone and water vapor) (Cantrell *et al.*, 1990; Hurkuck *et al.*, 2012).

4.4.1. Bacterial and non-bacterial origin of CH₄ in mammals

In the GI system of most mammals, especially in ruminants, large amounts of carbohydrates are fermented. CO₂ released during this process is subsequently reduced to CH₄ by the obligate anaerobe prokaryotes, the methanogenic Archaea (Conrad *et al.*, 1999). The rate-limiting last step of CO₂ oxidation is catalyzed by methyl coenzyme M reductase (EC 2.8.4.1) (Wongnate *et al.*, 2015). The gas produced is then excreted with the flatus, leaves through the skin or can be measured in roughly one third of humans in the exhaled breath (Nose *et al.*, 2005; de Lacy Costello *et al.*, 2013).

The non-microbial formation of CH₄ in eukaryotic mitochondria was first demonstrated *in vitro* in 2003 (Ghyczy *et al.*, 2003). It was later shown that during oxido-reductive stress conditions in animal and plant cell cultures and also *in vivo*, measurable quantities of CH₄ are formed (Ghyczy *et al.*, 2008; Wishkerman *et al.*, 2011; Keppler *et al.*, 2016). Possible ways of *in vivo* (bio)chemical reactions that are able to produce CH₄ were reviewed recently (Boros *et al.*, 2015).

4.4.2. Biological effects

Much attention has recently been paid to the bioactivity of CH₄ in eukaryotes and to its possible therapeutic projections. The biological effect of CH₄ in the mammalian organism was first shown by Pimentel and colleagues (Pimentel *et al.*, 2006). They reported that CH₄ slows the small intestinal propulsive motility and augments contractile activity, thereby influencing the enteric nervous/smooth muscle system. Correlations of detectable CH₄ in the exhaled air of humans and peristalsis changes during irritable bowel syndrome have been repeatedly shown as well (Lee *et al.*, 2013; Pozuelo *et al.*, 2015).

An anti-inflammatory potential for CH₄ was first reported by Boros and colleagues in experimental mesenteric IR (Boros *et al.*, 2012). Over the past few years many other papers have been published on the anti-inflammatory effects of CH₄ in various animal models of IR, hypoxia and sterile inflammation (see Table 2). Most publications addressed four aspects of CH₄ activity, namely (1) the modulation of pro-inflammatory cytokine release (TNF- α , IFN- γ , IL-6 and IL-1b); (2) anti-apoptotic effects evidenced by the reduced number of apoptotic cells, normalized caspase-3 and caspase-9 activity, decreased Bax and increased Bcl-2 levels and/or gene expression; (3) the suppressed generation of oxidative stress biomarkers (malondialdehyde, 8-oxo-2'-deoxyguanosine, 4-hydroxynonenal) with concurrent potentiating of endogenous antioxidant systems (superoxide dismutase, catalase, glutathione peroxidase); and (4) improved organ functions. The key publications on the subject are summarized below (Table 1).

Topics	Source	Main findings
IR injury of skin	Song K et al. 2015 BMC Surgery	CH ₄ -rich saline attenuates the IR injury of abdominal skin flaps in rats
Diabetic retinopathy	Wu J et al. 2015 BBRC	Protective effects of CH ₄ -rich saline on retinopathy in a streptozotocin-induced diabetic rat model
Liver IR injury	Ye Z et al. 2015 Shock	CH ₄ attenuates hepatic IR injury in rats through antiapoptotic and antioxidative actions
Liver IR injury	Strifler et al. 2016 PLOS One	Inhaled CH ₄ limits the mitochondrial electron transport chain dysfunction during experimental liver IR injury
Myocardial IR injury	Chen O et al. 2016 FRBM	CH ₄ attenuates myocardial ischemia injury in rats through anti-oxidative, anti-apoptotic and anti-inflammatory actions
Concanavalin A-induced hepatitis	He R et al. 2016 BBRC	CH ₄ -rich saline protects against concanavalin A-induced autoimmune hepatitis in mice through anti-inflammatory and anti-oxidative pathways
Exhaustive exercise	Xin L et al. 2016 PLOS One	CH ₄ -rich saline influences stress induced by one-time exhaustive exercise in rats
Neuroprotection in acute CO poisoning	Fan D et al. 2016 Brain Res	Neuroprotective effects of exogenous CH ₄ in a rat model of acute carbon monoxide poisoning
Retinal IR injury	Liu L et al. 2016 Brain Res	CH ₄ attenuates retinal IR injury via anti-oxidative and anti-apoptotic pathways
LPS-induced immune response	Zhang X et al. 2016 Sci Rep	CH ₄ limits LPS-induced NF-κB/MAPKs signal in macrophages and suppresses immune response in mice by enhancing PI3K/AKT/GSK-3β-mediated IL-10 expression
Acute CO toxicity	Shen M et al. 2016 J Neurol Sci	Neuroprotective effects of CH ₄ -rich saline on experimental acute CO toxicity

Table 1. Recent publications on CH₄ bioactivity

4.5. Relationship between O₂ and other gases

Disturbances in macro- and microcirculation either by primarily vascular origin or during secondary inflammatory disorders can profoundly alter the oxygenation profile of tissues. Many of the endogenous enzymes that utilize O₂ are able to bind other gases as well, but often with lower affinity. The complex interplay of gaseous compounds is of particular importance in hypoxic pathologies, since in the lack of O₂ new reactions can emerge, highlighting those processes which are usually in the background. A well-explored area of such interactions is that of O₂ with NO (van Faassen *et al.* 2009).

NO-linked effects in hypoxia

NO production by the NOS isoforms is an O₂-dependent mechanism. Under hypoxic and anoxic conditions NO can be formed from nitrite (NO₂⁻) and from nitrate (NO₃⁻), with further reduction through NO₂⁻. The main NO₂⁻ reductases in humans are as follows.

- **XOR.** The majority of Mo-containing enzymes are bacterial proteins (Hille 2013). There are four known Mo-containing enzymes in mammals and all four can act as NO_2^- reductase, namely XOR, sulfite oxidase, aldehyde oxidase and mitochondrial amidoxime reductase (Sparacino-Watkins *et al.*, 2014; Maia *et al.*, 2015; Wang *et al.*, 2015). Especially in the intestine and in the liver, XOR has multiple roles during hypoxia and reoxygenation (ROS production by XOR during the reperfusion has been reviewed before in section 2.2), but a crucial catalytic activity of the enzyme is the reduction of NO_2^- to NO. XOR is structurally similar to bacterial $\text{NO}_2^-/\text{NO}_3^-$ reductases (Zhang *et al.*, 1998), and the NO_2^- reductase activity of XO increases with acidosis and in hypoxic conditions (Hassoun *et al.*, 1994; Kayyali *et al.*, 2001).
- **Direct uncatalyzed reduction** occurs at measurable rates in the stomach at low pH (and perhaps in ischemic tissues) (Lundberg *et al.*, 1994).
- **Deoxygenated forms of Hb (and Mb)** are allosterically regulated NO_2^- reductases. Hb produces the NO with the highest rate at the half-loading point of O_2 (about 35 μM) (Huang *et al.*, 2005). The rate of NO_2^- reduction is dependent on the O_2 tension: with higher O_2 concentrations more Hb is saturated and the number of available ferrous hemes for NO decreases.
- **Mitochondrial respiratory chain** complexes III and IV can act as NO_2^- reductases as well (Kozlov *et al.*, 1999). Having a negative charge, NO_2^- cannot diffuse freely into mitochondria, and only a fraction of cytoplasmic NO_2^- is available intramitochondrially in the form of HNO_2 (Samouilov *et al.*, 2007).

The NO formation from NO_2^- under hypoxia is viewed as a salvage mechanism, as under such conditions physiological NO_2^- concentrations are already sufficient to induce vasorelaxation (Dalsgaard *et al.*, 2007).

NO and reperfusion

- NO_2^- -dependent cytoprotection was repeatedly shown in various IR models (Webb *et al.*, 2004; Duranski *et al.*, 2005; Jung *et al.*, 2006; Tripatara *et al.*, 2007). NO is able to inhibit cell surface adhesion molecules both on endothelial cells and leukocytes, thereby preventing extravasation and the activation of PMNs (De Caterina *et al.*, 1995). Furthermore, by the S-nitrosation of inflammatory transcription factors (e.g. activator protein-1 and NF- κB), NO exerts anti-inflammatory effects at the transcriptional level (Pilz *et al.*, 1995).
- **S-nitrosation** (one electron oxidation of a thiol group to form S-NO) of proteins and peptides leads to the formation of S-nitrosothiols (RSNOs) (Stamler *et al.*, 1992; Jia *et al.*,

1996), a cGMP-independent signaling pathway of NO. The RSNOs formed are stable and can act as NO donor molecules. Reactive nitrogen species support the formation of RSNOs. On the other hand, *in vitro* evidence suggests that XO-derived $O_2^{\bullet-}$ is able to decompose RSNO and lead to ONOO formation (Trujillo *et al.*, 1998).

- **Peroxynitrite (ONOO)** formation and the deleterious effects of NO are mostly connected. ONOO is an oxidant species formed in the diffusion-controlled reaction of NO with $O_2^{\bullet-}$ (Beckman *et al.*, 1990). XOR producing NO under hypoxia and $O_2^{\bullet-}$ in the postischemic tissue may be an important source of ONOO (Godber *et al.*, 2000). ONOO and ONOO-derived radicals (e.g. lipid hydroperoxides) can readily oxidize and/or nitrate biomolecules including tyrosine residues, thiols, DNA and unsaturated fatty-acid-containing phospholipids. Irreversible tyrosine nitration at CI is a well known adverse effect of ONOO (Brown *et al.*, 2004), whereas high ONOO concentrations lead to generalized protein and lipid modifications (Gadelha *et al.*, 1997; Szabo *et al.*, 2007).

5. Mitochondrion: the hub of interactions of gaseous compounds

While not much is known about the interactions between various gases in the eukaryotic cells, the importance of the topic is increasingly recognized. Protein complexes of the ETS are partially embedded in the IMM, exposing parts of them to the lipid bilayer and, therefore, to the apolar milieu. This is an important point to consider, since some of the most abundant gas mediators (e.g. NO) are far more soluble in lipids than in the aqueous phase. The mitochondrial ETS, particularly cytochrome c oxidase (CIV), is the target of the whole machinery of aerobic organisms which seek to transport molecular O_2 . This enzyme catalyzes the reduction of O_2 by ferricytochrome to H_2O ; hence it is quite predictable that other gases can interact with it as well. Indeed, CO, NO, H_2S and HCN all inhibit CIV (Cooper *et al.*, 2008). More importantly, it has been shown that CIV inhibition by sodium azide leads to endogenous CH_4 generation in plant cells and mammals (Wishkerman *et al.*, 2011; Tuboly *et al.*, 2013b). Furthermore, it has been shown that CH_4 inhalation improved mitochondrial OXPHOS in a partial hepatic IR model, at least indirectly (Strifler *et al.*, 2016).

If O_2 levels are in the physiologic range, the affinity of NO for sGC is about 50 times higher than that for the cytochrome c oxidase (data from brain tissue at O_2 levels of 30 μM (Brown *et al.*, 1994)). However, with lower O_2 concentrations this apparent gap decreases and NO inhibits CIV in competition with O_2 .

Usually, the reduction of molecular O_2 at CIV is not rate-limiting in mitochondrial respiration, but the partial inhibition of CIV by one gas will make it more rate limiting for

respiration, and in consequence more sensitive to other gases (Cooper *et al.*, 2008). Quite significantly, hypoxia is an inhibitor of CIV as well. Therefore, under hypoxic conditions, the inhibitory effects exerted by other gases can be amplified. The K_M of CIV is about 0.5 μM of O_2 , well below the O_2 levels in the body; hence CIV is not rate-limited. (We should mention here that because of the physiologically low tissue O_2 levels in the intestine, the mitochondria there might be prone to interactions with other gases). In contrast, if 60 nM of NO is present, K_M of O_2 rises to 30 μM , well above O_2 levels in the intestine (Brown *et al.*, 1994). In most cases, the reciprocal control of O_2 and NO relieves inhibition. Under profound ischemia, however, vasodilation is ineffective and compensatory mechanisms fail to replenish O_2 to mitochondria, resulting in gradually increasing NO concentrations.

Under physiological conditions, when an excess capacity of OXPHOS is present, the partial inhibition of CIV by NO may operate in parallel with ATP production matching the needs of the cell, while providing a “regulated” ROS formation for signal transduction. The term “nitroxia” was suggested by Shiva and coworkers for a dysregulated state, having some of the characteristics of hypoxia, when excess NO substantially inhibits CIV, leading to ATP depletion despite sufficient O_2 levels. In this case, enhanced ROS formation can be measured (Shiva *et al.*, 2005).

It is recognized that gaseous compounds are able to interact with other components of the mitochondrial electron transport machinery as well. The inhibition of CI by S-nitrosation is thought to be beneficial, protecting downstream complexes from pathologically increased ROS production under reperfusion (Brown *et al.*, 2004; Shiva *et al.*, 2009). Another mitochondrial target of gases is cytochrome c; CO, NO and H_2S can also interact with this hemoprotein with different degrees of affinity. Notably, interactions of cytochrome c with cardiolipin, a component of the IMM, dramatically enhance the ligand interactions of cytochrome c, so gas interactions are far more likely (Kagan *et al.*, 2005; Vlasova *et al.*, 2006).

Interestingly, p66Shc, an alternatively spliced transcript from the SHC1 gene, can produce ROS acting as an oxidoreductase and can transfer electrons from mitochondrial cytochrome c to O_2 (Giorgio *et al.*, 2005). The role of mitochondrial p66Shc was postulated in the ROS-induced ROS release (Zorov *et al.*, 2000; Galimov *et al.*, 2014), ultimately leading to the induction of mitochondrial permeability transition. How the binding of gas molecules to cytochrome c influences this process has not yet been explored.

Apart from reversible interactions of gases with mitochondrial enzymes, ONOO can inhibit CI, CII, CIV, ATP synthase, aconitase, Mn-SOD, creatine kinase, and probably many

other mitochondrial proteins in an uncontrolled manner (Cassina *et al.*, 1996; Gadelha *et al.*, 1997). This obviously harmful consequence of pathologically elevated mtROS and NO levels can only be prevented by the timely inhibition of abnormal ROS and NO production.

The interplay of CH₄ with NO in mammals has not yet been investigated systematically. On the one hand, it was demonstrated that normoxic CH₄ ventilation decreases tyrosine nitrosylation after an IR injury (Boros *et al.*, 2012), a process which involves NO. On the other hand, it has been shown that the inhibition of CIV, an important target of NO under hypoxia, leads to CH₄ generation (Wishkerman *et al.*, 2011; Tuboly *et al.*, 2013b). It should be added that the heme a₃/CuB O₂ reduction site of CIV is hydrophobic (Wikstrom, 2004), increasing the probability of interaction with gases that are more soluble in lipids than in the water phase. Moreover, some of the effects exerted by CH₄ in model systems of inflammation can be explained by the indirect modulation of functions of NO. It is likely that the two gases are able to modulate the effect of each other at membrane interfaces, where their concentration is at its peak. The reaction of NO with O₂ in hydrophobic membranes thought to be faster than in the aqueous phase. Based on model calculations, even with only 3% of lipids in the tissue volume, the majority of reactions of NO with O₂ will occur within membranes (Liu *et al.*, 1998). Taking into account the solubility of CH₄, comparable changes can be estimated.

II. Aims

1. Our first general aim was to explore the consequences and effects of normoxic CH₄ administration in a small animal model of IR-induced inflammation; the following specific goals have been addressed.

- The first specific goal was to characterize the changes in epithelial and endothelial permeabilities induced by IR challenge in the rat small intestine. With this aim, we monitored the early and later biochemical consequences, the structural and hemodynamic changes in the intestinal mucosa;
- The second goal was to detect specific biochemical markers of both oxidative and nitrosative stress to investigate the possible mechanism of action of exogenous CH₄-based treatments;
- The third goal is based on our hypothesis that CH₄ has direct influence on cell membrane rigidity. Therefore an additional goal was to investigate the influence of CH₄ on changes in erythrocyte deformability provoked by oxidative stress *in vitro*.

2. Our second general aim was to better understand the mechanisms of CH₄ action in models of tissue hypoxia with elevated ROS and RNS levels, with special emphasis on the possible relationship of CH₄ with other biological gases, such as NO; the following specific goals have been addressed in this part of the study.

- The first goal was to establish a bacterial endotoxin-based *in vivo* experimental model of inflammation to follow the changes in locally generated NO levels;
- The second goal was to develop a specific and sensitive method to detect NO generation in the model established in the first goal;
- The third goal was to examine the potential interplay of CH₄ with NO directly investigating *in vivo* and *ex vivo* effects of CH₄ administrations on the rates of NO generation in biological samples by means of electron paramagnetic resonance (EPR) spectroscopy technique and on-line chemiluminescence analyses.

III. Materials and methods

1. An *in vivo* study to investigate the effects of CH₄ treatments in mesenteric IR

1.1. Animals

The experiments were performed on 76 male Sprague-Dawley rats (280-320 g bw) in accordance with National Institutes of Health guidelines on the handling and care of experimental animals and EU Directive 2010/63 for the protection of animals used for scientific purposes, and it was approved by the National Scientific Ethical Committee on Animal Experimentation (National Competent Authority), with the license number V/148/2013. The animals were housed in plastic cages under 12/12 hr day-night cycle, standard air temperature and humidity conditions. All chemicals were obtained from Sigma-Aldrich Inc. (Budapest, Hungary) unless otherwise stated.

1.2. Surgical procedure and study design

Rats fed on a normal laboratory diet with tap water *ad libitum*, were randomly allocated into one or other of the experimental groups. After overnight fasting, the animals were anaesthetized with sodium pentobarbital (50 mg/kg bw ip) and placed in a supine position on a heating pad. The trachea was dissected free and cannulated with a silicone tube, and the right jugular vein was cannulated with PE50 tubing for fluid administration and infusion of Ringer's lactate (10 mL/kg/h) during the experiments.

The experiments were performed in two series (see Figure 1). In study 1 (the “early reperfusion” study), the animals were killed 60 min after the re-establishment of the mesenteric blood flow, while in the second set (the “late reperfusion” study), the reperfusion period and the corresponding control phase in the sham-operated animals lasted for 180 min.

After a midline laparotomy, the superior mesenteric artery (SMA) was dissected free. Group 1 (n=6) served as a sham-operated control, while in Group 2 (IR, n=6) the SMA was occluded using an atraumatic vascular clip for 45 min. In CH₄-treated Group 3 (IR+CH₄, n=6) an artificial gas mixture containing 2.2% CH₄, 21% O₂ and 76.8% N₂ (Linde Gas, Budapest, Hungary) was administered for 5 min before the end of the 45-min ischemia and for 10 min at the beginning of the reperfusion (Figure 1). In study 2 the protocol followed was identical, but the length of the observation and the reperfusion phase was different.

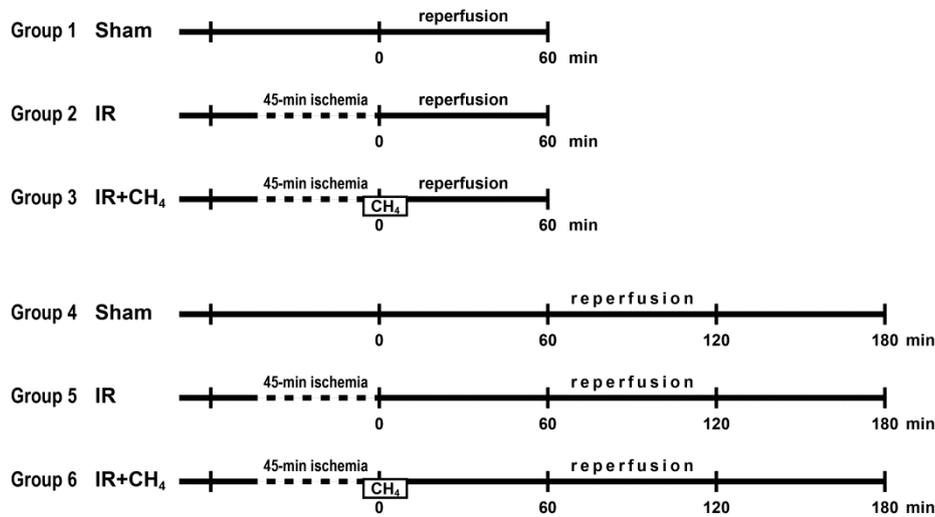


Figure 1. Experimental scheme

1.3. Mucosal permeability measurements

The epithelial permeability was determined with the 4 kDa fluorescein isothiocyanate-dextran (FD4) method as described previously (Cuzzocrea *et al.*, 1997). In short, a 5 cm long segment of the terminal ileum supplied by 3 blood vessel arcades was isolated at a distance of 10 cm from the ileocecal valve. Silicone cannulas were placed and fixed into the oral and aboral ends of the segment and the lumen was gently flushed with 5 mL of 37°C saline and 5 mL air, then the distal end was closed. Before performing measurements, the renal pedicles were ligated. Exactly at the moment of reperfusion (the “early reperfusion” study) or 120 min after (the “late reperfusion” study), the lumen was filled with 0.5 mL of warmed (37°C) FD4 solution (25 mg/mL). Blood samples (0.3 mL) were then taken 5, 10, 20, 30 and 40 min later for plasma fluorescein concentration measurements with a fluorescence spectrophotometer (F-2000 Hitachi, Japan, Ex: 492 nm; Em: 515 nm). The blood samples were stored on ice in the dark and centrifuged at 100g for 10 min. At the end of the experiment, the bowel segment was removed and weighed. The epithelial permeability index was expressed as a percentage of FD4 measured in the plasma and it was calculated using the formula [arterial FD4 concentration (ng/mL) / luminal FD4 concentration (ng/mL)] *100.

The vascular permeability index was determined using the azo dye Evans blue method as described previously (Szentpali *et al.*, 2001). Stated briefly, 30 min before the end of the experiments 20 mg/mL/kg of Evans blue was given in iv bolus and 30 min later a blood sample was taken from the caval vein, together with a whole-thickness tissue sample from the ileum. The biopsy specimen was placed in 5 mL of formamide and homogenized for 1 min in a glass Potter homogenizer. The homogenate was incubated at room temperature for 20 h and

then centrifuged at 2500g for 30 min. The absorbance of the supernatant was determined at 650 nm against a formamide blank with a UV-1601 spectrophotometer (Shimadzu, Japan). The concentration of Evans blue was determined from a standard curve and it was normalized to the protein content of the samples (Lowry *et al.*, 1951). Similarly, blood samples were centrifuged at 600g at 4°C for 10 min and the absorbance of the 100-fold-diluted plasma was measured. The vascular permeability index was defined as the ratio of the tissue and plasma concentrations of Evans blue, i.e. [tissue Evans blue concentration / plasma Evans blue concentration] *100.

1.4. Morphological analysis

Histopathological analysis

Full-thickness ileum biopsies were taken after 30 min of reperfusion in each group. The tissue was rinsed in ice-cold saline, fixed in 10 % buffered formalin, embedded in paraffin, cut into 4 µm-thick sections, and stained with hematoxylin and eosin. Photomicrographs were recorded with a 40x objective connected to a digital camera.

***In vivo* histology**

The extent of superficial epithelial damage of the terminal ileum was evaluated by means of fluorescence confocal laser scanning endomicroscopy (CLSEM, Five1 Optiscan Pty. Ltd, Melbourne, Victoria, Australia) developed for *in vivo* histology. The mucosal surface of the terminal ileum 10 cm proximal to the cecum was exposed surgically and laid flat for examination. The injury of the mucosal architecture was examined after topical application of the fluorescent dye acriflavin. The surplus dye was washed off with saline, then the objective of the device was placed onto the mucosal surface of the ileum, and confocal imaging was performed 2 min after dye administration (1 scan per image, 1024 x 1024 pixels and 475 x 475 mm per image). Non-overlapping fields were processed and evaluated by a modified semiquantitative scoring system (Érces *et al.*, 2016). The grading was performed using five criteria: I. denudation of villi (0 = no denudation, 1 = at least one denuded area per field of view, 2 = more than one area without any recognizable villus structure per field of view; II. edema (0 = no edema, 1 = moderate epithelial swelling, 2 = severe edema); III. shedding (0 = a normal, clearly, well-defined villus structure without shedding cells, 1 = some shedding cells; fewer than 30 cells per field of view, 2 = shedding cells, more than 30 cells per field of view, 3 = severe debris); IV. epithelial gap (0 = no gap, 1 = less than 5 gaps per villi, 2 = more than 5 gaps per villi); V. longitudinal fissure on villi (0 = no fissure, 1 = presence of fissure). A blinded analysis of the same images was performed twice off-line.

1.5. Hemodynamics

The SMA flow signals (T206 Animal Research Flowmeter; Transonic Systems Inc, Ithaca, NY, USA) were measured continuously and recorded with a computerized data acquisition system (Experimetria Ltd, Budapest, Hungary).

An intravital orthogonal polarization spectral imaging technique (Cytoscan A/R, Cytometrics, Philadelphia, PA, USA) was used for the visualization of the serosal microcirculation of the ileum. This technique uses reflected polarized light at the wavelength of the isosbestic point of oxyhemoglobin and deoxyhemoglobin (548 nm). Because polarization is preserved in reflection, only photons scattered from a depth of 200 μm contribute to image formation. A 10x objective was placed onto the serosal surface of the ileum, and microscopic images were recorded with an S-VHS video recorder (Panasonic AG-TL 700, Matsushita Electric Ind. Co. Ltd, Osaka, Japan). The quantitative assessment of the microcirculatory parameters was performed off-line by a frame-to frame blinded analysis of the videotaped images. Changes in red blood cell (RBC) velocity (mm/s) in the postcapillary venules were determined in 3 separate fields by means of a computer-assisted image analysis system (IVM Pictron, Budapest, Hungary). All microcirculatory evaluations were performed by the same investigator.

1.6. Biochemical measurements

Ileum biopsies kept on ice were homogenized in a phosphate buffer (pH 7.4) containing Tris-HCl (50 mM, Reanal, Budapest, Hungary), EDTA (0.1 mM), dithiothreitol (0.5 mM), phenylmethylsulfonyl fluoride (1 mM), soybean trypsin inhibitor (10 $\mu\text{g}/\text{mL}$) and leupeptin (10 $\mu\text{g}/\text{mL}$). The homogenate was centrifuged at 4°C for 20 min at 24 000g (Amicon Centricon-100, Millipore Corporation, Bedford, Massachusetts, USA). Tissue nitrotyrosine (NTyr) concentration was determined in the supernatant, while myeloperoxidase (MPO) activity was measured in the pellet of the homogenate.

Tissue myeloperoxidase (MPO) activity

The activity of MPO as a marker of tissue leukocyte infiltration was measured in the pellet of the homogenate using the modified method of Kübler et al. (Kuebler *et al.*, 1996). Stated briefly, the pellet was resuspended in K_3PO_4 buffer (0.05 M; pH 6.0) containing 0.5 % hexa-1,6-bis-decyltriethylammonium bromide. After three repeated freeze-thaw procedures, the material was centrifuged at 24 000g at 4°C for 20 min and the supernatant was used for MPO determination. Afterwards, 0.15 mL of 3,3',5,5'-tetramethylbenzidine (dissolved in DMSO; 1.6 mM) and 0.75 mL of hydrogen peroxide (dissolved in K_3PO_4 buffer; 0.6 mM) were added to 0.1 mL of the sample. The reaction led to the hydrogen peroxide-dependent

oxidation of tetramethylbenzidine, which was detected spectrophotometrically at 450 nm (UV-1601 spectrophotometer; Shimadzu, Kyoto, Japan). MPO activities were measured at 37°C; the reaction was stopped after 5 min by the addition of 0.2 mL of H₂SO₄ (2 M) and the resulting data were normalized to the protein content.

Plasma ET-1 levels

Blood samples (0.5 mL) were taken from the inferior caval vein and placed into chilled polypropylene tubes containing EDTA (1 mg/mL) at the end of the “late reperfusion” experiments (180 min after reperfusion), centrifuged at 1000g at 4°C for 30 min and then stored at -70°C until assay. The plasma ET-1 concentration was determined in duplicates by means of a commercially available enzyme-linked immunosorbent assay (ELISA) kit (Biochemica Hungaria Ltd., Budapest, Hungary) and expressed as fmol/mL.

Intestinal O₂^{•-} production

The rate of O₂^{•-} production in freshly minced intestinal biopsy samples was assessed by using the lucigenin-enhanced chemiluminescence assay described by Ferdinandy et al. (Ferdinandy *et al.*, 2000) In short, approx. 10 mg of intestinal tissue was placed in 1 mL of Dulbecco’s solution (pH 7.4) containing 5 μM of lucigenin. The manipulations were performed without external light 2 min after dark adaptation. Chemiluminescence was measured at room temperature in a liquid scintillation counter by using a single active photomultiplier positioned in out-of-coincidence mode, in the presence or absence of the O₂^{•-} scavenger nitroblue tetrazolium (NBT; 20 μl). NBT-inhibited chemiluminescence was interpreted as an indicator of intestinal O₂^{•-} generation.

Tissue nitrotyrosine level

Free NTyr, as a marker of ONOO generation, was measured by an enzyme-linked immunosorbent assay (Cayman Chemical; Ann Arbor, MI). Small intestinal tissue samples were homogenized and centrifuged at 15000g. The supernatants were collected and incubated overnight with anti-NTyr rabbit IgG and a NTyr acetylcholinesterase tracer in precoated (mouse anti-rabbit IgG) microplates, which were developed using Ellman’s reagent. The NTyr content was normalized to the protein content of the small intestinal homogenate and expressed in ng/mg.

1.7. CH₄ concentration measurement

The CH₄ concentration was measured by means of photoacoustic spectroscopy (PAS) (Tuboly *et al.*, 2013a). PAS is a special type of spectroscopy that measures optical absorption indirectly via the conversion of absorbed light energy into acoustic waves. The amplitude of the generated sound is directly proportional to the concentration of the absorbing gas

component. The light source of the system is a near-infrared diode laser that emits around the CH₄ absorption line at 1650.9 nm with an output power of 15 mW (NTT Electronics, Tokyo, Japan). The cross-sensitivity for common components of breath and ambient air were repeatedly examined, and no measurable instrument response was found for significant concentrations of CO₂ or H₂O vapour. The narrow line width of the diode laser provides high selectivity; the absorbance of CH₄ is several orders of magnitude greater than that of H₂O, CO₂ or CO at 1.65 μm, the wavelength we used. The CH₄ values were corrected for background levels and expressed in parts-per-million (ppm).

The device was previously calibrated with various gas mixtures prepared by the dilution of 1 vol % of CH₄ in synthetic air, and it has a dynamic range of 4 orders of magnitude; the minimum online detectable concentration of the sensor was found to be 0.25 ppm (3σ), with an integration time of 12 s (Tuboly et al. 2013).

In a separate set of experiments, tissue CH₄ concentration was measured multiple times in anesthetized rats after the inhalation of room air or artificial air containing 2.2% exogenous CH₄. Immediately after 15 min of inhalation, a 200 mg ileum sample was taken, excess fluid was immediately wiped off and the tissue specimen was placed in a glass vial with 20 mL headspace volume and closed so as to be air tight. Parallel to taking ileum biopsies, 1 mL of blood was taken from the common carotid artery of the same animals through a silicone cannula and was subsequently transferred to identical glass vials. The outlet of the vials was connected to the pump of the spectroscope and headspace gas was pumped into the chamber of the device with a rate of 10 mL/min.

2. *In vitro* and *ex vivo* studies for the detection of NO generation

2.1. Animals

The experiments were performed on male Sprague-Dawley rats (n=45; 250–300/390–540 g bw; Animal Research Laboratories, Himberg, Austria/Charles River, Germany) which were kept under controlled standard animal housing conditions. The animals had free access to standard laboratory rodent food and water. They were kept for seven days prior to usage in experiments for accommodation. All interventions were conducted in compliance with the National Institutes of Health's "Guide for the Care and Use of Laboratory Animals" with approval from the Animal Protocol Review Board of the city government of Vienna, Austria. All chemicals were obtained from Sigma-Aldrich (Vienna, Austria), unless otherwise stated.

2.2. LPS treatment and study design

LPS from *Escherichia coli* serotype 026:B6 (activity ≥500,000 EU/mg) was used for the dose dependence experiments. Six groups of rats were injected with various doses (0.2, 1.3,

2.5, 4.7, 6.3, and 8.5 mg/kg body weight) of LPS dissolved in saline (Fresenius Kabi, Bad Homburg vor der Höhe, Germany). Control animals were injected with saline only. Samples were then collected 16 h after the LPS injection. In a separate set of experiments, animals were divided into five groups. All rats were injected with the same dose of LPS (*Escherichia coli* serotype 026:B6, 8 mg/kg body weight, activity $\geq 10,000$ EU/mg) dissolved in saline, and samples were taken 2, 4, 8, and 16 h after LPS treatment. Control animals were injected with saline only. The LPS solution was vortexed for 1 min and sonicated for 30 s before application, and subsequently injected in the penis vein under isoflurane anesthesia in a volume ranging from 0.5 to 0.75 mL.

2.3. EPR spectroscopy

EPR spectra were recorded at liquid nitrogen temperature (-196°C) with a Magnetech MiniScope MS 200 EPR spectrometer (Magnetech, Berlin, Germany) in a quartz finger-type Dewar flask filled with liquid nitrogen. The EPR spectra were then recorded at short- and long-scale ranges. The general settings for a short range were as follows: modulation frequency, 100 kHz; microwave frequency, 9.425 GHz; microwave power, 8.3 mW; modulation amplitude, 5 G; and gain, 200. NO–Hb complexes were recorded at 3300 ± 200 G. The general settings for the long range were as follows: modulation frequency, 100 kHz; microwave frequency, 9.429 GHz; microwave power, 30 mW; and modulation amplitude, 6 G. Liver spectra were recorded at 3200 ± 500 G. The spectra were quantified by the determination of magnitudes of different components of spectra and by double integration of EPR spectra.

2.4. NO trapping

Iron-bound NO in the intestinal tissue of rats with CH_4 treatment was measured using the method described earlier (Kozlov *et al.*, 2001b). Stated briefly, rats were injected subcutaneously with sodium diethyldithiocarbamate (Na-DETC; 500 mg/kg) and FeSO_4 (50 mg/kg, dissolved in 0.18 mM sodium citrate) separately with 5 minutes between the injections, 20 minutes before SMA occlusion. DETC forms a water-insoluble DETC-Fe complex with iron in the tissue compartment, which is able to bind and stabilize NO as a DETC-Fe-NO complex, and this permits the measurement of NO in the natural compartment where it was formed.

3. An *in vitro* microrheological study

3.1. Study design, induction of oxidative stress and CH₄ treatment

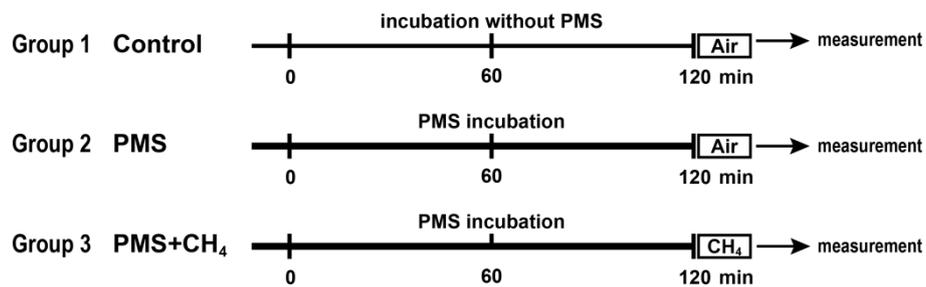


Figure 2. Experimental protocol of the microhemorheological study

Venous blood from healthy male volunteers was collected in lithium heparin coated tubes. Blood samples were placed into three groups, and incubated for 120 minutes at 37°C on a rollerbed before RBC aggregation and deformability measurements. A non-treated sample served as the negative control. Oxidative stress was induced with the addition of phenazine methosulfate (PMS, dissolved in phosphate buffered saline, final concentration 200 µM) and an incubation for 120 minutes (Rabai *et al.*, 2010). Before the study, a dose-response experiment was performed with PMS concentrations between 0 and 400 µM, to determine the effective concentration. In the third, CH₄-treated group, the headspace of the sample was continuously perfused with a gas mixture containing 2.2 % CH₄ in normoxic air (Messer Hungarogáz, Budapest, Hungary) for 10 minutes after at the end of the PMS incubation protocol.

RBC deformability and the aggregation of samples were determined by means of ektacytometry and light-transmission aggregometry immediately after the incubation period. Samples were taken from the same vials for both measurements.

3.2. RBC deformability

RBC deformability in response to shear forces was determined via a LORCA ektacytometer (Laser-assisted Optical Rotational Cell Analyzer; R&R Mechatronics, Hoorn, Netherlands).

Immediately after the treatment protocol, 20 µL of blood was suspended in 4 mL of a polyvinylpyrrolidon solution, with a viscosity of 29.8 mPas, and injected into the cylinder of the ektacytometer. During the measurement, the temperature was kept at 37°C.

Deformation is characterized by the elongation index (EI) calculated from the diffraction pattern of laser light on elongated RBCs. The light was captured and analyzed by a video camera and a computer system that calculated an EI as the (length - width) / (length + width)

of the pattern for 9 different shear stress values ranging from 0.5 Pa to 50 Pa (Hardeman *et al.*, 1994).

3.3. RBC aggregation

The aggregation characteristics of erythrocytes were assessed by using a light transmission method described earlier (Kiesewetter *et al.*, 1982). Immediately after the treatment protocol, 30 μL of blood was transferred to a Myrenne MA-1 Aggregometer (Myrenne GmbH, Roetgen, Germany).

The blood sample was first sheared at 600 s^{-1} to disperse all pre-existing aggregates, and then the shear rate decreased rapidly to low shear rates. The degree of aggregation was characterized by the aggregation index, which is calculated using the surface area below the light intensity curve in a 10 second period. Measurements were then performed at ambient room temperature.

4. *In vitro* analysis of anoxic NO release

4.1. Sample preparation

Male Sprague-Dawley rats (350-450 g) were euthanized by decapitation under isoflurane anesthesia. The liver was excised and placed in ice-cold Ringer solution (B. Braun Melsungen AG, Germany), cut into small pieces, snap-frozen in liquid nitrogen and stored at -80°C until assay. After thawing, the tissue was weighted and supplemented in a 1:10 ratio with an incubation buffer containing 106 mM KCl, 5 mM KH_2PO_4 , 20 mM Tris-HCl and 0.5 mM EDTA. The pH of the solution was set to 6.0 to mimic ischemic tissue with severe acidosis. Afterwards, the liver sample was homogenized with an overhead stirrer (RW16, IKA Werke, Germany).

4.2. NO measurement protocol

NO release from liver homogenate was measured on-line by means of chemiluminescence (Sievers 280i, General Electric, USA) under anoxic conditions, pH 6.0 and 37°C . 0.5 mL of diluted liver homogenate supplemented with 19 mL of incubation buffer and 38 μL antifoam was added into the glass chamber of the device. Then the samples were equilibrated with nitrogen or nitrogen containing 2.2% CH_4 (Messer Hungarogáz, Budapest, Hungary) under continuous gas flow for 10 minutes. Afterwards, 4.4 mM NaNO_2 was added and the NO released from the homogenate was measured in real-time for a 15 minute period.

5. Statistical analysis

For a statistical evaluation of the data, GraphPad Prism 5.01 for Windows (GraphPad Software, La Jolla, California, USA) was used. The statistical analysis was performed by a two-way analysis of variance of repeated measures followed by Bonferroni post-hoc test in

normally distributed data and Kruskal-Wallis one-way analysis of variance on ranks combined with Dunn's method for pairwise multiple comparisons in groups showing a non-Gaussian distribution. Pearson's test was used to assess the statistical significance ($p < 0.0001$) of correlations. Taking into account the fact that parts of the *in vivo* data were not normally distributed (not Gaussian), we displayed the data as box plots, where possible. Median values and 75th and 25th percentiles are given and in each case, p values < 0.05 were considered significant.

IV. Results

1. The kinetics of CH₄ transport

The CH₄ concentration in the baseline samples of non-CH₄-producer animals remained below the background levels in the arterial blood and in the ileum tissue prior to the beginning of the experiment (figures 3A and 3B, “Baseline”). After 5 minutes of normoxic CH₄ inhalation with a flow rate of 300 mL/min, at the end of the SMA ischemia substantially increased CH₄ concentrations were detected in the systemic arterial blood and there was a slight increase in the ileum as well (figures 3A and 3B, “Isch 45’ – CH₄ 5’”). 10 minutes into the reperfusion phase, at the end of the 15 min inhalation of normoxic CH₄-air mixture, CH₄ concentration in the ileal tissue also increased (Figure 3B, “Rep 10’ – CH₄ 15’”). In samples taken at the 60th minute of the reperfusion, 50 minutes after the end of CH₄ treatment, no significant amounts of CH₄ were found in intestinal or blood samples (figures 3A and 3B, “Rep 60’”).

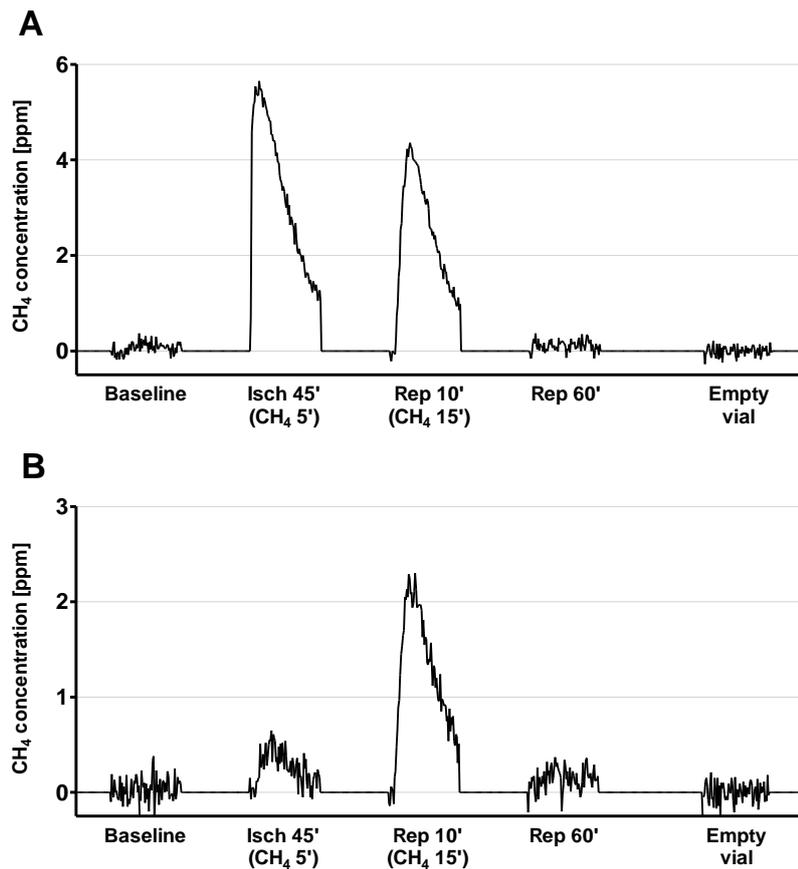


Figure 3. CH₄ transport to the intestine. CH₄ concentrations in blood (A) and ileum (B) samples taken during the CH₄-air inhalation protocol in a pilot study. An original recording of photoacoustic signals as a function of time. At the indicated time point during the experiments, a tissue sample of the ileum, weighing approximately 200 mg was taken and placed immediately in a glass vial. Simultaneously, blood samples (1 mL each) were taken from the common carotid artery into glass vials. Gas samples from the headspace (20 mL) were thereafter transferred to the photoacoustic

detection system and each tissue specimen was measured for 10 minutes. Values are expressed as parts-per-million (ppm) and are corrected for background CH₄ levels. Baseline – prior to CH₄ inhalation and ischemia. Isch 45' (CH₄ 5') – end of mesenteric ischemia, 5 minutes of normoxic CH₄ inhalation. Rep 10' (CH₄ 15') – 10th minute of reperfusion, end of the 15-min normoxic CH₄ inhalation. Rep 60' – 60th minute of reperfusion. Empty vial - CH₄ concentration without biological sample, used for background-correction.

2. The small intestinal permeability

Small intestinal epithelial barrier function

The epithelial permeability (EP) index was determined to assess the barrier function of the intestinal mucosa during the re-establishment of the blood flow to the previously ischemic tissues (**Figure 4A**). The EP did not change in sham-operated, control animals, while the plasma levels of FD4 rose steeply in the IR group, which indicates a rapid loss of the epithelial barrier function. Normoxic CH₄ treatment resulted in significantly lower EP levels, implying there were preserved interepithelial junctions.

Later in the reperfusion (**Figure 4B, below**), the EP index in non-treated animals decreased, suggesting an improved barrier function as compared to that in the early phase; while in the CH₄-treated group, the change was similar to those observed in the early phase of reperfusion. The EP index remained at the baseline level in the control animals.

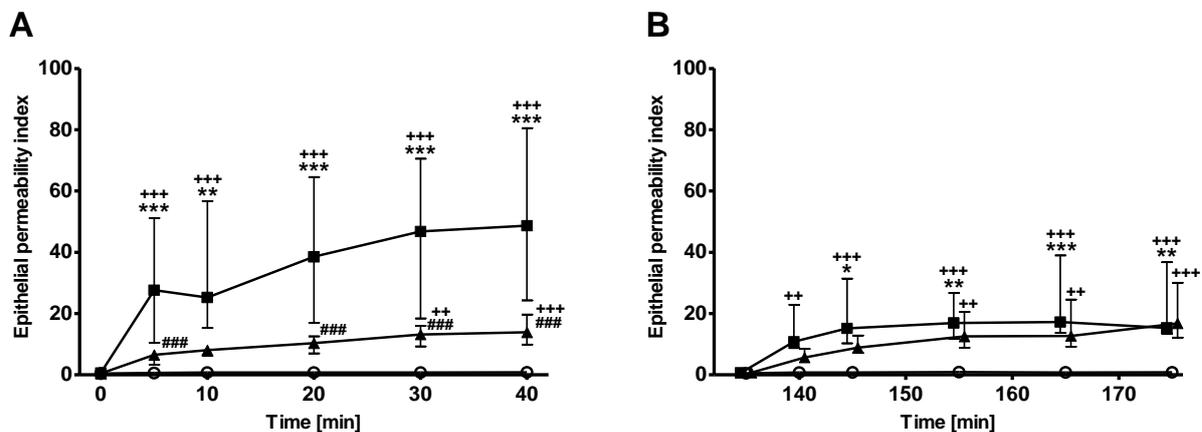


Figure 4. Epithelial permeability. Changes in epithelial permeability (EP) in the sham-operated (empty circles), mesenteric IR (black squares), and CH₄-treated IR (black triangles) groups during early (A) and late phases (B) of reperfusion. The EP was assessed in each period and the permeability index was calculated as described earlier (see the Methods section for a description of the experiments). The data are expressed as median, 25th and 75th percentiles. Here, ** means p<0.01 between groups vs. sham-operated group; ### means p<0.001 between CH₄-treated and IR groups, while + means p<0.05 and +++ means p<0.001 compared to baseline values within groups.

Small intestinal microvascular barrier function

The vascular permeability (VP) assessed by Evans blue extravasation increased in the early reperfusion relative to sham-operated control animals, but statistically significant differences between the experimental groups were not detected either in the early (**Figure 5A**) or the later (**Figure 5B**) phases of reperfusion.

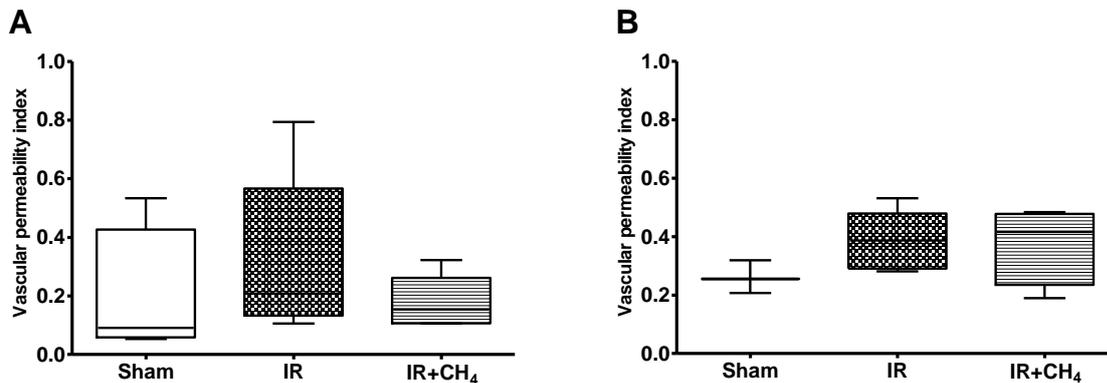


Figure 5. Vascular permeability. Changes in vascular permeability (VP) in the sham-operated (Sham, empty boxes), mesenteric IR (IR, checkered boxes), and CH₄-treated IR (IR+CH₄, hatched boxes) groups during early (A) and late phases (B) of reperfusion. The VP index was calculated using a spectrophotometric evaluation of Evans Blue extravasation (see Materials and methods). The results are expressed as median, 25th and 75th percentiles.

Micro- and macrocirculatory changes

Prior to the induction of SMA occlusion, the RBC velocity in the microvessels of the ileal serosa was similar in all groups (see **Figure 6**). In the 15th min of reperfusion, the intestinal microcirculation of the IR group was significantly impaired. In the IR+CH₄-treated groups the RBC velocity did not differ from the sham-operated groups, implying improved microcirculation. By the 120 min of the reperfusion, no differences could be seen among the groups.

The SMA blood flow was assessed continuously during the experiments (**Figure 7**). In the IR and IR+CH₄ groups, the complete cessation of blood flow was followed by different reactions during reperfusion. The SMA flow in IR group remained significantly low as compared to baseline levels; while in CH₄-treated animals, the SMA flow was significantly higher compared to that in the IR group.

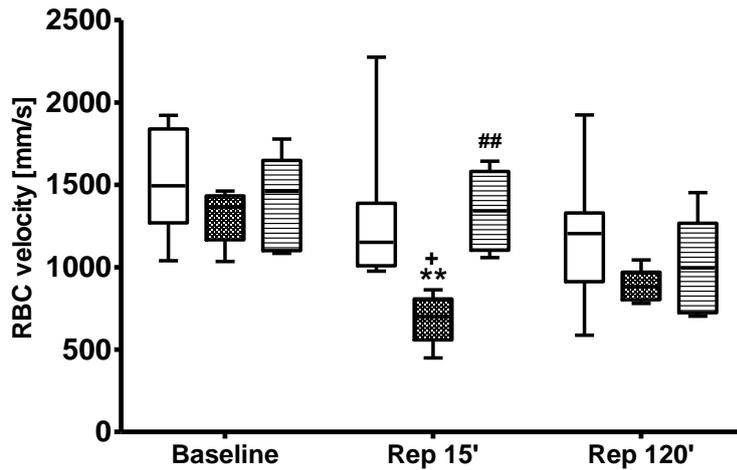


Figure 6. RBC velocity. Changes in red blood cell (RBC) velocity in the microvessels of the serosal surface of the ileum in response to a sham operation (empty boxes) or 45 min of mesenteric ischemia followed by 120 min of reperfusion (IR groups, checked boxes) and in rats treated with a normoxic (21% O₂) gas mixture containing 2.2% CH₄ for 5 min at the end of the ischemia and for 10 min at the beginning of the reperfusion period (IR+CH₄, hatched boxes). Measurements were performed at baseline conditions (Baseline), 15 min after reperfusion (Rep 15') and 120 min after reperfusion (Rep 120'). ** means p<0.01 between groups vs sham-operated group, ## means p<0.01 between IR vs IR+CH₄ groups, + p<0.05 compared with baseline values within groups.

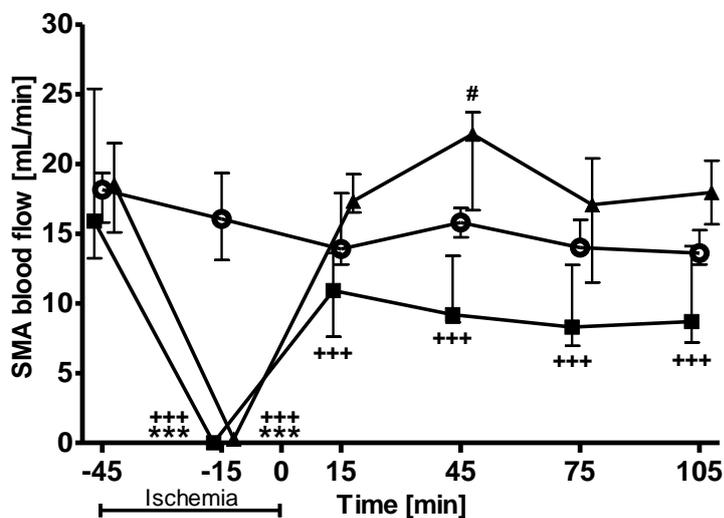


Figure 7. SMA blood flow. Changes in the superior mesenteric artery (SMA) blood flow in the sham-operated (empty circles), mesenteric IR (black squares), and CH₄-treated IR (black triangles) groups. 0 min on the X axis denotes the start of reperfusion. Results are expressed as median, 25th and 75th percentiles. Here, *** means p<0.001 between groups vs. sham-operated group; # indicates p<0.05 between CH₄-treated and IR groups, while +++ means p<0.001 compared with baseline values within groups.

ROS and RNS levels

Tissue NTyr concentration (**Figure 8A**) is an indicator of protein nitration produced by a chemical reaction associated with ONOO generation. The NTyr levels were elevated by the 180 min of reperfusion in the IR group as compared to those in the sham-operated controls, while the levels did not differ from the controls in the CH₄-treated group.

O₂^{•-} levels

A primary cellular ROS, O₂^{•-}, was detected in ileal biopsies at the beginning of experiments (**Figure 8B**), and no between-group differences were observed. At 15 min after the reestablishment of blood flow, the samples from the IR group contained significantly higher levels of O₂^{•-} than those from the control and CH₄-treated animals.

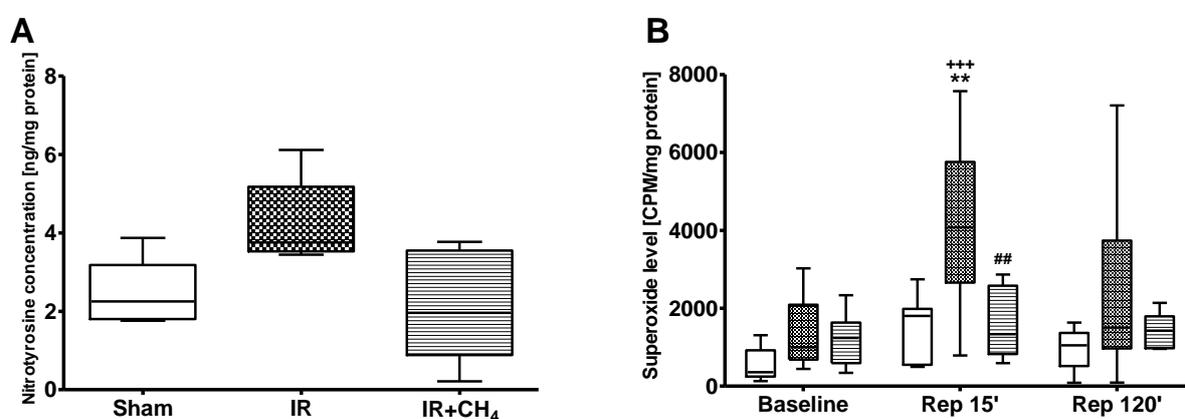


Figure 8. Intestinal nitrotyrosine levels and capacity to generate superoxide radical. Changes in mucosal NTyr (**panel A**) and O₂^{•-} (**panel B**) production in sham-operated control animals (Sham, clear boxes) and after 45 min of SMA occlusion and 180 min reperfusion (IR group, checked boxes). In the methane-treated group (IR+CH₄, hatched boxes) the animals inhaled a normoxic (21% O₂) gas mixture containing 2.2% CH₄ for 5 minutes in the final phase of the ischemia and for 10 minutes in the initial phase of the reperfusion period (or 15 minutes in total). The NTyr level was measured by means of EIA from samples taken after 180 min of reperfusion and normalized to the total protein content of the tissue. O₂^{•-} levels (panel B) were measured by chemiluminescence at the beginning of the experiment (Baseline), 15 min (Rep 15') and 120 min after reperfusion (Rep 120'), and they were normalized to the protein content of the tissue sample. Data are expressed as median, 25th and 75th percentiles. Here, ** means p<0.01 compared to the sham-operated group; ## means p<0.01 between the CH₄-treated group and IR group, while +++ means p<0.001 compared to baseline values within groups.

Tissue ET-1 levels

The ET-1 concentration was measured from plasma samples at the end of the 180-min reperfusion period. In the IR group, there was a significant elevation at 180 min after reperfusion relative to that in control animals. This elevation was significantly reduced in the IR+CH₄-treated group (**Figure 9A**).

Tissue MPO levels

The activity of MPO, a marker enzyme of PMN granulocytes, was assessed in intestinal homogenates at the end of the late reperfusion phase. A significant MPO elevation was present in both the IR and the CH₄-treated groups, indicating acute inflammation and extravasation of leukocytes into the tissue (**Figure 9B**).

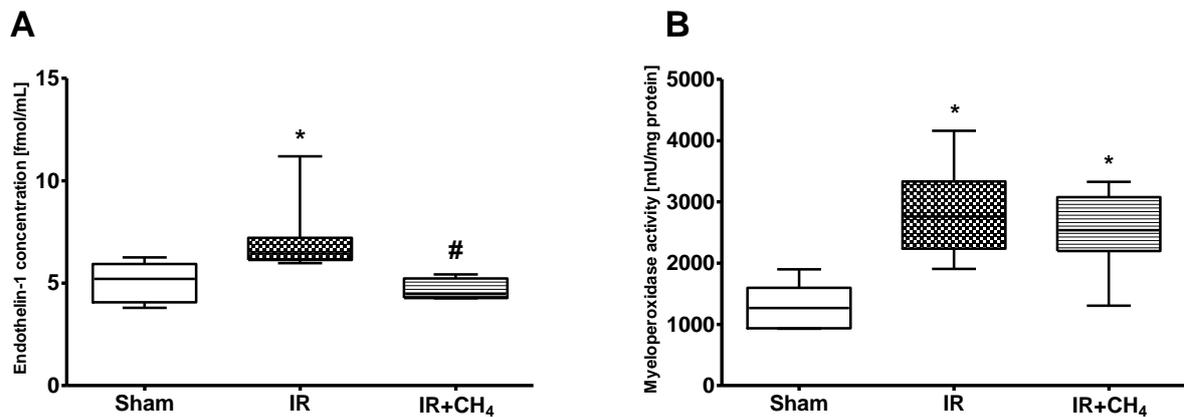


Figure 9. Endothelin-1 and MPO. Changes in tissue myeloperoxidase (MPO) activity and the plasma endothelin-1 (ET-1) concentration at the end of the 180 min reperfusion in the sham-operated (Sham, clear box), IR (checkered box), and CH₄-treated groups (IR+CH₄, hatched boxes). Here, * means $p < 0.05$ between groups vs sham-operated group, # means $p < 0.05$ between IR vs IR+CH₄ groups.

Structural integrity of small intestinal mucosa

Tissue samples were taken in the early reperfusion phase for conventional histology to provide structural data on the ileum mucosa. In the sham-operated group (**Figure 10A**), villus morphology was normal, but in the IR group (**Figure 10B**) extensively damaged, denuded and progressively shrinking villi were typically found in association with increased luminal debris formation. Furthermore, the congestion of RBCs in the microvessels of the villi was seen. CH₄ treatment (**Figure 10C**) preserved the integrity of the mucosal layer with moderate debris formation and slight alterations of the lamina propria.

Intravital CLSEM images were recorded to get information on the structural condition of the surface of the epithelium and the microscopic histology data were evaluated using a semiquantitative scoring system (**Figure 11D**). Normal villi with intact epithelial cells were observed in the control group (**Figure 11A**). Relative to the typically continuous, unbroken epithelial lining in the sham-operated animals, the mucosa was severely damaged after a 30 min reperfusion. Epithelial defects stretching across the villi were regularly seen (**Figure 11B**). No epithelial disruptions on the lumen surface were present in the CH₄-treated group (**Figure 11C**) and the microstructural damage (pathological changes?) reflected in the injury score was significantly lower than that observed in non-treated IR animals.

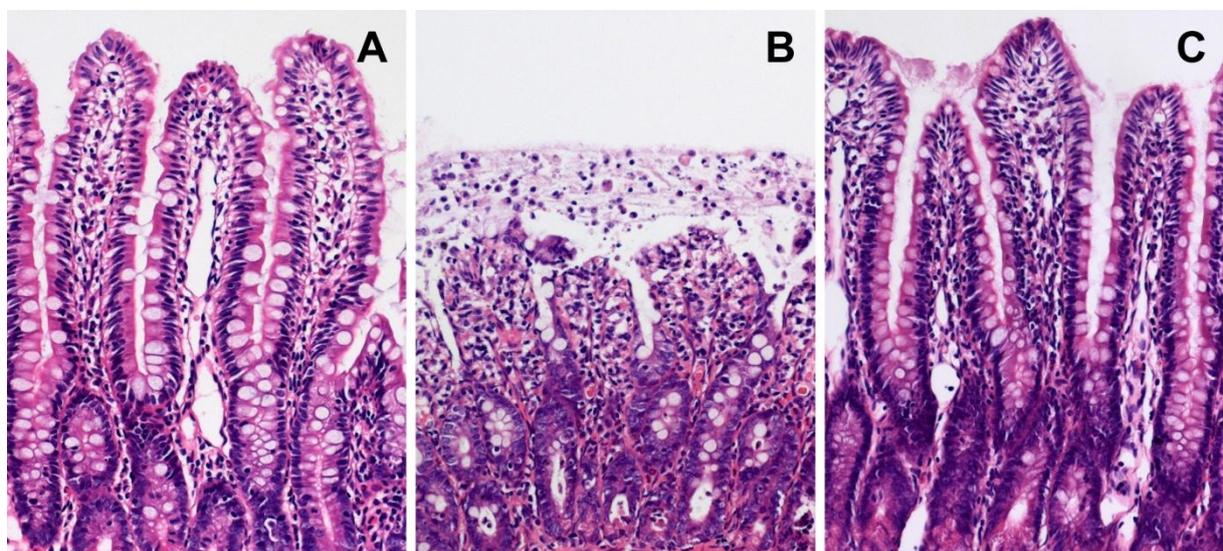


Figure 10. Conventional histology. Representative photomicrographs of the ileum mucosa 30 minutes after the end of the ischemia in sham operated animals (**A**), ischemia-reperfusion (**B**) and ischemia-reperfusion and CH₄ treated group (**C**). The images were taken with a 40x objective after hematoxylin-eosin staining.

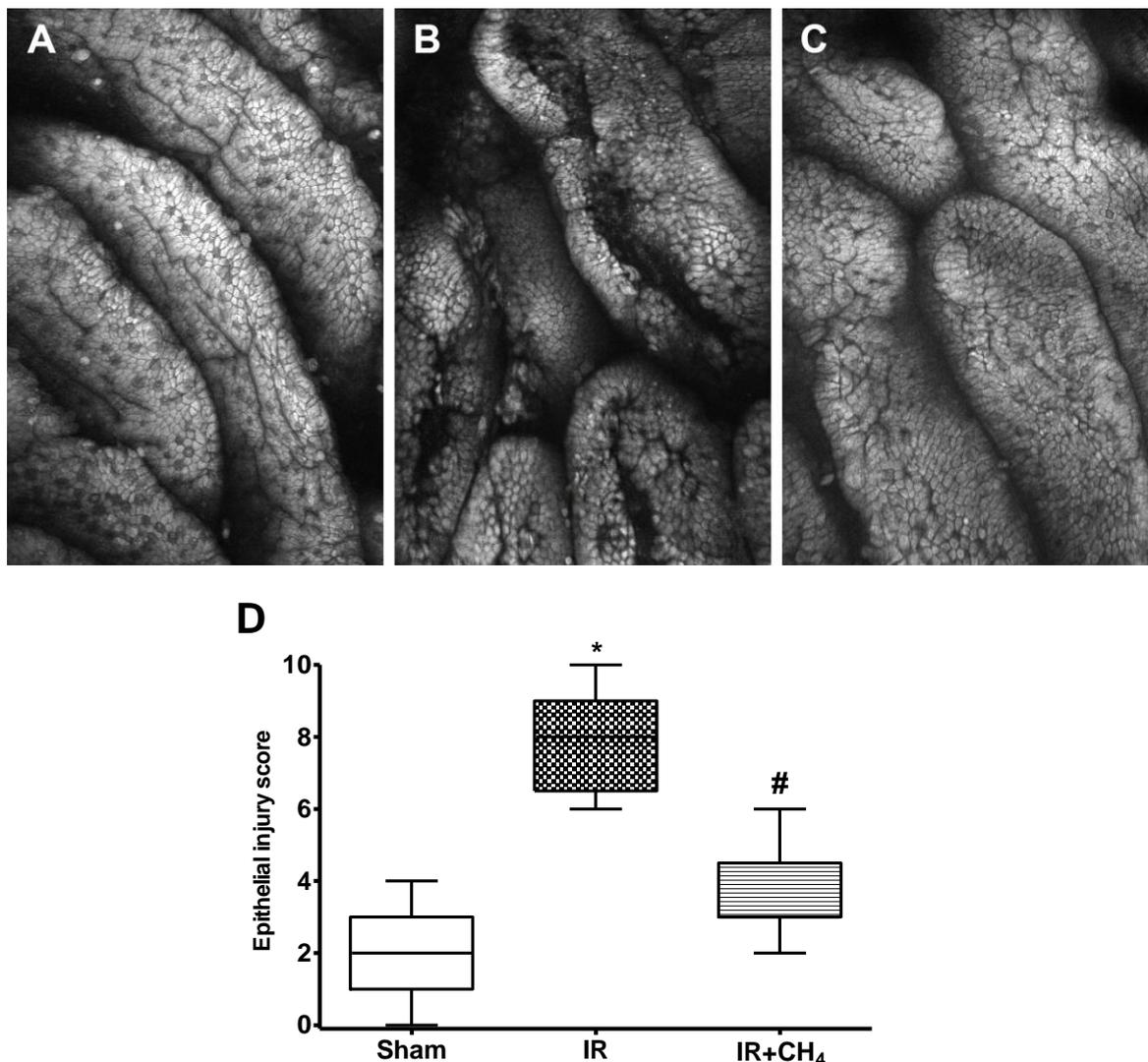


Figure 11. Confocal laser scanning endomicroscopic images of the intestinal epithelium. Top panel *In vivo* histology images recorded by confocal laser scanning endomicroscopy (CLSEM) after the topical administration of sodium acriflavine: **A:** Sham-operated group, **B:** IR group, 45 min of SMA occlusion and 30 min reperfusion, **C:** IR+CH₄ treated group, 45 min of SMA ischemia and 30 min reperfusion. Bottom panel Grading of *in vivo* histology on a semiquantitative scoring system. The plots show the median and the 25th and 75th percentiles. Here, * means $p < 0.05$ between groups vs. sham-operated group, # means $p < 0.05$ between IR vs IR+CH₄ groups.

3. The detection of NO by EPR without exogenously added spin-trapping molecules

EPR spectroscopy was applied to validate measurements of NO levels using endogenous trapping molecules. 16 h after an LPS injection, there was a dose-dependent increase at $g = 2.075$ and $g = 2.042$ in the complex EPR signal of liver, corresponding to hemoglobin-bound NO (NO-Hb) and Fe-bound NO (NO-Fe; **Figure 12B**). In blood samples taken from the same animals, only NO-Hb signals at $g = 2.075$ were present (**Figure 12A**). There was a

good correlation of increasing LPS doses with NO–Hb and NO–Fe signals (**Figure 12C-E**), which had been calibrated previously (data not shown). The correlation was strong between NO–Hb and NO–Fe signals as well (**Figure 12F-H**).

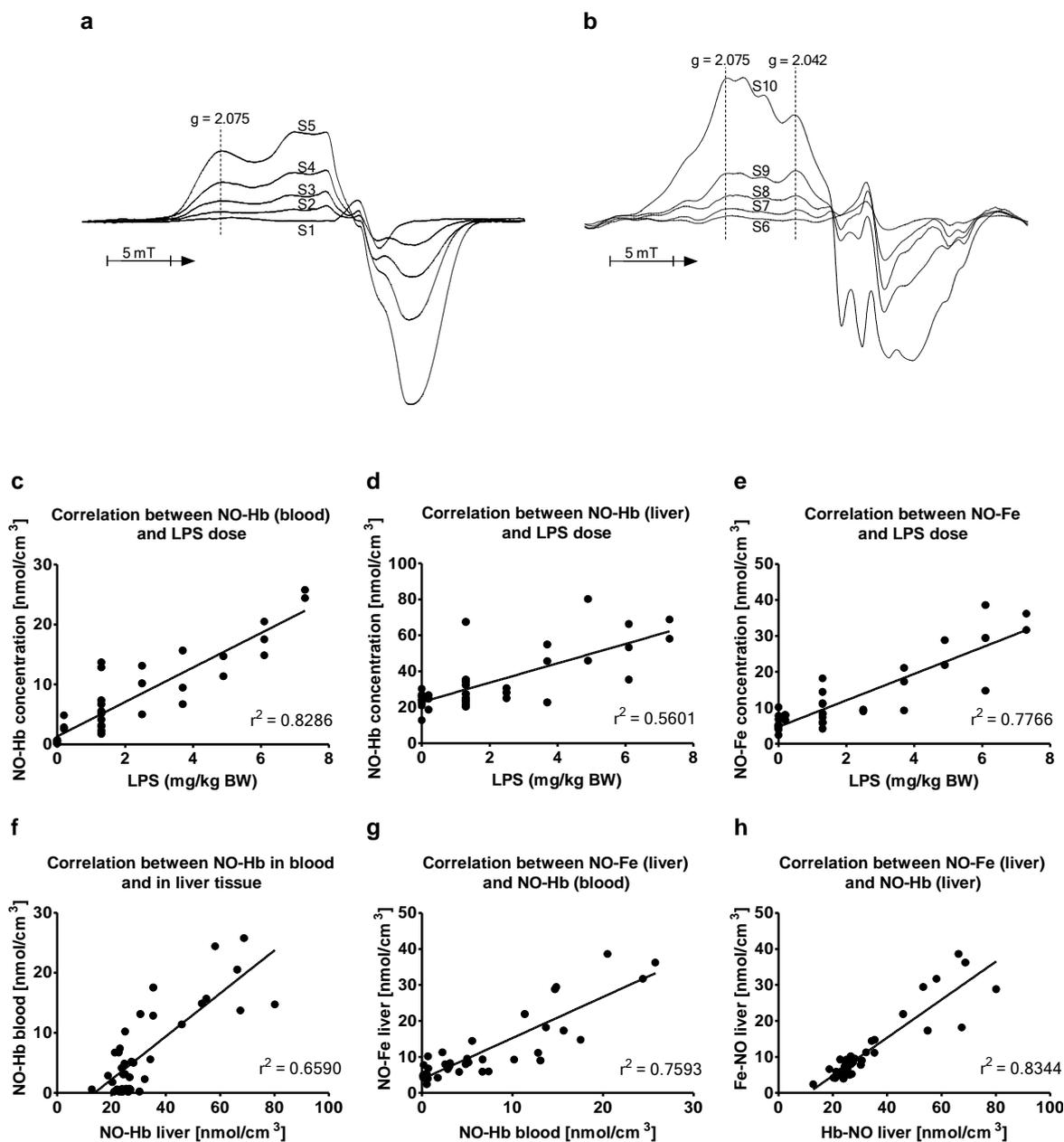


Figure 12. Quantification of NO formed in cells and in blood upon LPS treatment and the characterization of NO pools. The dose of LPS varied from 0.2 to 7.3 mg/kg body weight. Blood (A) and liver (B) EPR spectra were recorded 16 h after an injection of 0.2 (S2, S7), 3.7 (S3, S8), 4.9 (S4, S9), and 8.5 (S5, S10) mg/kg body weight LPS or an equivalent volume of NaCl (S1, S6). NO–Hb in blood (C) and liver (D), and NO–Fe in liver tissue (E) as a function of LPS dose. The correlation between NO–Hb in blood and liver (F), NO–Fe in liver and NO–Hb in blood (G) and NO–Fe and NO–Hb in liver (H). Abbreviations: m, slope; r, coefficient of determination; S, signal

4. The effects of CH₄ on NO release under ischemia

Using Na-DETC and exogenously added iron to form stable, intracellular NO-Fe-DETC complexes for EPR analysis, we measured NO levels in the duodenum and ileum of rats (see **Figure 13**). There was a trend showing increased NO concentrations at the end of a 45-min ischemia in both parts of the intestine as compared to those in sham-operated animals. The inhalation of a normoxic gas mixture containing 2.2% CH₄ for 10 min resulted in lower levels of NO. During the reperfusion phase, we did not see any similar trends (data not shown).

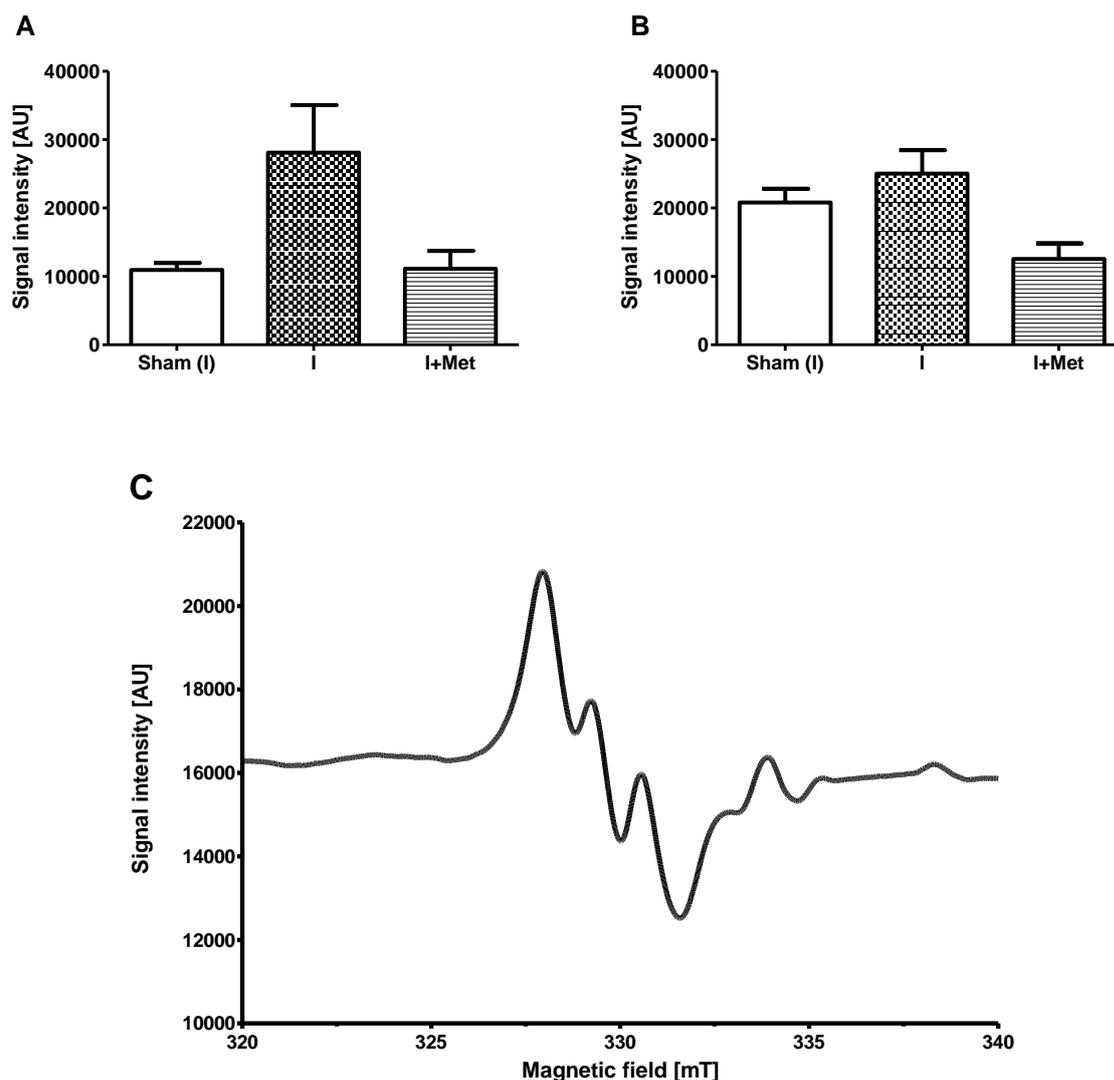


Figure 13. NO-Fe-DETC levels in samples taken from the duodenum (**panel A**) and ileum (**panel B**) of rats during mesenteric IR with or without CH₄ treatments. Control animals (Sham, empty bars), animals after 45 min of SMA occlusion without reperfusion (I group, checkered bars). In the CH₄-treated group (I+CH₄, hatched bars), the animals inhaled a normoxic (21% O₂) gas mixture containing 2.2% CH₄ for 10 min in the final phase of ischemia. Here, data are expressed as the mean ± SD (n=3 each). **Panel C:** characteristic EPR signal of NO-Fe-DETC.

To estimate whether or not CH₄ can modulate NO release from NO₂⁻ in anoxic tissues, liver homogenate was incubated with pure N₂ or N₂ supplemented with 2.2% CH₄. NO production was alleviated in the presence of CH₄; reaching significantly lower levels as compared to those for the N₂ only group from the 10th min after adding NaNO₂ (**Figure 14**).

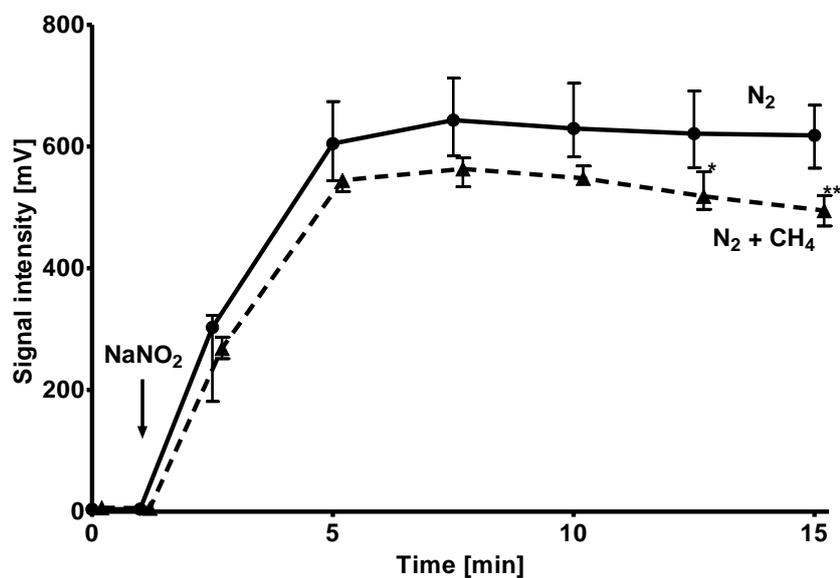


Figure 14. The NO generation of liver homogenate under anoxia. After equilibration with N₂ or N₂ containing 2.2% CH₄, 4.4 mM NaNO₂ was added (arrow). NO release was detected in real-time and it is plotted against the time elapsed. The signal intensity is proportional to the NO generated. Median values and the range are given. * means p < 0.05 and ** means p < 0.01 N₂+CH₄ vs N₂ group (n=3 each).

5. The effects of CH₄ on the microhemorheological parameters of whole blood

The deformability of erythrocytes taken from human blood was measured using a laser-assisted optical rotational method (Figure 15). Oxidative stress, induced by *in vitro* treatment with the oxidizer PMS, resulted in a significantly decreased elongation index from low to moderately high shear stress rates, as compared to that for the non-treated control samples. Normoxic CH₄ incubation, applied *after* the oxidizer incubation, was able to partly counteract the decreased rigidity of RBCs at moderate levels of shear stress (Figure 15B-D).

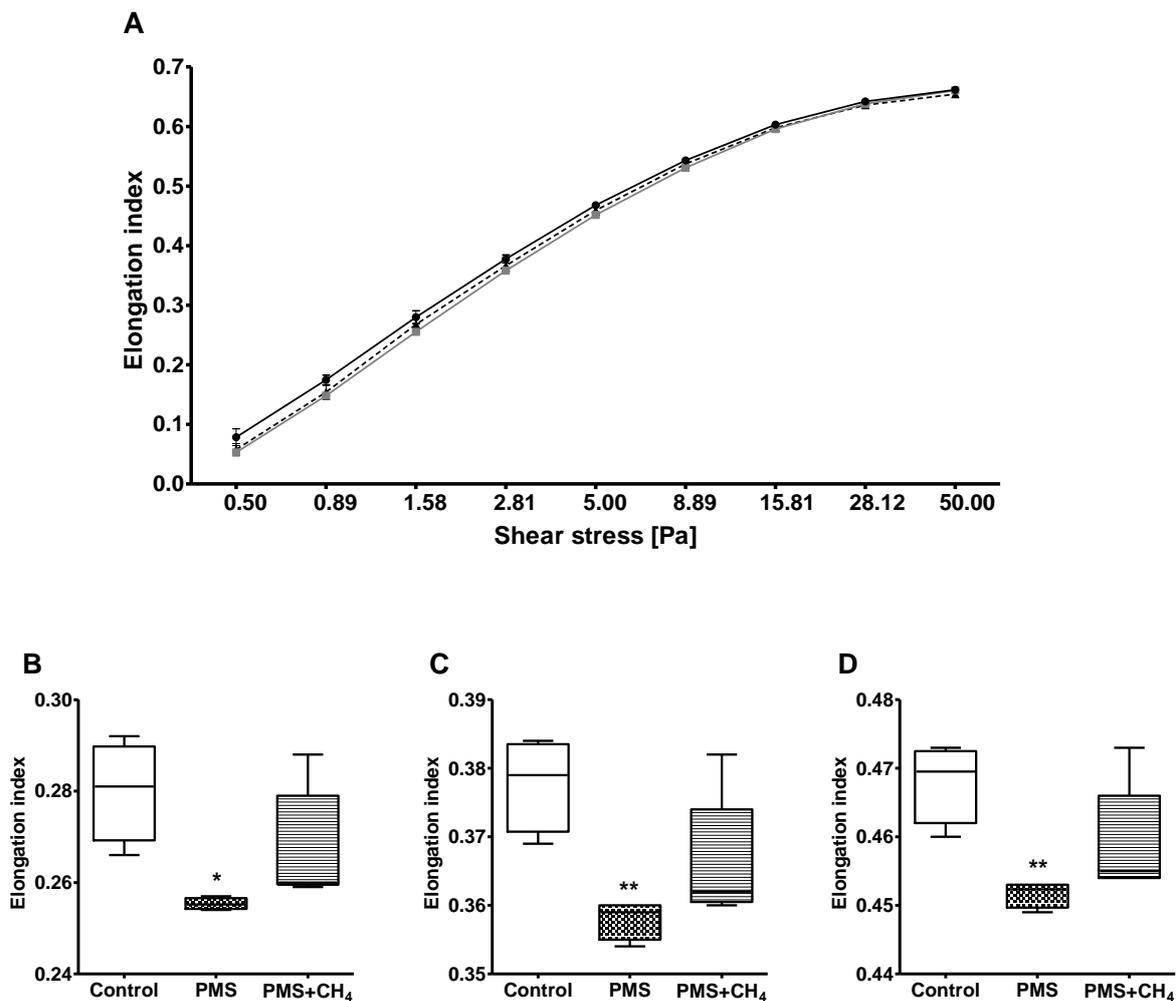


Figure 15. RBC deformability. Deformability changes of RBCs *in vitro* under increasing shear stress rates (panel A) and values at 1.57 Pa (B), 2.81 Pa (C) and 5.00 Pa (D) depicted as box plots. The higher elongation indices represent more elongated, hence more oval erythrocytes. Control samples (black line and empty boxes), samples incubated with PMS for 2 hours to induce oxidative stress (PMS groups, grey line and checkered boxes) and blood samples treated with a normoxic (21% O₂) gas mixture containing 2.2% CH₄ for 10 min after 2 h of PMS challenge (PMS+CH₄, dashed line and hatched boxes). On the shear-stress curve means \pm SD are given, while the box plots show the median and the 25th and 75th percentiles. Here, * means $p < 0.05$ and ** means $p < 0.01$ PMS vs sham-operated group and $n = 4-5$.

Oxidative stress *in vitro* increased the aggregation of erythrocytes at low shear-stress as compared to that in control samples (**Figure 16**). After applying CH₄, these values significantly decreased to the level of non-treated control samples.

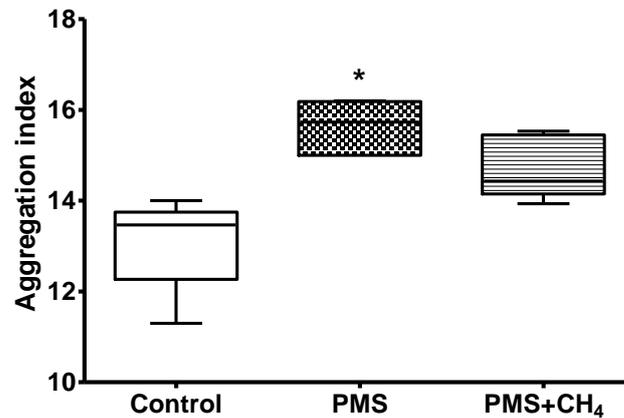


Figure 16. RBC aggregation. The light-transmission of rapidly aggregating samples at low shear rates was measured for 10 s and the aggregation index was calculated. Control samples (empty boxes), samples incubated with PMS for 2 hours to induce oxidative stress (PMS groups, checkered boxes) and blood samples treated with a normoxic (21% O₂) gas mixture containing 2.2% CH₄ for 10 min after 2 h of PMS challenge (PMS+CH₄, hatched boxes). The plots show the median and the 25th and 75th percentiles. Here, * means p<0.05 PMS vs sham-operated group and n = 4-5.

V. Discussion

The main purpose of the thesis was to summarize and interpret previous data on the *in vivo* properties of exogenous CH₄ in order to design and conduct further experiments, which help to extend our knowledge on the possible role and mechanisms of action of CH₄ in mammals. Today it is recognized that a variety of enzymes are generating gas mediators, molecules that exert important physiological functions. Also, there is a growing body of evidence which suggests that CH₄ can be produced in the eukaryotic cell. The anti-inflammatory potential of exogenously supplemented CH₄ has been repeatedly confirmed as well, but it is still unclear whether endogenous CH₄ levels are able to influence physiological or pathophysiological processes and important details of the mechanism of action still need to be clarified. In fact, there is an apparent difference between the amounts of CH₄ produced endogenously and the concentrations used exogenously, which might raise concerns about the physiological relevance of CH₄ as a gas messenger. From a pharmacological aspect, however, this is not a critical issue. Understanding the mechanism of action is the key to efficient and safe therapies, and on the basis of accumulated knowledge, the presumed physiological role of a compound can be clarified. In this work, the efficiency of a CH₄-based therapy was investigated in a small animal model of oxido-reductive stress and in a second step of accompanying *in vitro* studies, we tried to delineate a feasible mechanism of action. We should note that the results still cannot fully explain the effectiveness of CH₄, but they rather reflect the multifaceted responses of complex biological systems.

1. The kinetics of CH₄ transport *in vivo*

In the case of NO or H₂S, precursors stimulating endogenous release of the gas can be administered or enzymatic synthesis of the compounds can be induced (Szczesny *et al.*, 2014; Gero *et al.*, 2016). However, with CH₄, direct delivery is the only feasible option. One approach to apply effective CH₄ doses is by the enrichment of fluids, saline or other physiological solutions with the gas (Shen *et al.*, 2016; Xin *et al.*, 2016; Zhang *et al.*, 2016). This delivery method is similar to H₂-rich saline administration, which has been repeatedly used *in vivo* in animal models (Mao *et al.*, 2009; Zheng *et al.*, 2009; Chen *et al.*, 2015). While giving the right dose may be easier, the storage of supersaturated CH₄-containing solutions might be a challenging problem. Another important issue is the relatively low solubility of CH₄ in aqueous solutions, and this limits the amount of CH₄ which can be administered intravenously or intraperitoneally. Nevertheless, for chronic treatments it may be the method of choice, since CH₄ can be given in repeated bolus doses.

In our studies, gas inhalation was used with artificial air containing 21% O₂ and 2.2-2.5% CH₄. The solubility of CH₄ is rather low in water, but much higher in the lipid phase, hence we hypothesized that inhalation provides a quick way of saturating tissues with CH₄. The large alveolar surface of the lungs and the thin layer of alveolar cells with adjacent endothelium provide an ideal way to dissolve CH₄ in the blood. Red blood cells contribute to 40-50% of the total volume of the circulating blood; therefore erythrocyte lipid membranes are good candidates to carry the gas. Also, it has been shown that in this concentration range, CH₄ in the inhaled air has no measurable impact on O₂ delivery and saturation (Boros *et al.*, 2012).

Despite the fact that the efficacy of CH₄ after inhalation has already been demonstrated in different animal models of diseases, the distribution of the molecule has yet to be determined. With this in mind, we sought to assess the kinetics of tissue and blood levels of CH₄ during and after the inhalation protocol was used *in vivo* (see figures 3A and 3B). We found that even a brief inhalation of CH₄ resulted in high amounts of gas being dissolved and transported by the blood. Due to the apolar property of the molecule, the majority of gas is transported by RBCs (and lipoproteins) and was reasonable to suppose that the dissolved CH₄ influences properties of the RBC membrane. Upon reperfusion, the CH₄ content of the systemic blood immediately starts to equilibrate with the CH₄ level in the previously ischemic tissue, reducing the time needed to reach the target cells.

In parallel with blood CH₄ measurements, levels of the gas in the intestinal tissue were also assessed. There was a slight increase in CH₄ levels during ischemia as a consequence of the omnidirectional diffusion of the gas. Due to its non-polar properties, CH₄ can reach tissues without perfusion; hence it can exert effects under deep hypoxia. Upon reperfusion, as the blood flow restarted in the tissue, the levels of the gas rose significantly. Thereafter, about 50 min after the end of normoxic CH₄ inhalation no CH₄ is measurable either in blood or in the intestine. One limitation is that direct measurement of the actual levels of CH₄ dissolved in the tissue is not possible using the methods described here. In our study the headspace gas concentrations were measured over 10 min, during which the previously dissolved blood or tissue CH₄ continuously equilibrated between the samples and the gas phase. During inhalation, the intracellular levels are definitely higher than the amounts measured.

2. The bioactivity of exogenous CH₄ in the intestine

Enterocytes are energetically very active, and there is a delicate steady-state of cell loss and re-epithelialization in the small intestine. The maintenance of physical and immunological barriers between the interior milieu and intestinal lumen is of vital importance,

but the intestine is sensitive to hypoxia and the GI system is among the most vulnerable organs during circulatory redistributions (Vajda *et al.*, 2004). Although small fluctuations of metabolites can be compensated by high-energy phosphate systems within the cells of the bowel (Kitzenberg *et al.*, 2016), longer periods of ischemia or ischemia followed by a rapid reestablishment of the blood flow leads to severe damage.

The anti-inflammatory properties of CH₄ have already been examined in intestinal IR (Boros *et al.*, 2012) and we sought to look at the function of mucosal barrier in this setting. Here we could show that even a brief, 15-min normoxic CH₄ treatment modulated epithelial component of transmucosal permeability significantly and effectively prevented the IR-induced barrier loss in the early reperfusion phase (see figures 4A and 4B). The rationale behind the timing and dosage of CH₄ treatment was provided by the previous reports.

The assessment of the lumen-to-plasma clearance of the 4 kDa MW FITC-labelled dextrane allowed us to determine paracellular permeability, mediated predominantly by TJs (Szabo *et al.*, 2006). TJs are membrane-bound protein complexes connected to the cytoskeleton (Laukoetter *et al.*, 2006). The maintenance of the closed conformation is energy dependent; therefore deteriorating intracellular ATP levels increase permeability (Wattanasirichaigoon *et al.*, 1999). CH₄ can in theory protect TJs from opening by directly influencing membrane fluidity or by preserving the ATP levels of epithelial cells; moreover, it can act indirectly, such as by a NO-mediated mechanism. Actually, the reduction of tissue NTyr levels does indeed suggest that CH₄ may improve the epithelial barrier function by limiting the inhibitory effect of NO on mitochondrial electron transport chain complexes I and IV (Brown, 2001). This, in turn, may elevate ATP production early in the reperfusion phase, allowing cells to attain sufficiently high energy levels to maintain TJ proteins in the tightly closed conformation.

In parallel with the improved barrier function, both conventional histology and CLSEM revealed preserved structure of the most luminal layers of the mucosa (see figures 10 and 11). There is a multitude of factors contributing to mucosal injury upon reperfusion (Granger *et al.*, 2015). In the current protocol the decrease in epithelial permeability after CH₄ inhalation was associated with reduced ROS and RNS generation and decreased ET-1 levels. In addition the CH₄ treatment influenced the PMN infiltration, a specific cellular component of inflammatory reactions (figures 8 and 9). CH₄ treatment improved both serosal microcirculation and the SMA flow facilitating oxygen delivery to cells and allowing effective oxidative phosphorylation in mitochondria (see figures 6 and 7). The normalized

ET-1 plasma levels in the CH₄-treated groups may reflect the improved local microcirculatory state on one hand and reduced inflammatory activation on the other hand.

There are, however, other links between the improved functional and structural status of the GI mucosa upon CH₄-treatment. It was shown previously that normoxic CH₄ treatment modulated XOR-derived O₂^{•-} (Boros *et al.*, 2012). However, extracellular, XOR-derived O₂^{•-} can activate PMNs through the TLR4 - NF-κB pathway (Lorne *et al.*, 2008). The significantly reduced ROS and RNS levels (O₂^{•-} and NTyr) after CH₄ treatment can further prevent overwhelmed PMN activation and, as a consequence, the consumption of O₂ by NADPH oxidase. Since one of the main determinants of mucosal microcirculation during reperfusion is an enhancement of the interactions between leukocytes and the endothelium (Vollmar *et al.*, 2011), a reduced number of activated PMNs can readily improve tissue perfusion.

Early barrier dysfunction is a crucial in the development of delayed and remote organ injury following mesenteric IR. Posts ischemic gut serves as an incubator, where endotoxin leaking from the lumen to the interstitium primes PMNs (Moore *et al.*, 1994). Significantly, the direct effect of normoxic CH₄ on NADPH oxidase activation was shown earlier (Boros 2012, Mészáros, unpublished data). This observation is in agreement with the findings presented in a recent publication of Zhang and coworkers, who provided conclusive evidence that CH₄-enriched saline solution suppresses TLR4-induced nuclear factor-kappa B (NF-κB) / mitogen-activated kinase (MAPK) activation in cultured macrophages.

The study design allowed us to determine the barrier function immediately after reperfusion and in the later phases, the effects of CH₄ inhalation were also studied 60 min and 180 min after the restoration of the mesenteric blood flow. The epithelial barrier of the ileum was already partially restored in the later phase of the reperfusion, as indicated by the lower FD4 clearances (see Figure 4.). In fact, the restoration of the epithelial barrier is a remarkably speedy process, involving the flattening of epithelial cells to cover injured villus surface and the shrinkage of the villi to reduce the damaged area (Derikx *et al.*, 2008; Matthijsen *et al.*, 2009; Grootjans *et al.*, 2011). Improved micro- and macrocirculation, restored O₂ supply and the normalization of the energetic state of the intestinal cells can all contribute to the efficacy of this process, leading to the closure of TJs. However, the difference between the intensities of early and later changes in epithelial permeability suggest that upon re-establishment of the blood and oxygen supply, the endogenous defense mechanisms cannot immediately control or counteract the damaging reactions. Salvage therapies should target the initial steps to avoid long-term or distant consequences of the barrier damage, and CH₄ treatment can fill this need.

Even a brief 15-min CH₄ inhalation preserved the function of the postischemic gut, hence continuous or longer CH₄ supplementation might have additional beneficial effects.

The effects of CH₄ on the IR-induced changes in vascular permeability are less clear because in this model the epithelial and endothelial permeability-specific changes were affected differentially by the IR cycle. Epithelial permeability significantly deteriorated immediately after reperfusion, while the extravasation of the circulating endothelial tracer did not increase to a great extent. Hence, the relative effect of CH₄ administration on microvascular permeability could not be evaluated simultaneously. Nevertheless, possible technical limitations (i.e. interaction between FITC-labeled dextran and Evans blue measurements) cannot be ruled out. The epithelial and endothelial permeability measurements were conducted simultaneously in the same animal for technical and ethical reasons, and this approach could influence the sensitivity of these methods. Therefore, future studies are needed to examine the influence of CH₄ on microvascular permeability directly and separately.

3. The detection of NO in biological samples

Along the above lines, one might suppose that CH₄ accumulates transiently at membrane interfaces, and a disproportionate increase in relationship with other gases, such as NO, may alter enzyme-linked processes, and ultimately mitochondrial function. Although much is known about the role of NO in physiology and pathophysiology, the determination of its actual levels in biological samples is not an easy task. NO has a half-life of milliseconds in tissues, much shorter than other inflammatory mediators. The determination of NO₂⁻/NO₃⁻ amounts by the Griess reaction or with chemiluminescence methods with higher sensitivity is an option, but in hypoxia, when nitrite reductases consume NO₂⁻, these can lead to erroneous results (Griendling *et al.*, 2016). EPR spectroscopy is currently the most suitable method used to measure NO directly. It has very high specificity for NO as it forms complexes with exogenously added spin traps (e.g. Na-DETC and FeSO₄). Unfortunately, these compounds are toxic. Moreover, such molecules can interfere with endogenous NO utilizing pathways through their high affinity to NO.

An alternative is to utilize endogenous traps of NO for this purpose. Intracellularly produced NO reacts with ferrous ions to form dinitrosyl-iron complexes (NO-Fe) (Sergent *et al.*, 2005). In contrast, in the vasculature NO forms mononitrosyl-hemoglobin complexes (NO-Hb) (Gow *et al.*, 1998) with hemoglobin. Since no ferrous ions occur in the blood, the two complexes, located at $g = 2.075$ and $g = 2.042$ are characteristic for NO-Hb and NO-Fe, respectively.

LPS administration is a standard way of inducing acute TLR4 - NF- κ B-mediated inflammation. We were able to demonstrate in intact, non processed, frozen samples that LPS increased hepatic NO levels in a dose-dependent way, both in the intracellular and intravascular compartments and NO-Hb levels in the systemic circulation (see Figure 12). The concentrations of NO in naturally occurring complexes were substantially lower compared with previous values stated in the literature of intracellular NO using the specific NO trap iron-diethylthiocarbamate in a similar model (Kozlov *et al.*, 2005). We may conclude that only part of NO produced intracellularly is scavenged by dinitrosyl-iron complexes and the residual NO contributes to physiological NO signaling or diffuses in the blood (Kozlov *et al.*, 2001a).

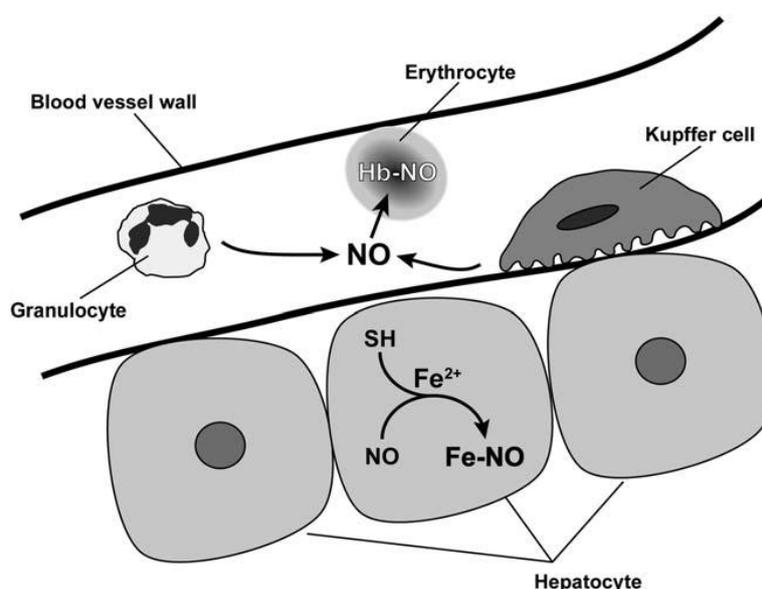


Figure 17. Endogenous traps of NO inside parenchymal cells and small blood vasculature of the liver. This scheme shows that NO-Hb complexes predominantly reflect extracellular NO generation by immune cells, whereas NO-Fe is a matter of intracellular NO synthesis. After being synthesized by hepatocytes, NO is trapped by ferrous ions (Fe^{2+}) from the cytosolic free iron pool, which forms a dinitrosyl-iron complex (NO-Fe) with two NO molecules and two thiol groups of proteins (SH). We assume that a small fraction of the NO molecules synthesized in hepatocytes diffuses into the microvasculature, where together with the NO molecules synthesized by immune cells (granulocytes, Kupffer cells), it forms mononitrosyl-hemoglobin complexes (NO-Hb).

4. The interaction of CH_4 with NO-related nitrosative stress

NO is an exemplary double edged sword in biology. In nanomolar concentrations it is vasorelaxant and antiadhesive, but in the micromolar range NO inhibits mitochondrial respiration in a reversible way (Cooper *et al.*, 2007; Cooper *et al.*, 2008). It is unlikely that NO can inhibit cytochrome c oxidase during normoxia due to the high K_M of the enzyme for NO. But under hypoxia and with high NO concentrations it is a possible pathway. In parallel, the half-life of NO is substantially longer if O_2 levels are low (Thomas *et al.*, 2001). In fact,

elevated NO levels could be present during the ischemic phase, as data obtained by means of EPR indicates. With CH₄ treatment, however, the overproduction of NO could be reduced (see Figure 13). Significantly, we detected NTyr formation in the intestinal tissue, an indicator of ONOO, which requires both O₂^{•-} and NO to form. The non-specific nitration of ETS proteins by ONOO is an additional, non-reversible inhibition of mitochondrial OXPHOS (Brown *et al.*, 2004). Thus, the reduction of NTyr levels after CH₄ treatment is an indirect sign of the influence of mitochondrial and NO-linked reactions.

The source of NO under hypoxia is thought to be NO₂⁻, reduced mostly by enzymatic reactions and by small molecules under acidic pH. Apart from hemoglobin within RBCs and the mitochondrial ETC itself XO acts as NO₂⁻ reductase as well. It has been shown that CH₄ inhibits XOR under normoxic conditions, a feature which has important role in alleviating ROS-induced damage during reperfusion, when XO is one of the main ROS generating enzymes. The same enzyme can function as NO₂⁻ reductase under hypoxia and low pH, conditions, which are fulfilled during ischemia.

Considering this information, we set out to measure NO release in real-time *in vitro* from liver tissue homogenate under anoxic conditions, mimicking ischemic tissue. After addition of Na-NO₂⁻ as a source of NO, in the anoxic-CH₄-incubated group, the production rate of NO was significantly lower than that with N₂ atmosphere only (see Figure 14). Endogenous levels of XDH are similarly high in the liver and in the small intestine. Accordingly, it can be assumed that modulation of XO by CH₄ contributes to the reduced NO levels, but other mechanisms cannot be excluded with certainty. How exactly CH₄ regulates NO levels is still unclear, but the interaction of the molecule with any of the O₂-dependent NOS isoforms seems unlikely, considering the deep hypoxia during ischemia. Nevertheless, modulation of iNOS induction during the reperfusion cannot be ruled out.

Taken together, we were able to demonstrate that increased CH₄ input reduces NO production and NTyr levels in hypoxic organs and in tissue homogenates under anoxia. Apparently, there is a contradiction between the reduction of anoxic/hypoxic NO generation and improved microcirculation upon reperfusion. Nevertheless, the maximal vasodilating effect of NO requires much lower levels than those needed for inhibition of cytochrome c oxidase. What is more, it was demonstrated previously that intracellular NO readily diffuses out from the cells in the circulation, but intravascular NO does not diffuse into the cells (Kozlov *et al.*, 2001a). Spatial differences of NO production and utilization thus play a substantial role, but the explanation of this phenomenon requires more study.

5. The direct action of CH₄ on erythrocyte deformability

The tissue NTyr level is an indicator of protein nitration, associated with elevated ONOO levels. ONOO is a potent initiator of membrane lipid peroxidation (Hogg *et al.*, 1999) and it was reported earlier that IR-associated lipid peroxidation decreases membrane fluidity in various tissues (Dobretsov *et al.*, 1977) and in erythrocytes as well (Watanabe *et al.*, 1990). RBCs are rich in iron, which catalyzes ROS formation through the Fenton-reaction, making them highly susceptible to IR damage (McCord, 1985; Baskurt *et al.*, 1998; Nanobashvili *et al.*, 2002). The membrane and cytoskeleton are together responsible for altering the shape of erythrocytes (Reinhart, 2001). Lipid peroxidation breaks the connection between the two components (Mohandas *et al.*, 1993) and, consequently, both deformability and aggregation of the RBCs is influenced in a detrimental way (Kayar *et al.*, 2001). Brath *et al.* reported worsened RBC deformability and increased aggregation in the early reperfusion after experimental mesenteric ischemia in the rat portal vein (Brath *et al.*, 2010). Normal erythrocytes are approximately 25% larger than the mean diameter of capillaries, so RBC deformability is a prerequisite in normal capillary blood flow (Reinhart *et al.*, 1985; Baskurt *et al.*, 2003).

Based on this background, we evaluated RBC velocity as a measure of perfusion in the serosal microvessels, and direct intravital data were obtained on the deranged intestinal microcirculation. Furthermore, CH₄ inhalation was associated with improved serosal microcirculation and the treatment also influenced the SMA flow. Hence, we hypothesized that CH₄ can modulate erythrocyte deformability upon being dissolved in the membrane and it may restore the bullet-shaped RBCs and allow them to flow freely through microvessels, thus improving microcirculation. With this in mind, an *in vitro* microrheological study with human whole blood was designed without the confounding *in vivo* effects of vasoactive metabolites. From methodological considerations, PMS, an oxidizing agent was used prior to normoxic CH₄ treatments, an approach analogous to *in vivo* CH₄ application during the reperfusion phase only. We demonstrated that PMS incubation significantly reduced the elongation index of RBCs, thereby confirming the contribution of lipid peroxidation in this model system (see Figure 15). Upon normoxic CH₄ treatment, the RBC deformability improved at low-to-moderate shear stress rates, suggesting a direct effect of CH₄ on membrane fluidity and/or membrane-cytoskeleton junctions.

Some studies claim that pathological RBC aggregation plays an even greater role in microcirculatory dysfunctions. Therefore, aggregation was measured in the same experimental setting at low shear stress rates. Once again, the increased aggregation index of

RBCs provided evidence for the role of oxidative stress in our set-up; and CH₄ treatment alleviated this response (see Figure 16). Moreover, with higher PMS concentrations, the effect of CH₄ treatment was even more pronounced (data not shown). Thus, these data provide evidence for the direct effects of CH₄ exerted in biomembranes of erythrocytes. Interestingly, XO inhibition with allopurinol prevents the RBC rigidity changes after IR *in vivo* (Nemeth *et al.*, 2006). Since CH₄ is able to inhibit XO (Boros *et al.*, 2012), an improvement of the microcirculation upon CH₄ treatment is the net result of a complex mechanism, inhibition of XO leads to a decreased O₂^{•-} production and less membrane damage, while, at the same time, the accumulation of CH₄ in the lipid phase further increases RBC deformability.

VI. Summary of the new findings

Previous reports with exogenously administered CH₄ corroborated the findings on the anti-inflammatory effects of the molecule, showing alleviated TNF- α , IL-6, IL-1 β levels, improved antioxidative defense mechanisms and reduced signs of oxidative damage. Based on these results two hypotheses for the mechanism of action were formulated and tested.

1. CH₄ is quickly transported from the lungs to the small intestine by the circulating blood. After the inhalation of normoxic air containing 2.2% CH₄ blood concentration of CH₄ increases at least 2-3-fold over its basal levels; this increase is sufficient to modulate the generation of ROS and RNS.
2. The early and later consequences of intestinal IR were characterized by biochemical, macro- and microcirculatory and morphological parameters. The IR-induced circulatory and structural damage was accompanied by increased epithelial permeability, demonstrating mucosal barrier damage. Normoxic CH₄ inhalation effectively prevented the elevation of intestinal epithelial permeability, maintaining structural integrity of the mucosa and improving biochemical signs of inflammation in the early reperfusion. These data support previous findings on CH₄ bioactivity and establish a mucosa-protective role for exogenous CH₄ to modulate the IR-induced pro-inflammatory activity locally in the small intestine.
3. A new, toxic spin trap-free EPR spectroscopy method was developed for NO detection. The method was validated for NO measurements from both the intra- and extracellular compartments of the same samples in tissue biopsies.
4. With EPR spectroscopy reduced NO levels were detected in the intestinal tissue after normoxic CH₄ administration, and this finding was supported by *in vitro* data collected in anoxic liver tissue during CH₄ incubation. Collectively, these results confirm that exogenous CH₄ administration effectively reduces NO levels in the ischemic tissue.
5. The results of this study revealed that CH₄ *in vitro* improves the deformability of red blood cells during simulated oxidative stress, which may contribute to the improvement of the effectiveness of mesenteric microcirculation in the posts ischemic small intestine.

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IX. Annex

